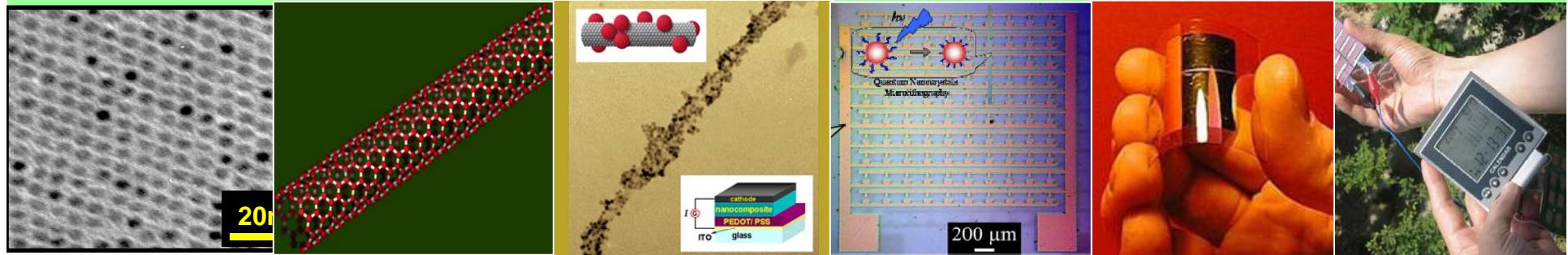




# Organic-Inorganic Hybrid Materials for IR-Photodetection and Photovoltaics



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Hannam University, Daejeon 305-811, Korea*

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- Introduction
- Quantum Dot-based Hybrids
  - \* Enhancing the Photocurrent Density
  - \* Photopatternable Quantum Dots
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- C<sub>60</sub> Derivatives
- Summary

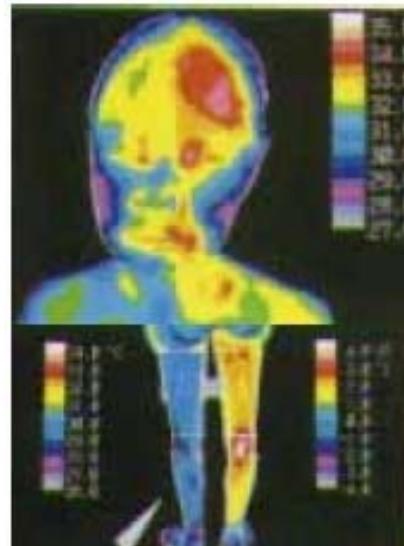
# IR Photodetection

Increase detectivity → Increase number of applications

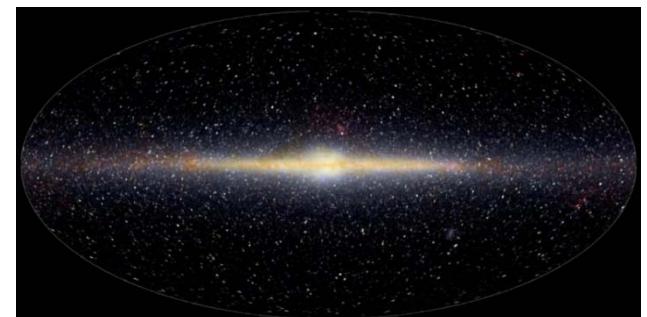
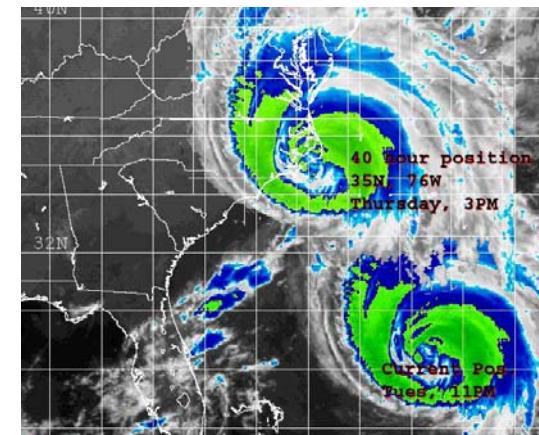
Military



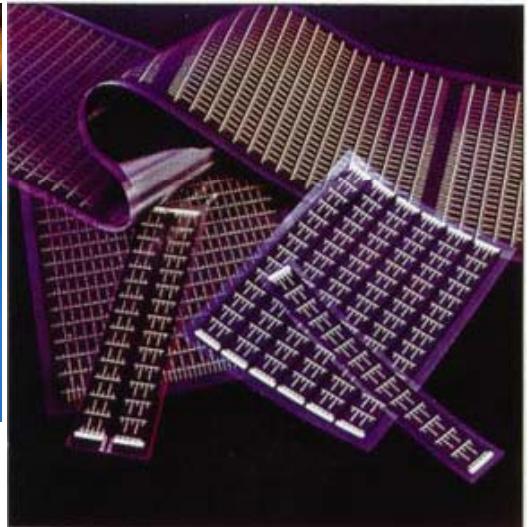
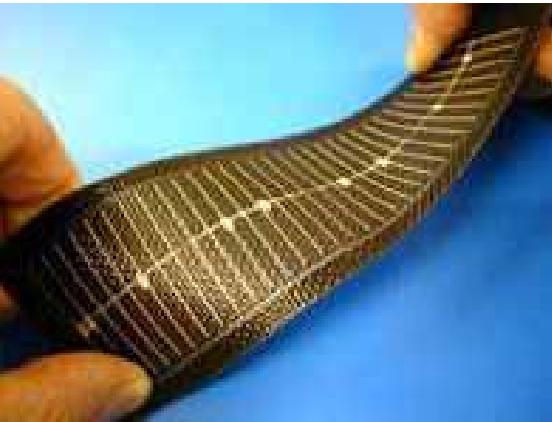
Medical  
(Thermal Imaging)



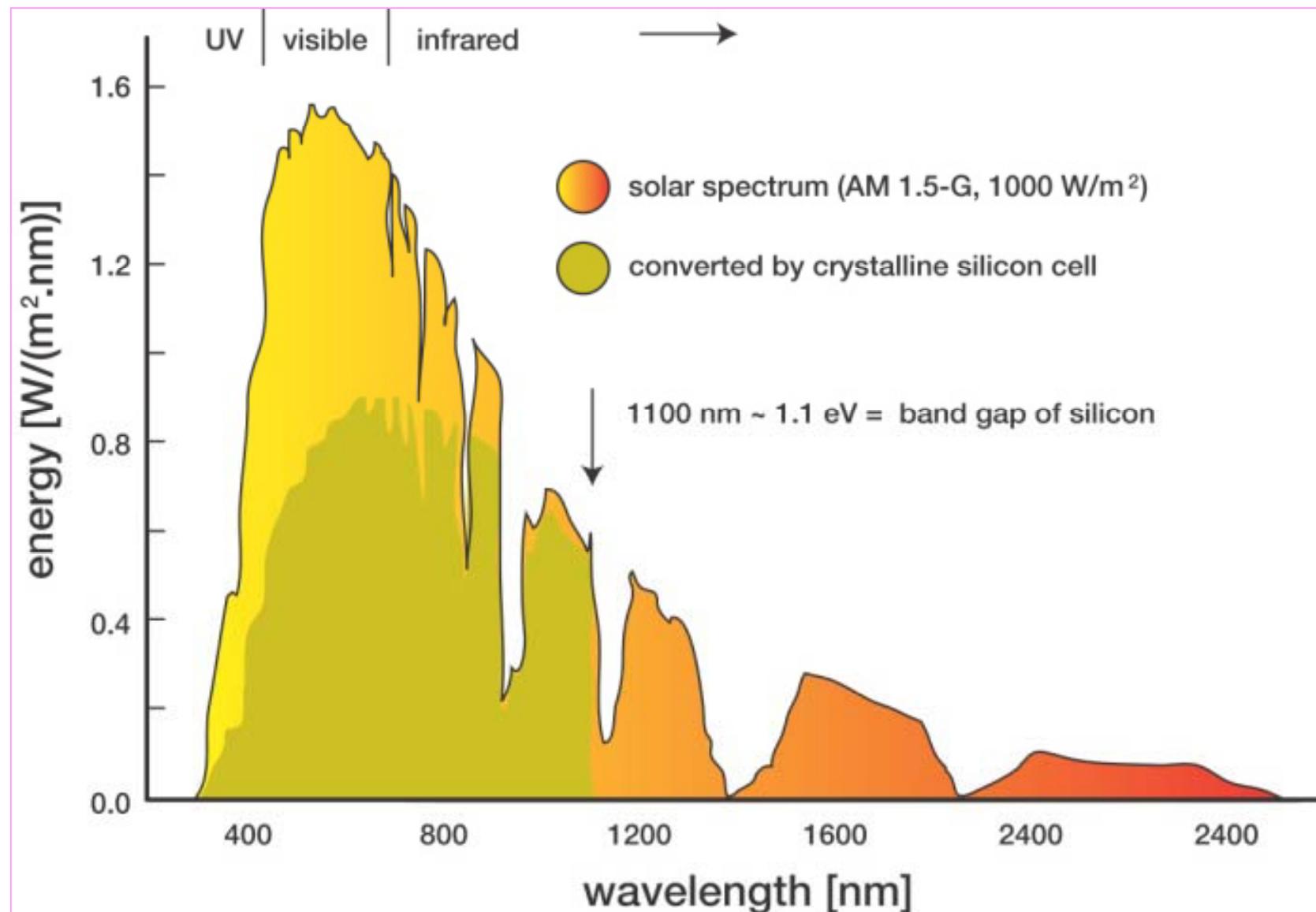
Weather



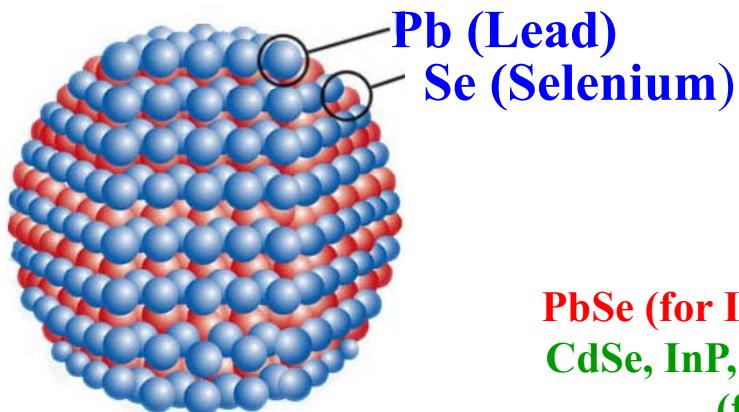
Astronomy:  
Infrared Image of  
the Milky Way



# Solar Spectrum



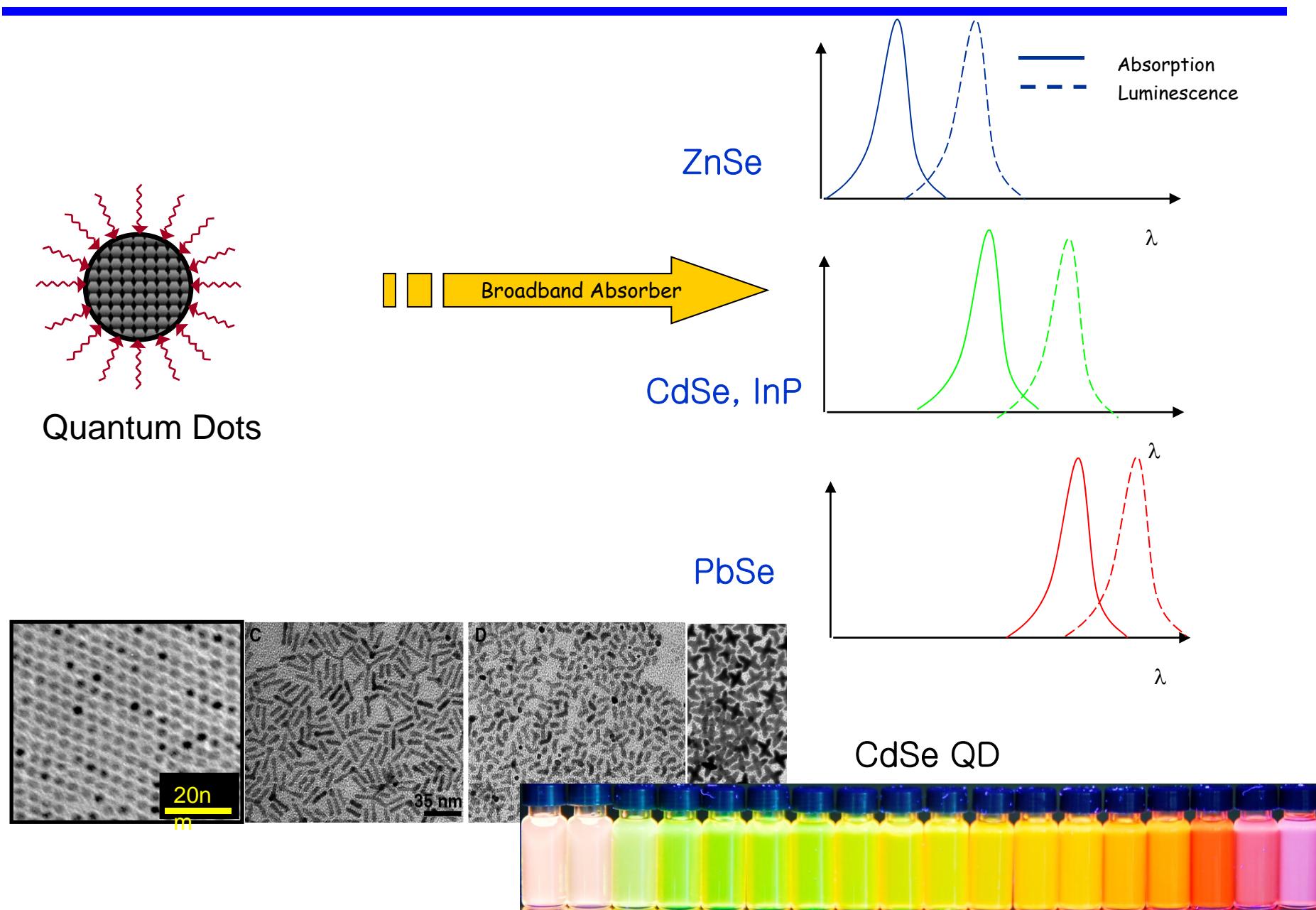
## (Nanocrystal Quantum Dots)



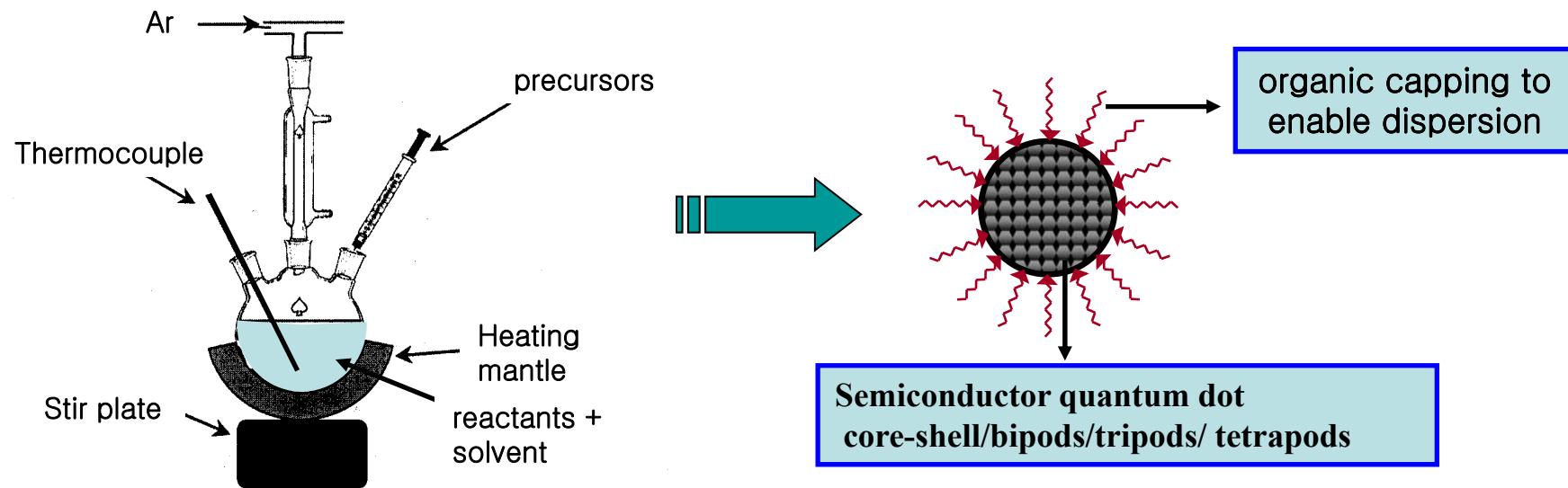
**PbSe (for IR)**  
**CdSe, InP, InP-CdS core-shell**  
**(for visible)**  
**ZnSe, CdS (for UV)**

- Semiconducting characteristics
- Excellent quantum size effects
- Efficient excitonic generation
- Tunable absorption and emission in the **UV to IR**

# Tuning of Spectral Response by Choosing Quantum Dots



# Preparation of PbSe Nanocrystals

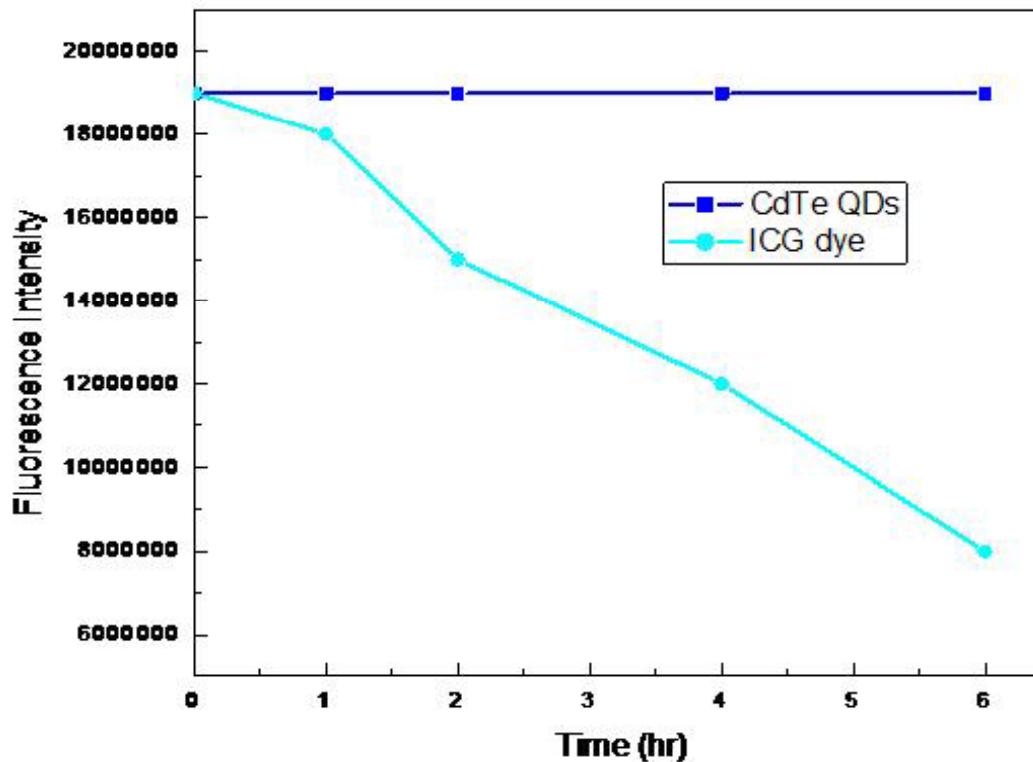


(PbO + Oleic acid in tri-*n*-octylamine solvent)+ TOP-Se or TBP-Se

**Comments:** Size and shape control + Surface functionalization of the particles are very important steps.

# Comparative Environmental Stabilities of IR QDs and IR Organic Dyes

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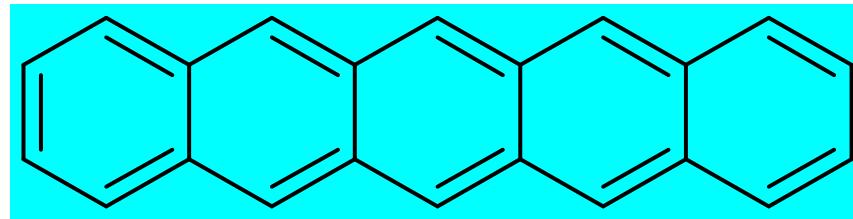
*Excitation - 720 nm  
Emission - 830 nm.*

*Normal laboratory  
environmental storage  
conditions.*

# Hybrid Materials for **IR** Photodetection

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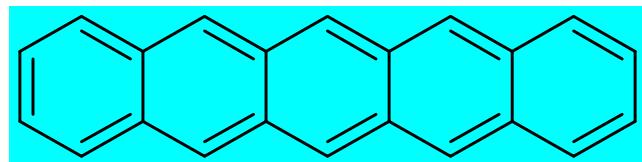
**QD / Pentacene / PVK**  
**Polymeric Nanocomposite**



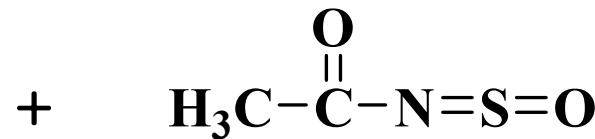
**Pentacene**

# Synthetic Route for Pentacene Precursor

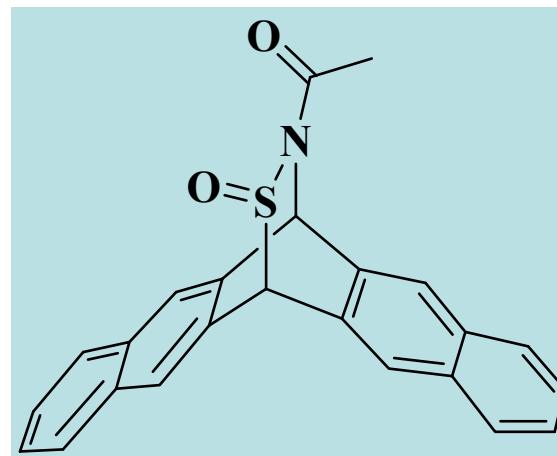
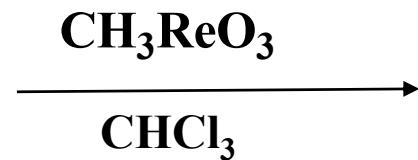
Ali Afzali, et.al, *J. Am. Chem. Soc.* **2002**, *124*, 8812.



Pentacene



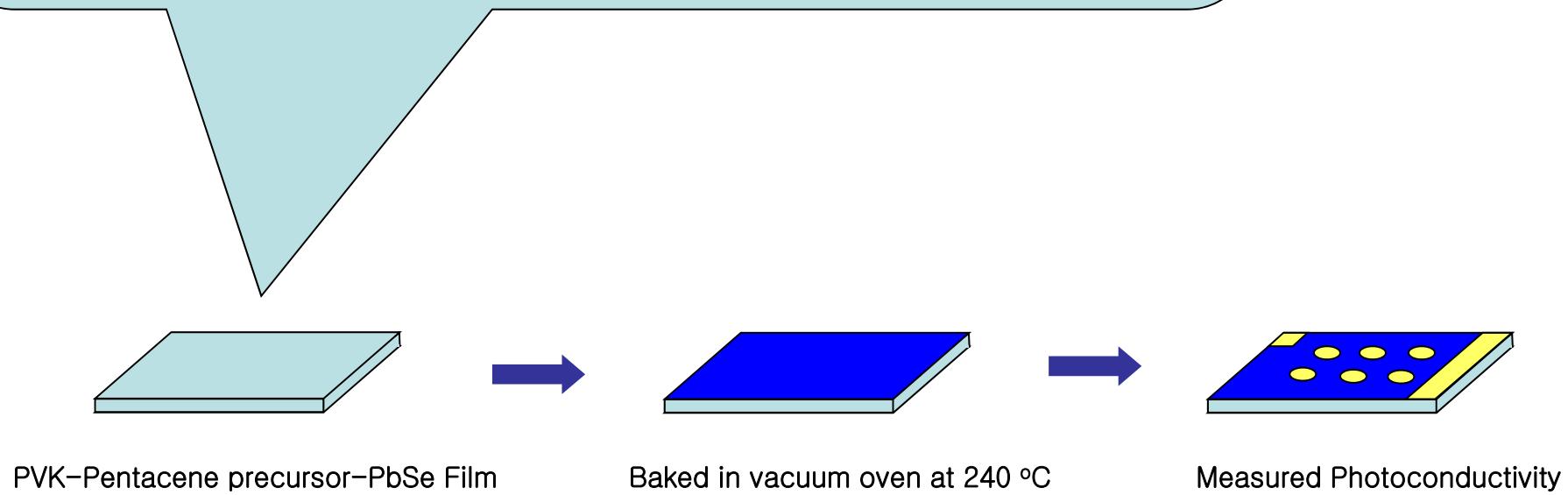
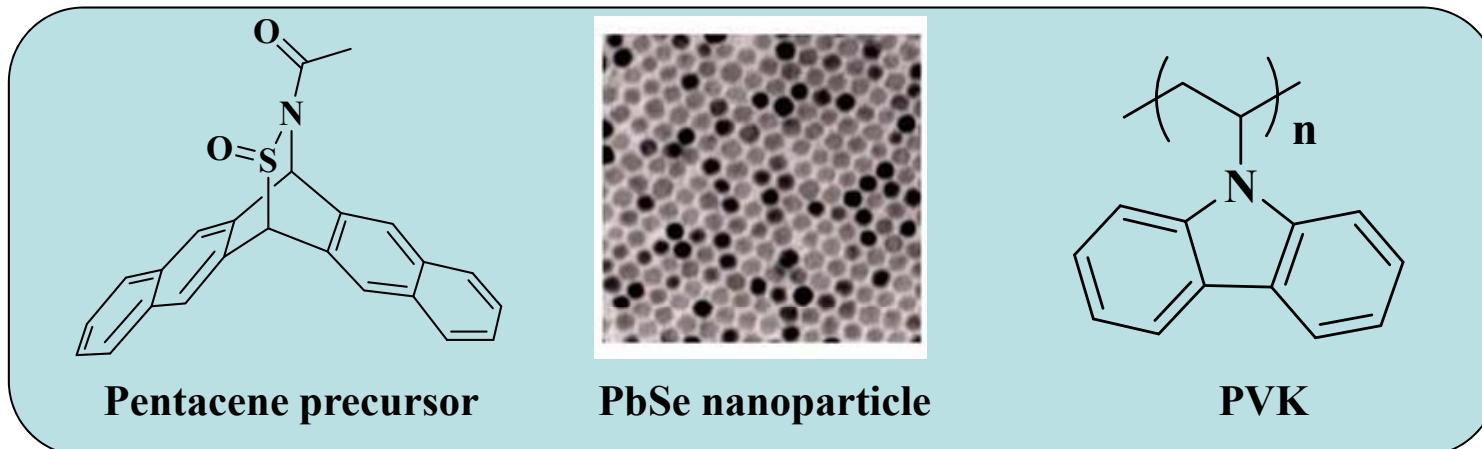
*N*-sulfinylacetamide



Pentacene precursor  
*“Soluble in CHCl<sub>3</sub>, CH<sub>2</sub>Cl<sub>2</sub>”*

Prepared by the Diels-Alder reaction between pentacene & N-sulfinylamide in the presence of a catalytic amount of methyltrioxorhenium,

# Device Processing for IR Photodetector



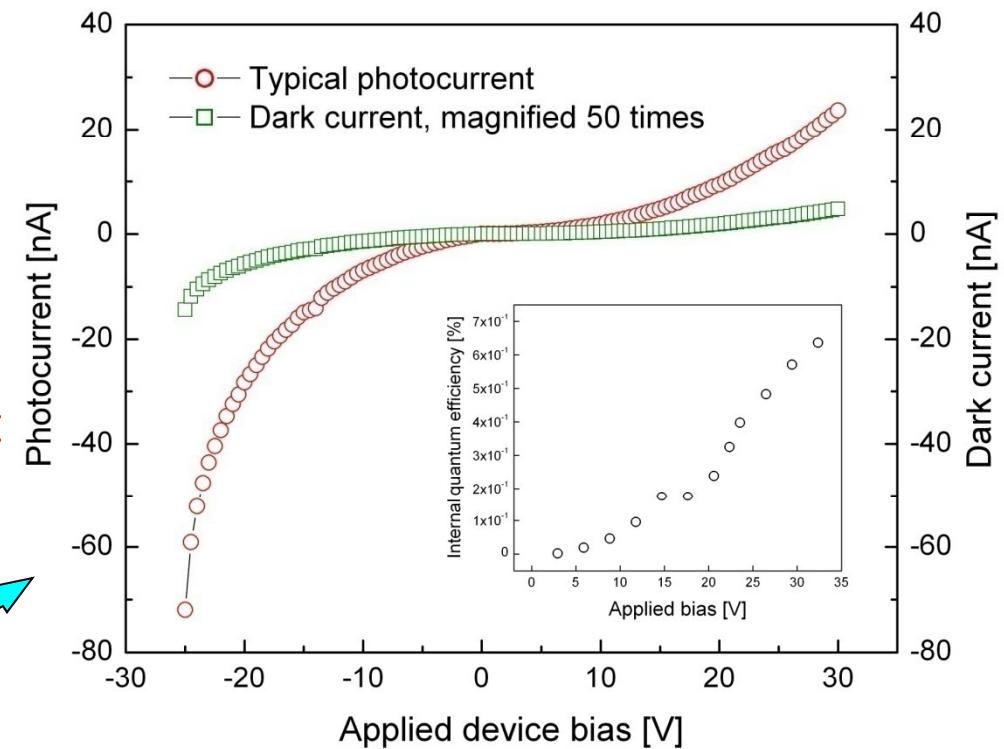
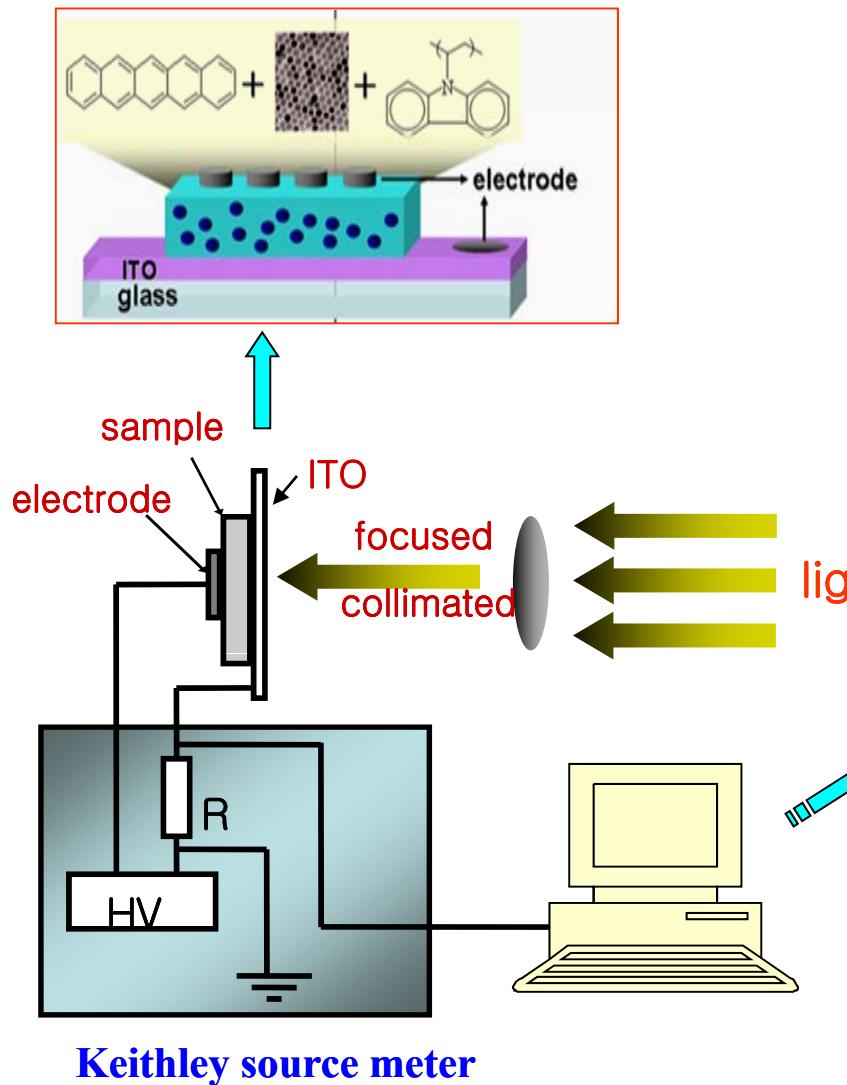
PVK-Pentacene precursor-PbSe Film

Baked in vacuum oven at 240 °C

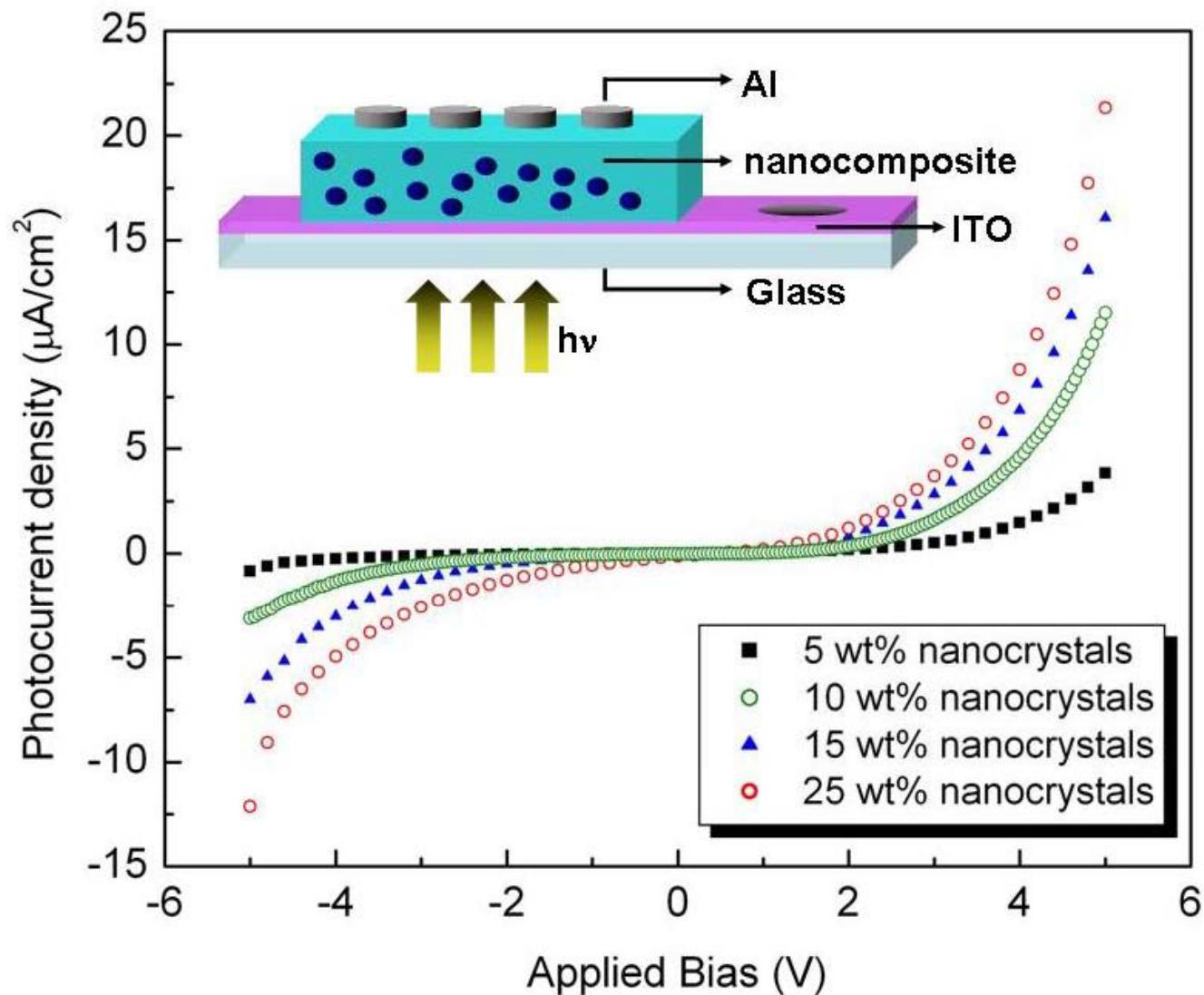
Measured Photoconductivity

**Film Composition:** PVK 30 wt%, Pentacene precursor 30 wt%, PbSe 40 wt%

# Evaluation of Photoconductivity

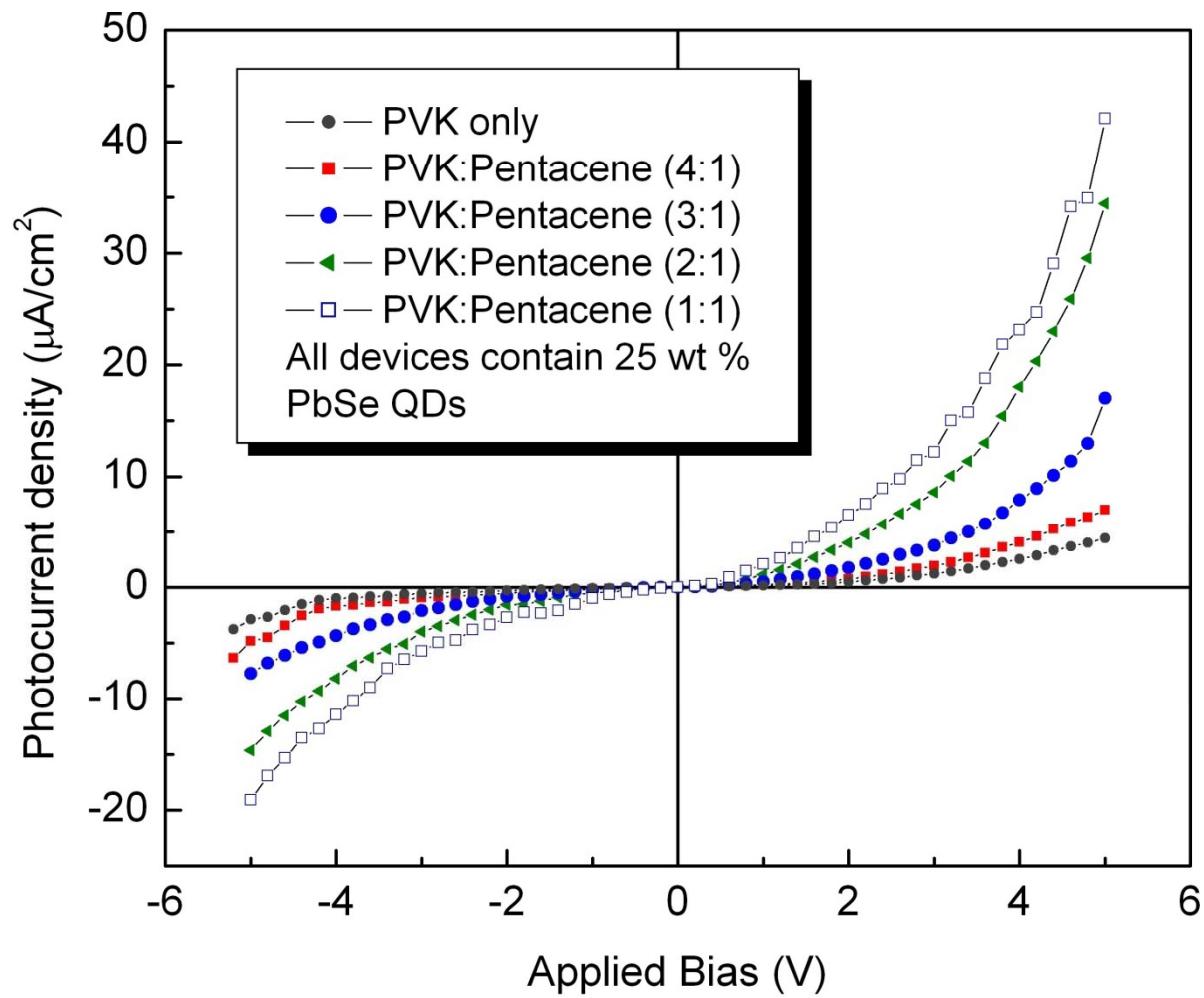


Typical I-V Curve



**Photocurrent density as a function of applied voltage in devices with the same proportion of PVK: pentacene (3:1) but various amounts of PbSe nanocrystals as indicated in the legend.**

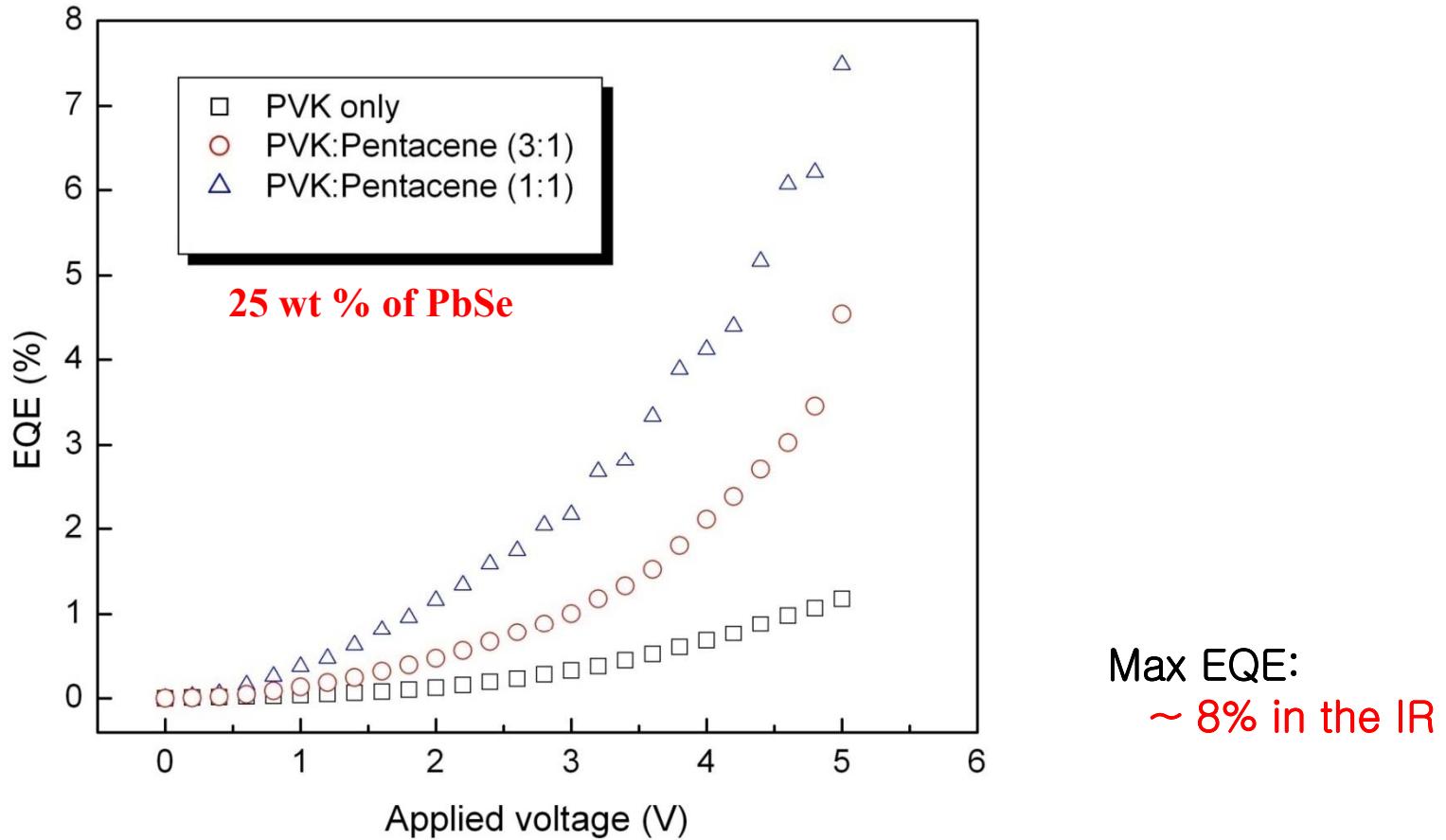
# Photocurrent density as a function of applied bias (1340 nm) in different devices with varying proportions of PVK and pentacene



The photocurrent increases significantly as the amount of pentacene in the composite increases. The best performance was extracted in devices with equal amounts of PVK and pentacene (having 25 wt % of PbSe QDs). The enhancement in photocurrent, compared to a PVK-PbSe film, is over 8 times.

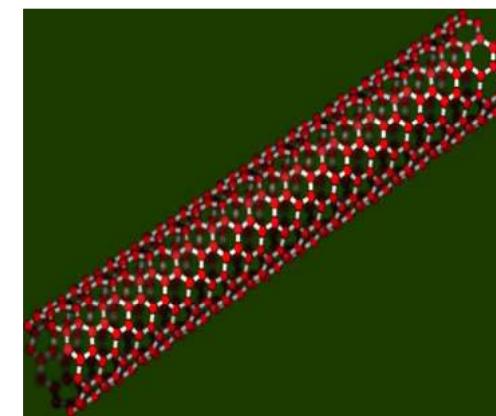
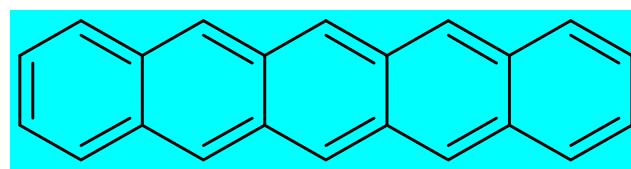
# Comparison of the external quantum efficiency of the composite devices

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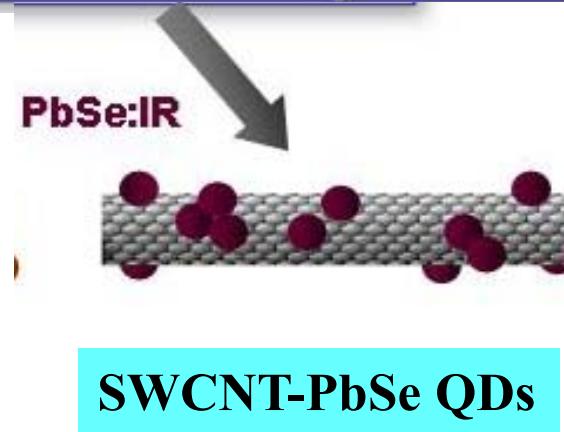
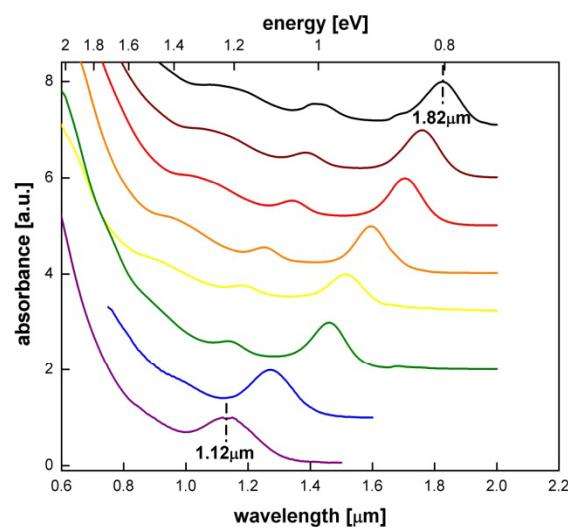
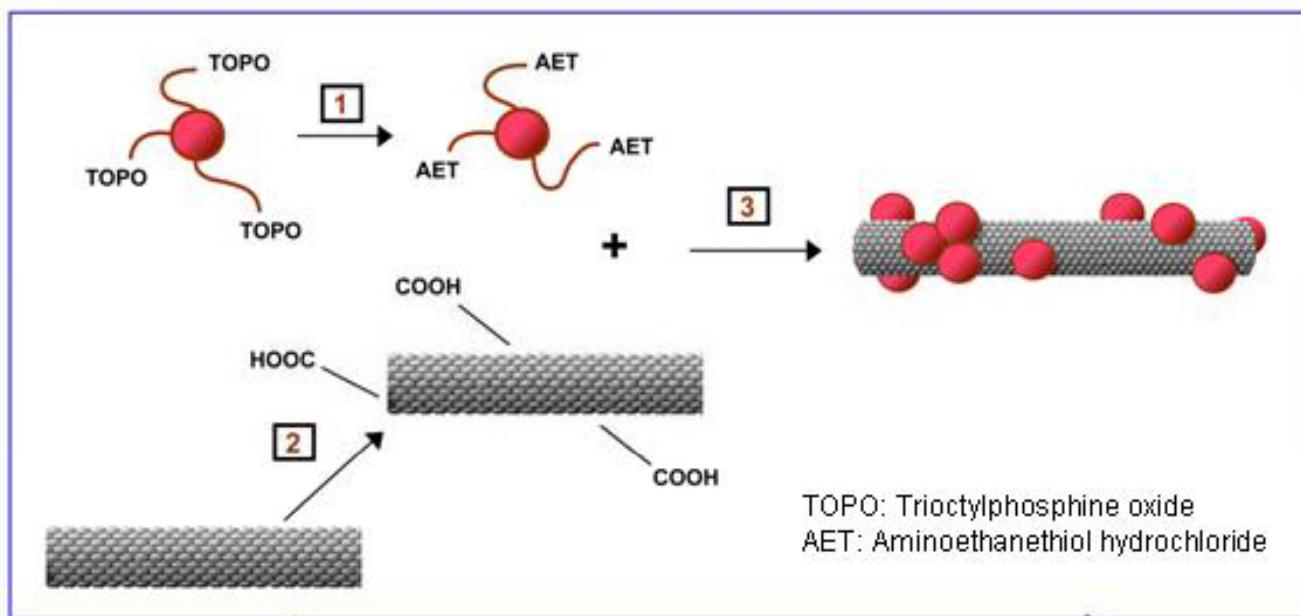


A maximum external quantum efficiency (EQE) of ~8% at an applied device bias of 5 V is achieved in the composite having equal amounts of PVK and pentacene. This is an improvement of eight times over the PVK:PbSe devices under similar experimental conditions.

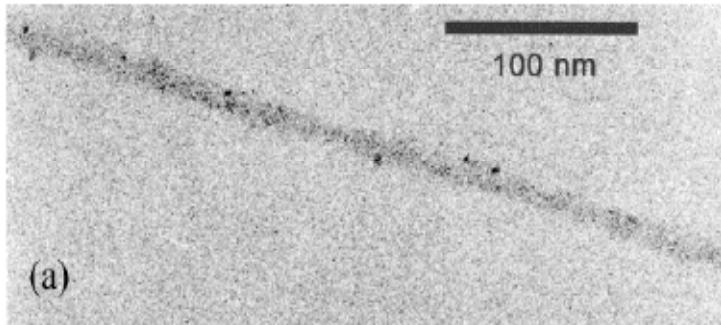
# **QD-Carbon Nanotube / PVK Polymeric Nanocomposite**



# Carbon Nanotubes Coupled with Quantum Dots

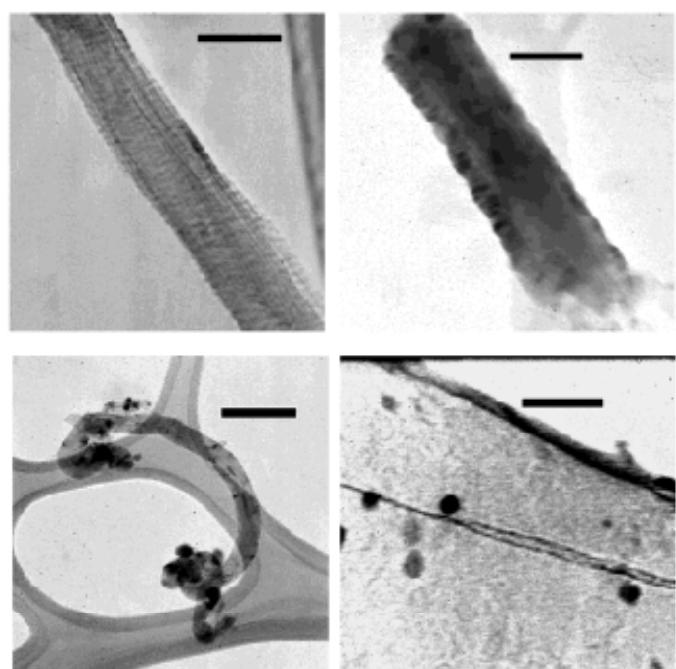


## SWCNT-CdSe



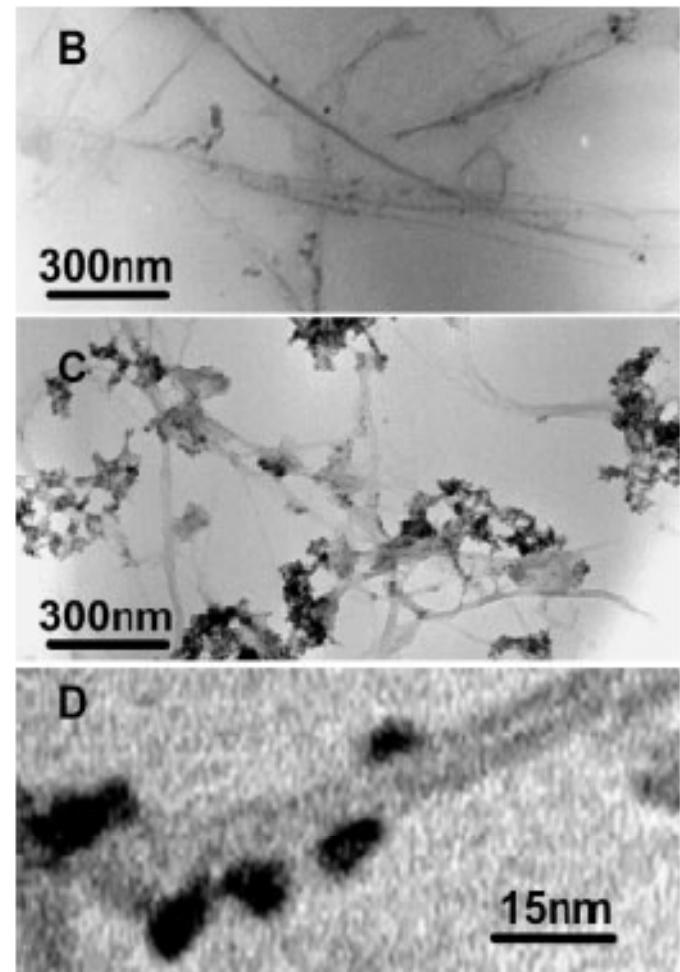
J. M. Haremza et al, Nano Lett. 2, 1253 (2002)

## SWCNT-CdSe



S. Banerjee et al, Nano Lett. 2, 195 (2002)

## SWCNT-CdS

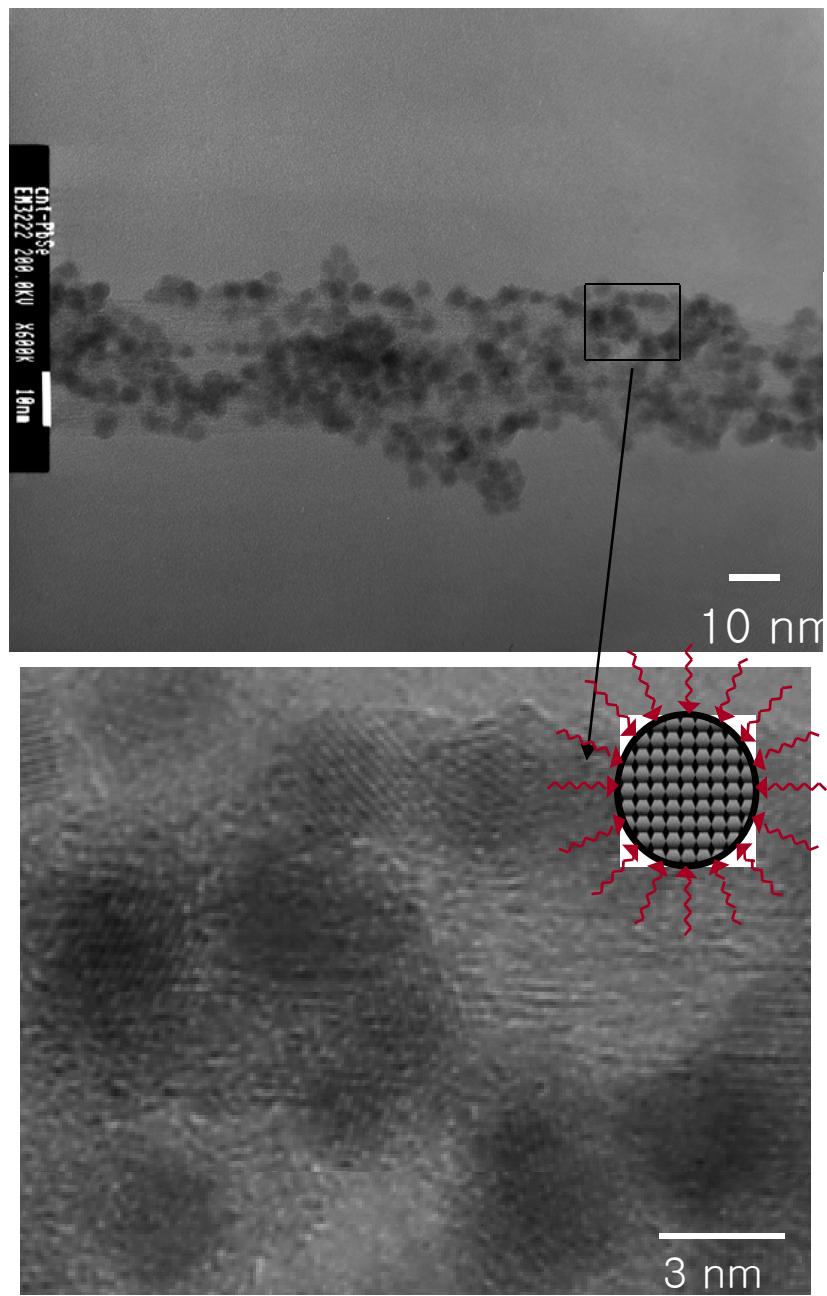
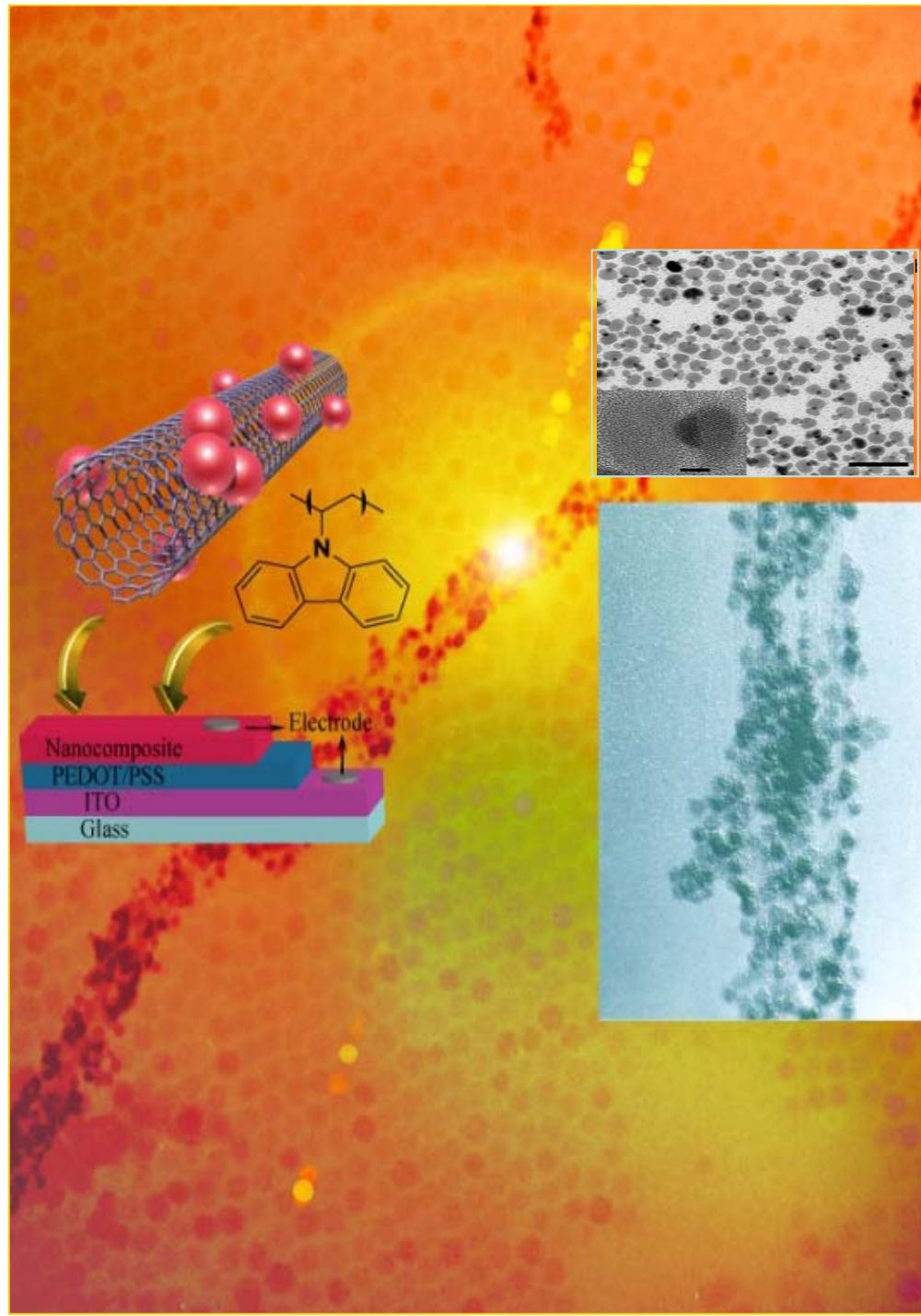


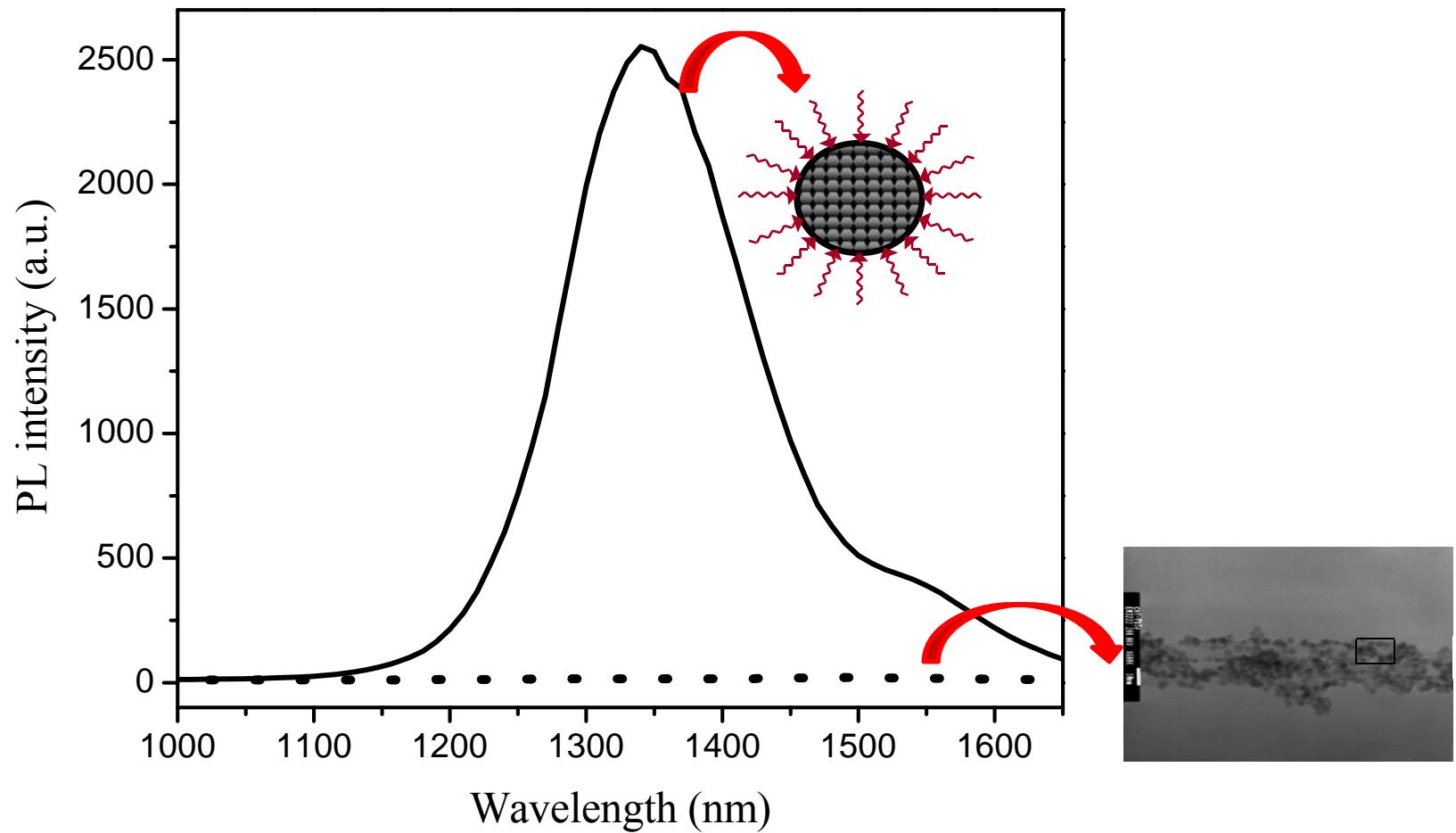
I. Robel et al,  
Adv. Mater. 17, 2458 (2005)

# QD-SWCNT-Polymer Nanocomposites for Photovoltaics



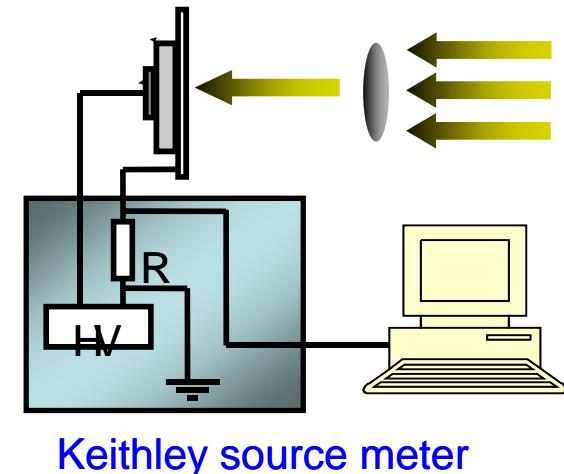
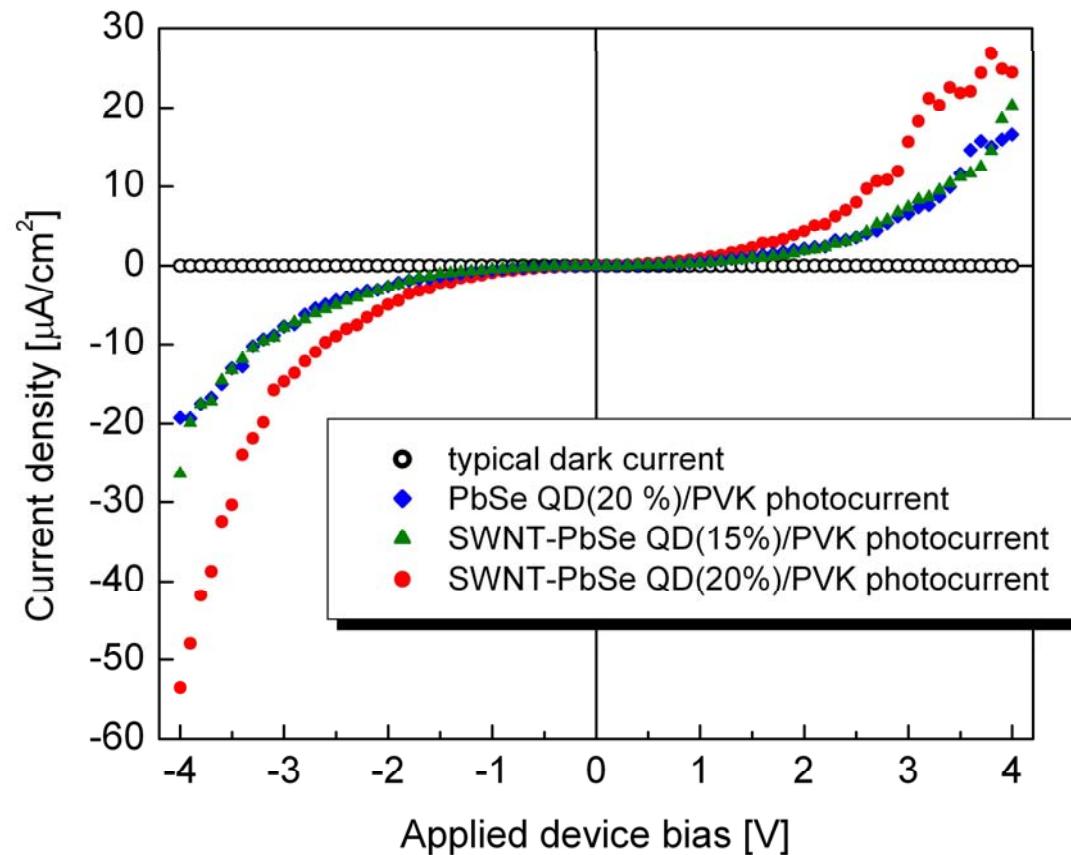
A variety of QDs active from UV to IR enables us to access a wide part of the solar spectrum



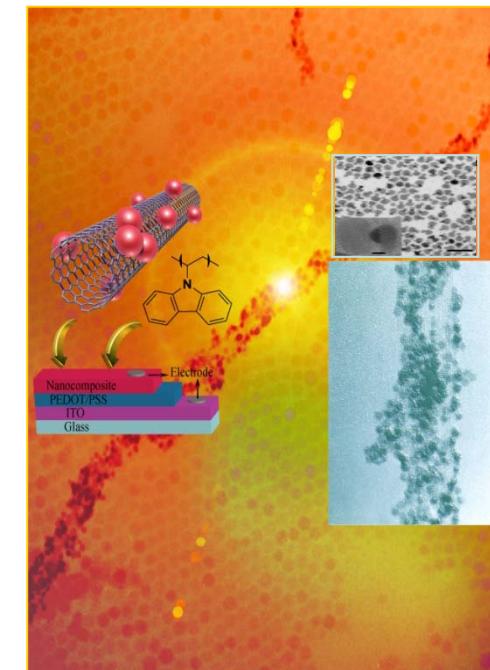


**PL spectra of PbSe QDs (solid line) and SWNT-PbSe (dotted line) in tetrachloroethylene colloidal suspensions. The concentration of PbSe QDs was the same in both cases.**

# **I-V Characteristics of PbSe QD/PVK and SWNT-PbSe /PVK Device in Dark and under Illumination**



Keithley source meter



*Adv. Mater., 19, 232-236 (2007) / US Patent 2010-0025662 A1, 2011  
ACS “Heart Cut” Research Highlight, 2007*

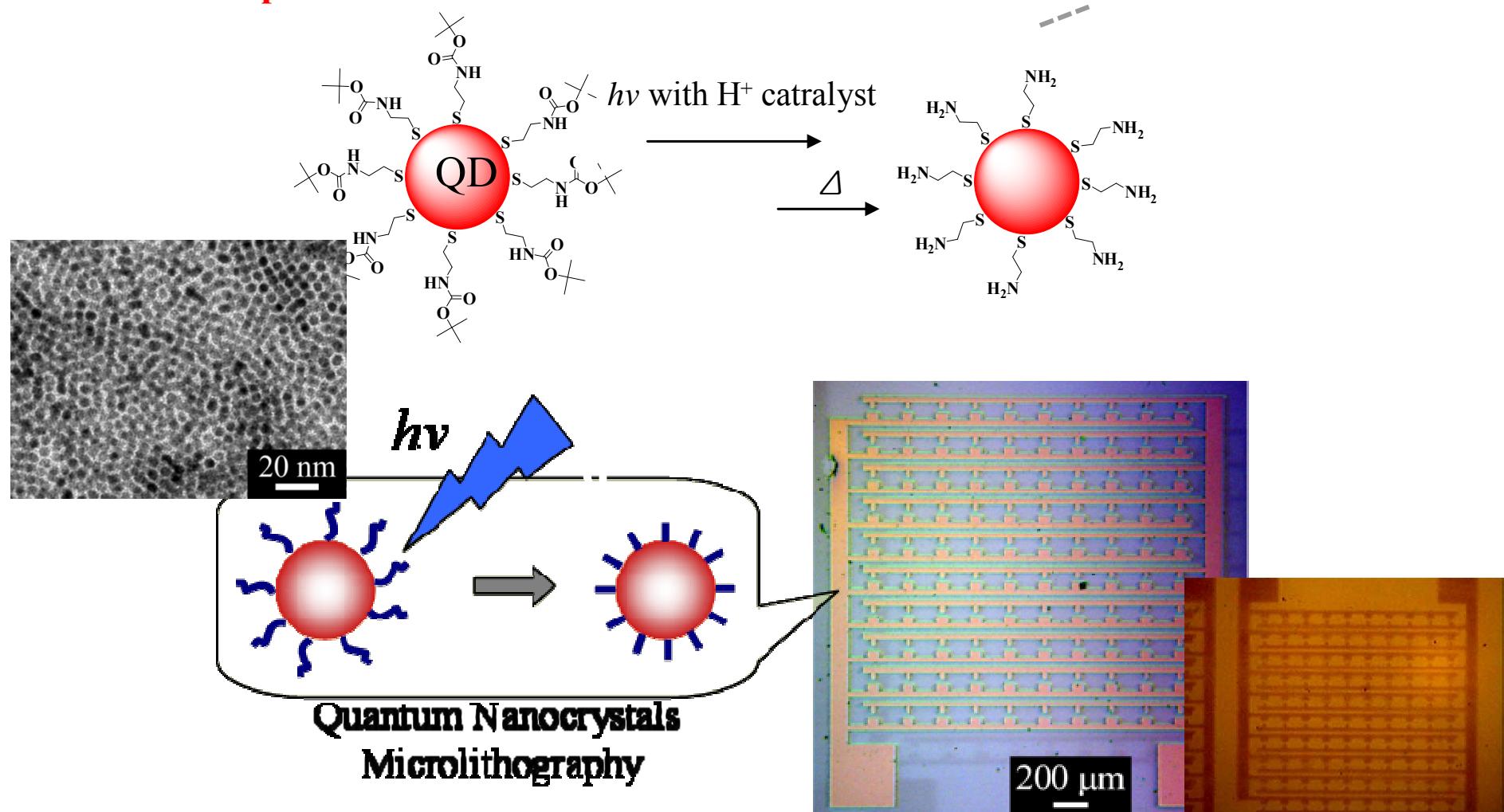
# Contents

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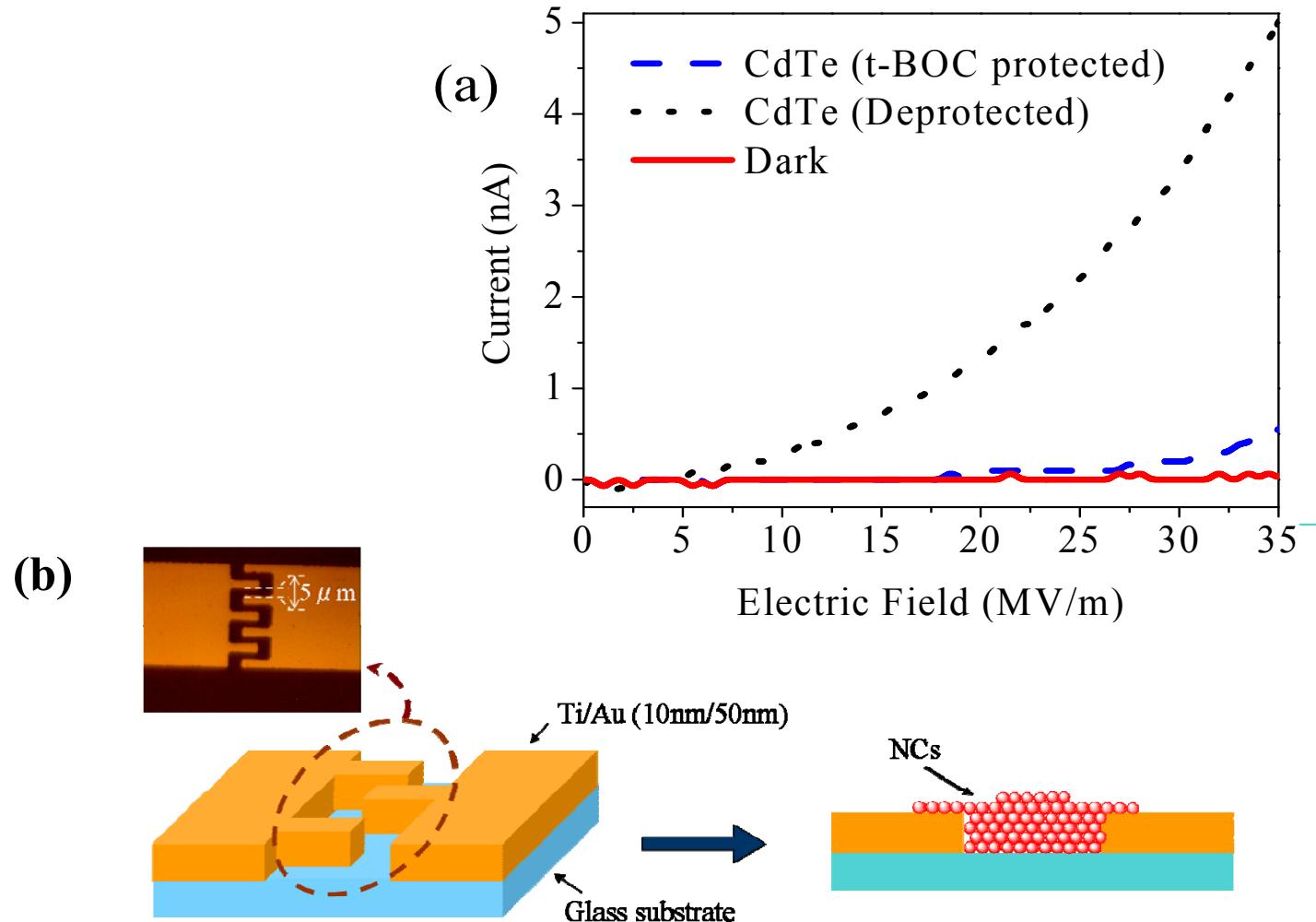
- Introduction
- Quantum Dot-based Hybrids
  - \* Enhancing the Photocurrent Density
  - \* Photopatternable Quantum Dots
- Low Bandgap Polymers
- C60 Derivatives
- Summary

# Surface Functionalization for Photo-Patternable QDs

## Chemical Amplification Reaction

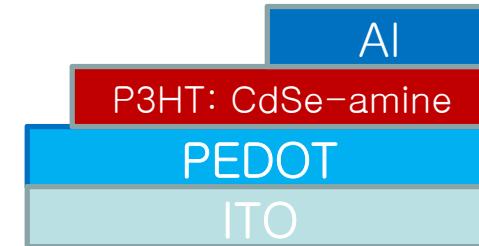
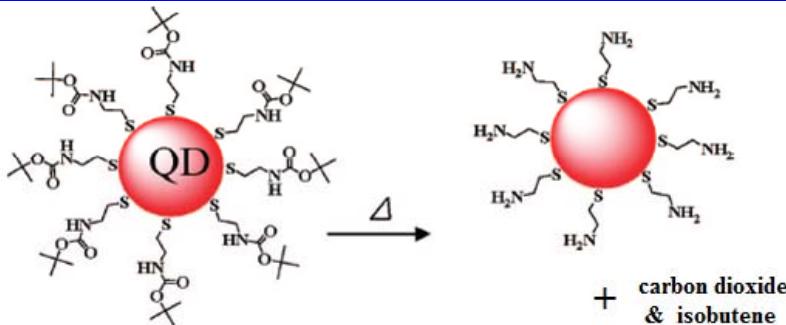


# Photocurrent Measurements of Photopatternable QDs



Current-voltage curves (a) in the dark or with white light ( $100 \text{ mW/cm}^2$ ) illumination for a film of *t*-BOC protected and deprotected CdTe nanocrystals (measured at the voltage scan rate is  $1 \text{ V/s}$ ). The channel length is  $5 \mu\text{m}$ . MSM device structure (b) for photoconductivity measurement.

# Polymer Nanocomposite Photovoltaics utilizing CdSe QDs Capped with a Thermally Cleavable Solubilizing Ligand



**Solar Cell Device Structure**

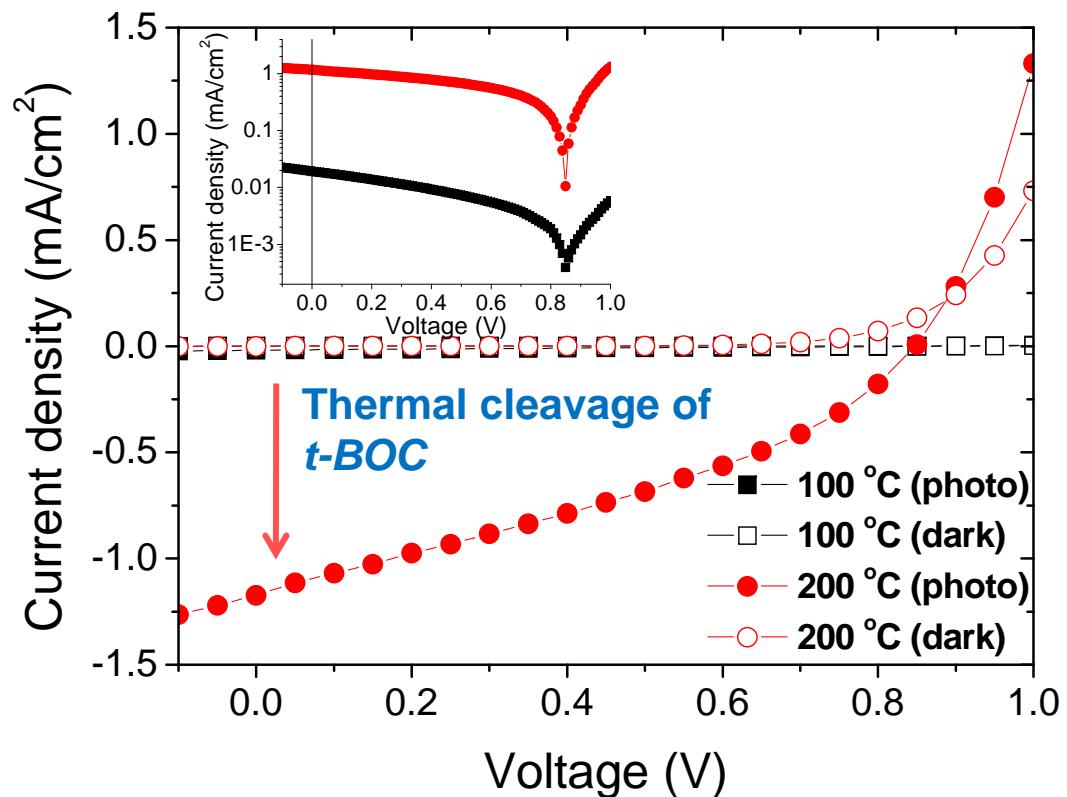
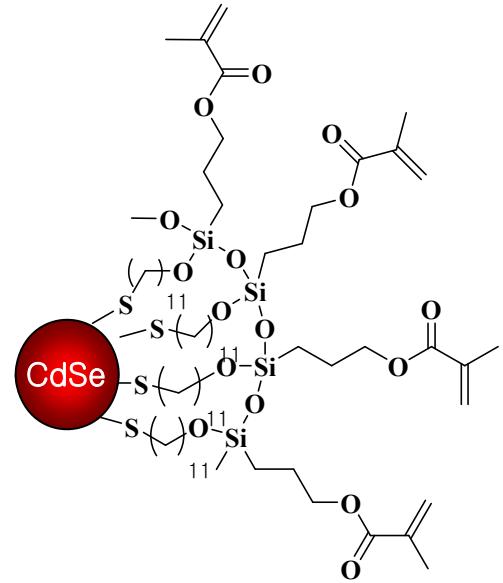


TABLE I. Summary of device performance in P3HT:CdSe-*t*Boc (10:90 wt %) blend film with different heating temperatures for *t*BOC deprotection

Temperature ( $^\circ\text{C}$ )	$V_{\text{oc}}$ (V)	$J_{\text{sc}}$ ( $\text{mA}/\text{cm}^2$ )	FF	PCE (%)
150	0.89	0.75	0.29	0.21
175	0.87	0.82	0.31	0.25
200	0.85	1.17	0.35	0.38
220	0.82	1.16	0.37	0.39
240	0.76	1.43	0.37	0.44
260	0.73	1.53	0.33	0.42
300	0.50	1.31	0.29	0.21



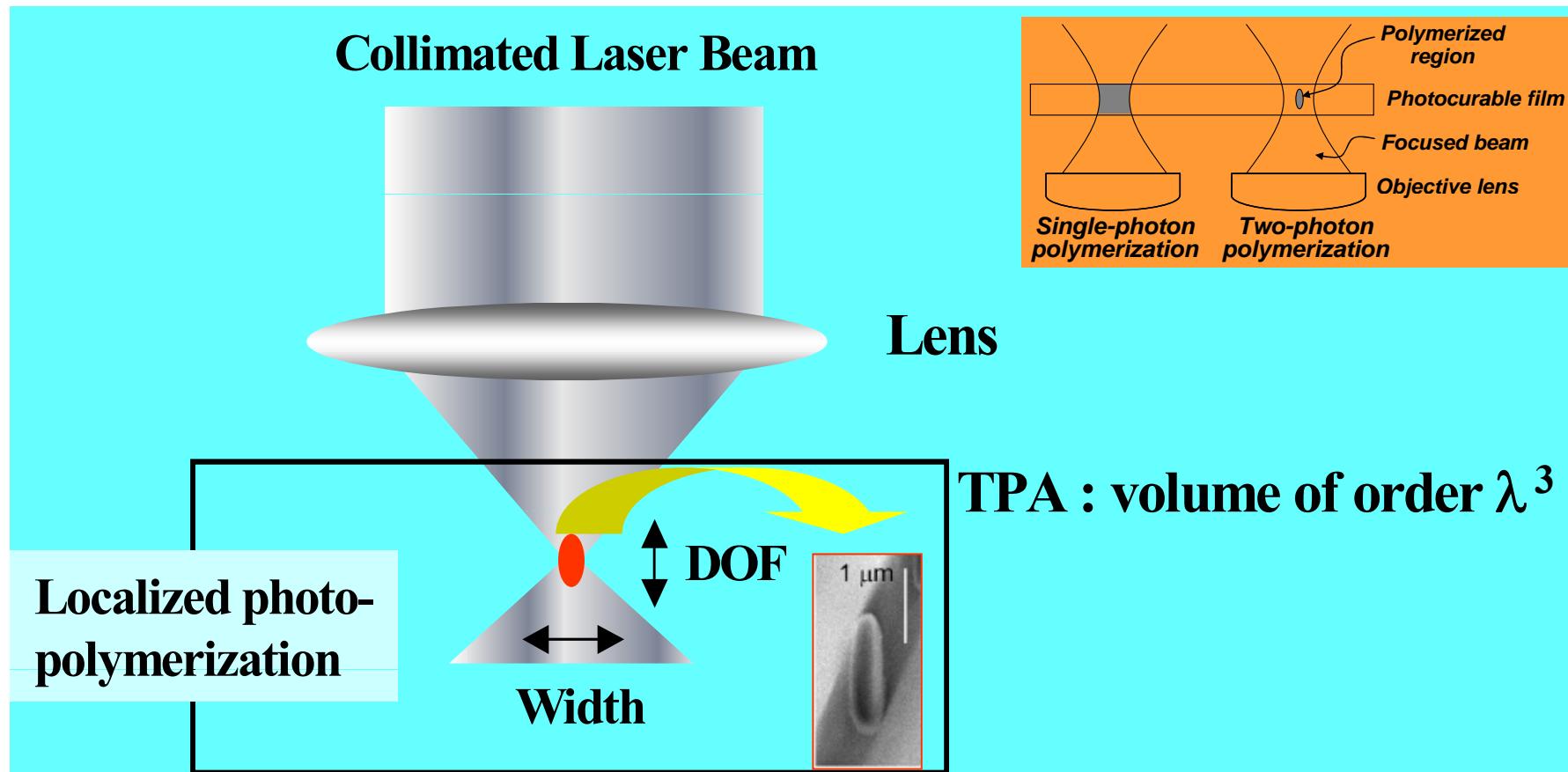
QD 1

UV: 552nm

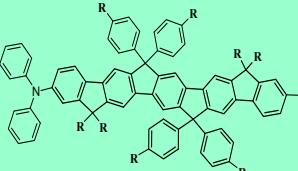
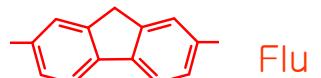
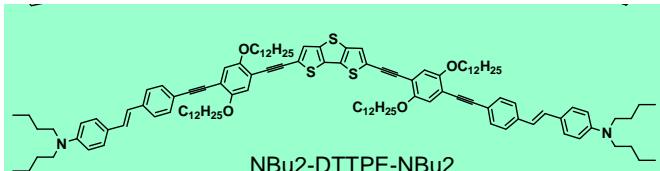
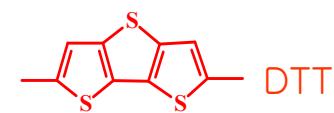
PL: 568nm

# 3-D Lithographic Microfabrication

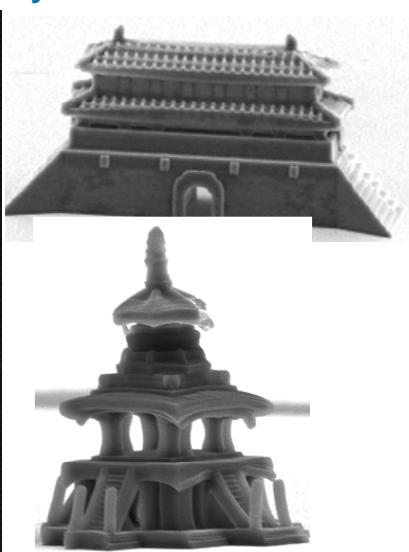
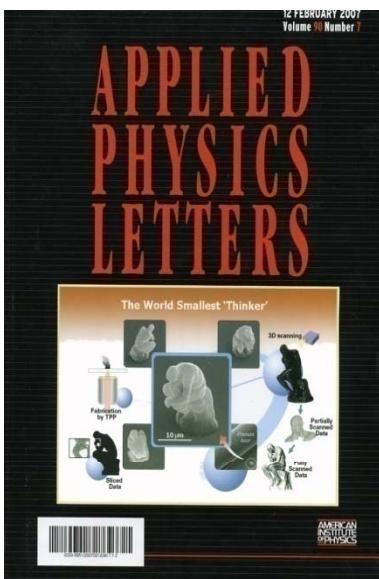
## ◆ Localized photochemistry by TPA process



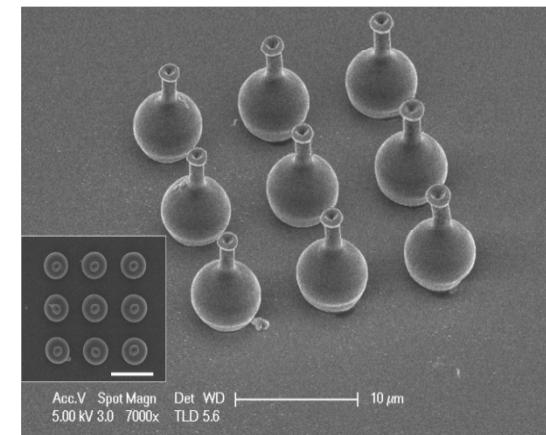
Use of longer radiation wavelengths in TPA process is possible to achieve the better penetration depth in the medium



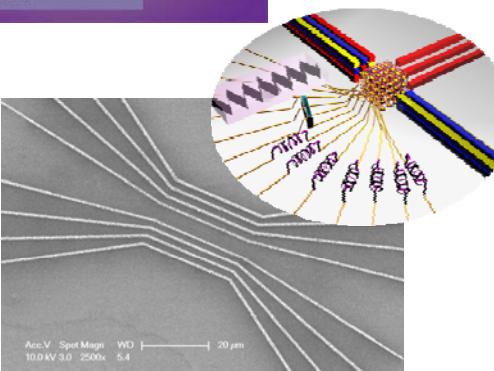
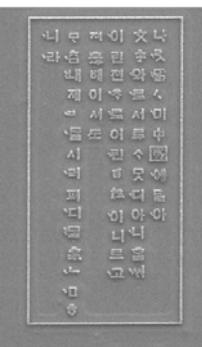
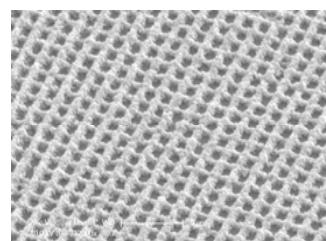
## Polymers



## Ceramics



## Metals



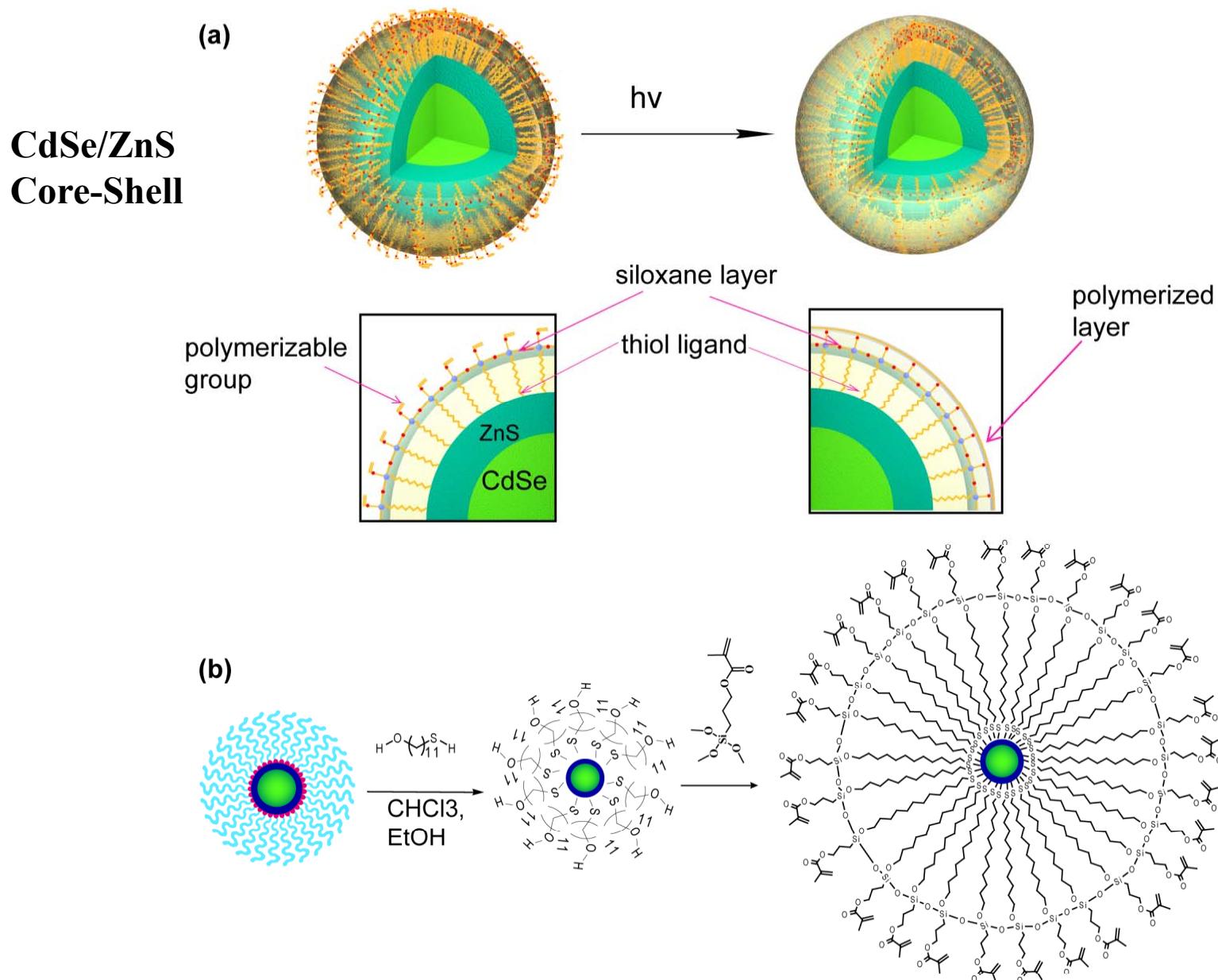
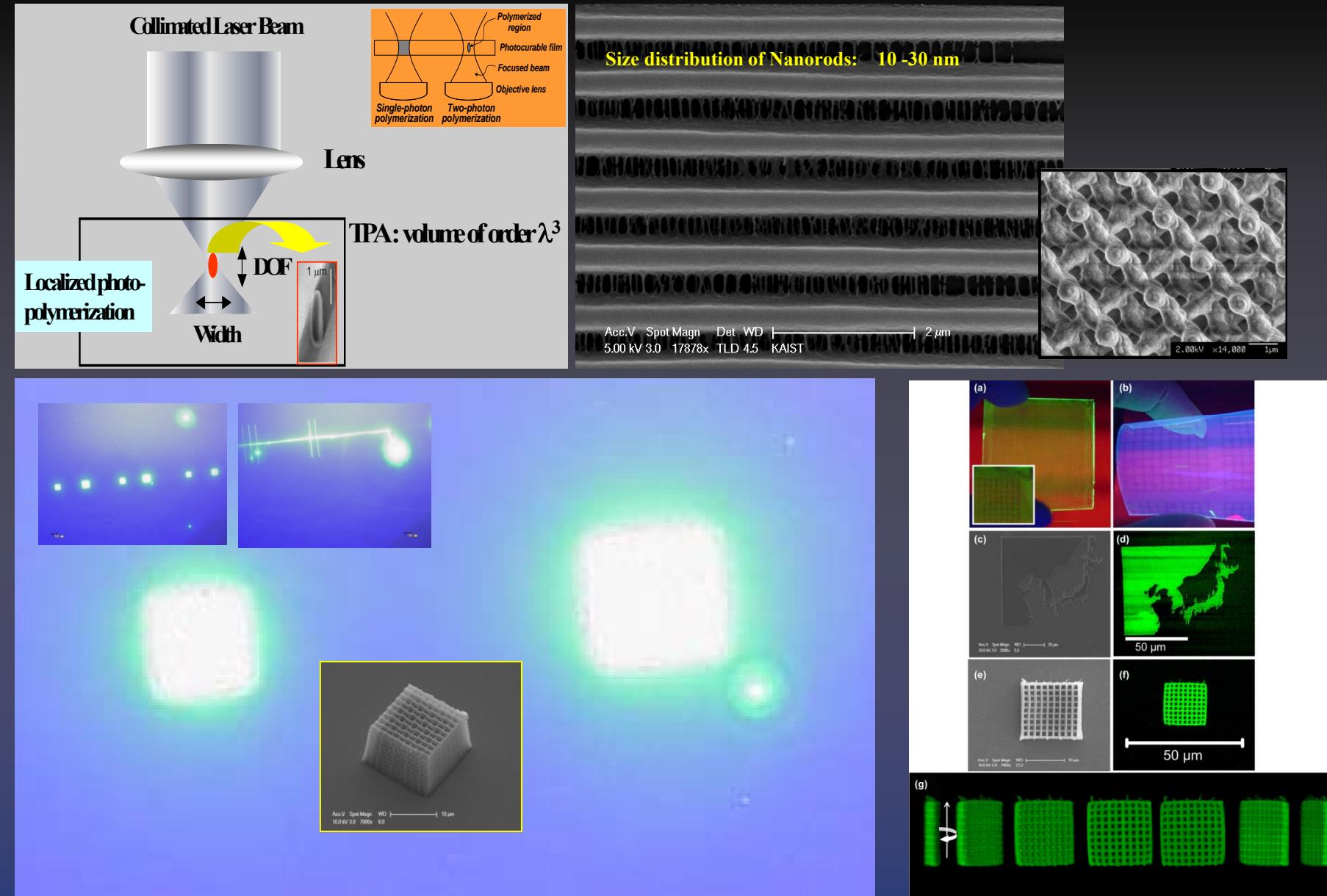
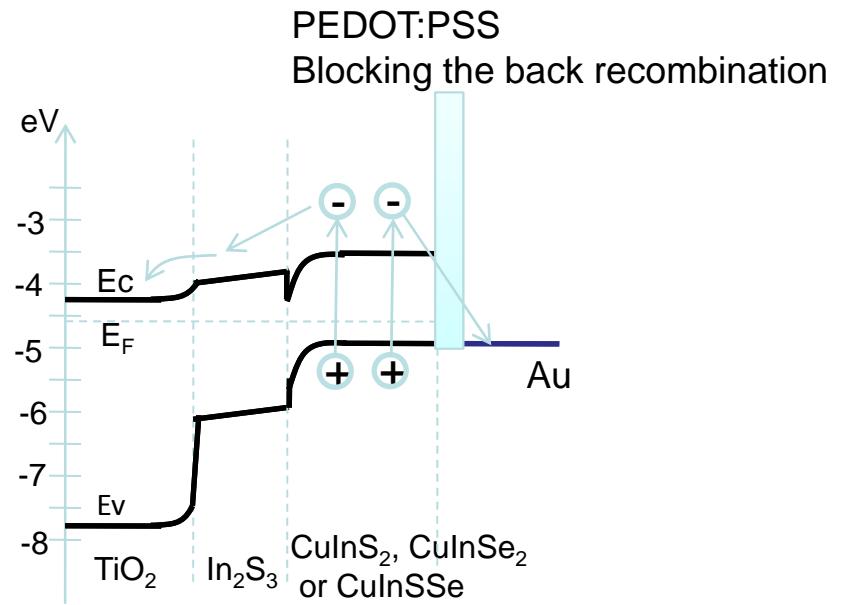
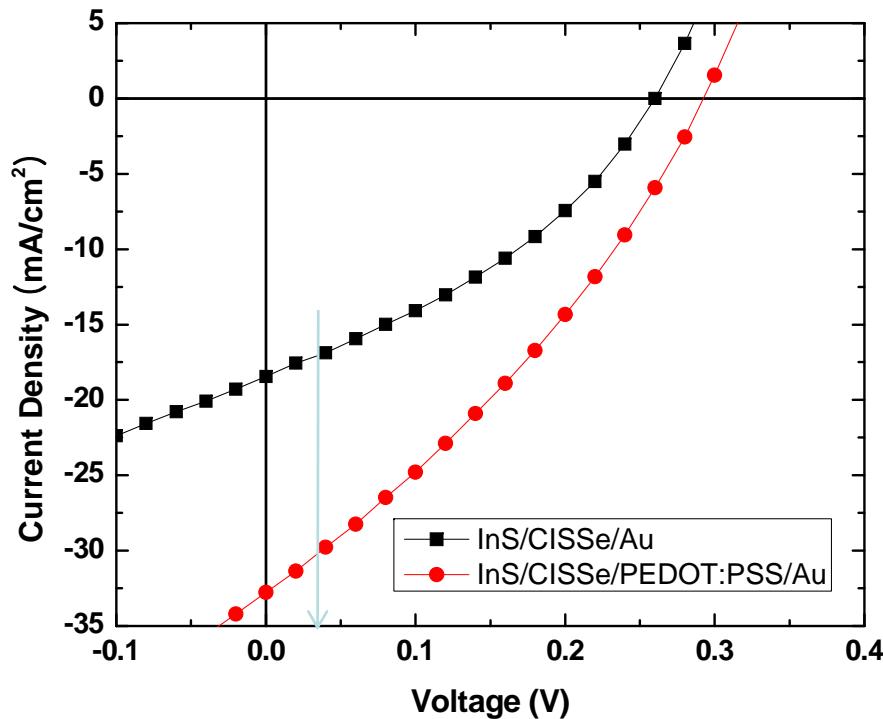


Figure (a) shows the photocuring of working acrylate functionalized quantum dots with its layer structure, (b) Shows the synthesis scheme for the acrylate functionalized photopatternable quantum dots.

# Photopatterns by Using CdSe/ZnS Core-Shell QDs



# Polymer Blocking Layer Effect on $\text{In}_2\text{S}_3/\text{CuInSSe}$ Solar Cell

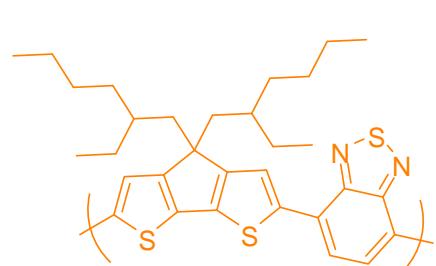


	V <sub>oc</sub> (V)	J <sub>sc</sub> (mA/cm <sup>2</sup> )	FF(%)	PCE(%)
w/o PEDOT:PSS	~0.26	18.4	35.3	1.69
with PEDOT:PSS	~0.29	33.5	31.9	3.16

**V<sub>oc</sub> is similar**  
**J<sub>sc</sub> shows significant increase**  
**FF is decreased**  
**PCE increases due to the J<sub>sc</sub> increment**

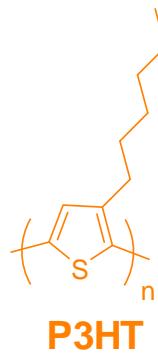
**→ PEDOT:PSS effectively blocks  
electrons & transports holes**

# Low-Bandgap Conjugated Polymers



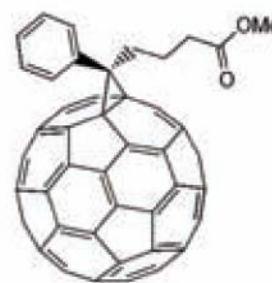
**PCPDTBT**

D-A concept

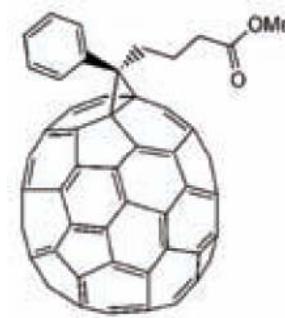


**P3HT**

Rigiospecific Structure



**PCBM**



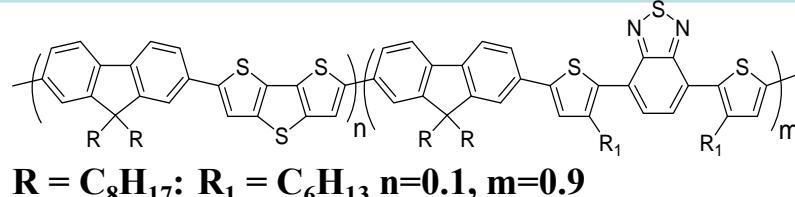
**PC<sub>70</sub>BM**

Optical Bandgap: 1.52 eV

1.90 eV

# Low Bandgap Conjugated Copolymers: PFTB & PCTB

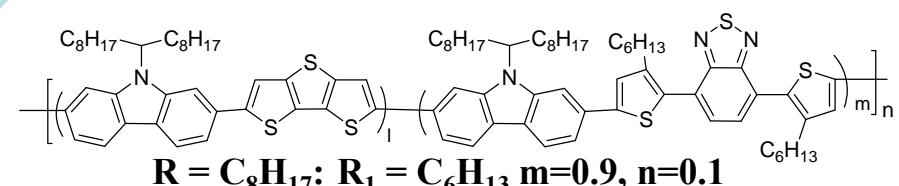
## PFTB



Abs:  $\lambda_{\max}$   
521 nm  
519 nm (with PCBM 1:4)

PL :  $\lambda_{\max}$   
642 nm  
642 nm (with PCBM 1:4)

## PCTB

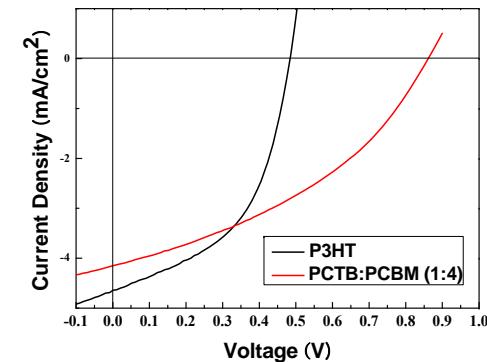
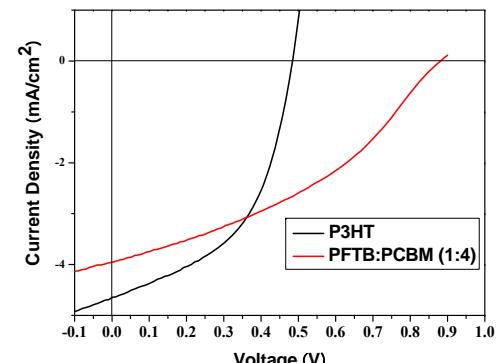


Abs:  $\lambda_{\max}$   
521 nm  
519 nm (with PCBM 1:4)

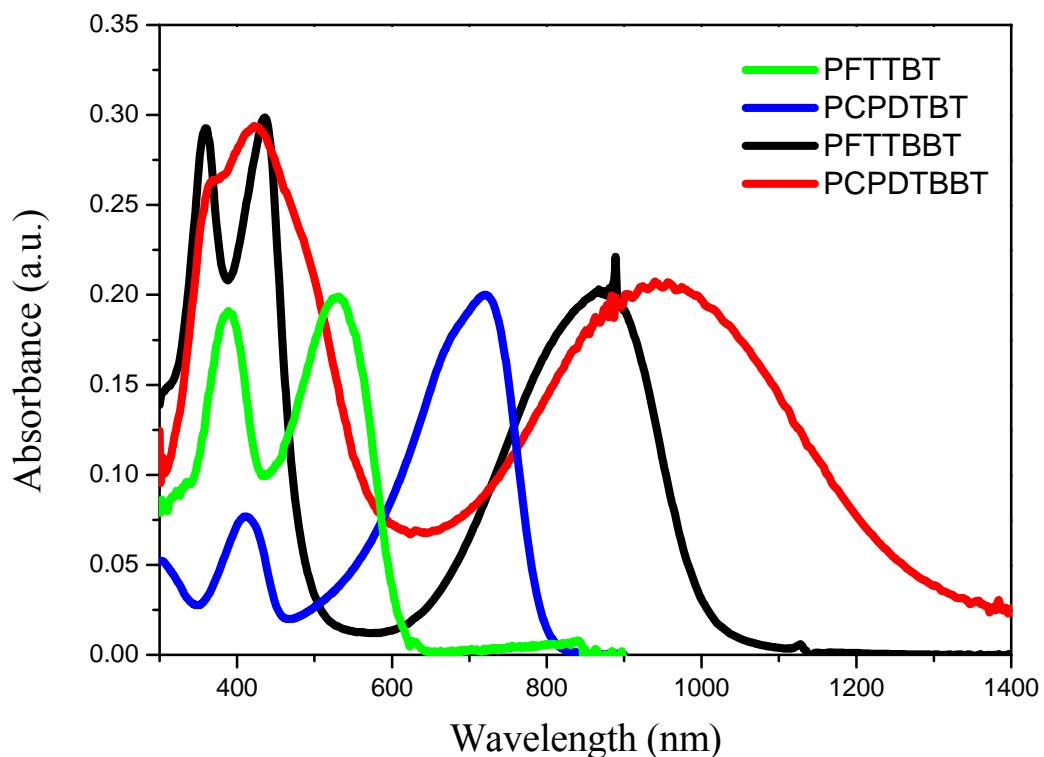
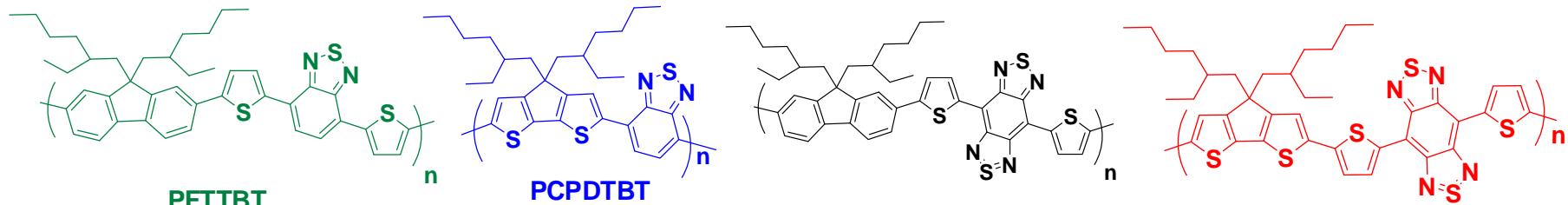
PL :  $\lambda_{\max}$   
642 nm  
643 nm (with PCBM 1:4)

## I-V Characteristics and Solar Cell Efficiency

Device	$J_{sc}$ (mA/cm <sup>2</sup> )	$V_{oc}$ (V)	$FF$ (%)	Solar Efficiency (%)
P3HT/PCBM (1:0.8)	4.234	0.485	49.68	1.02
<b>PFTB/PCBM (1:4)</b>	3.596	0.881	37.64	<b>1.19</b>
<b>PCTB/PCBM (1:4)</b>	<b>3.782</b>	<b>0.862</b>	<b>38.74</b>	<b>1.26</b>



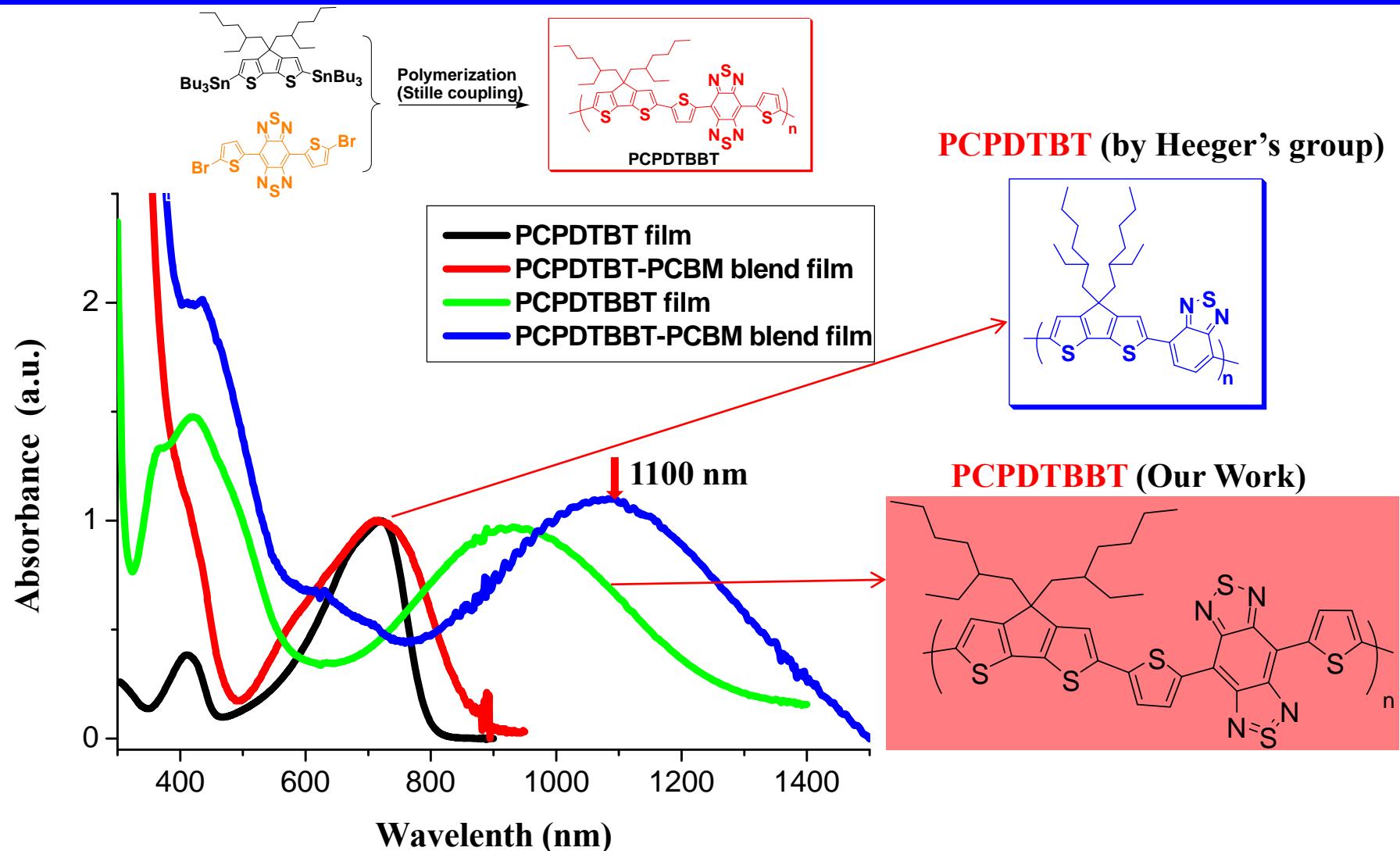
# Development of Efficient NIR Polymers; Synthesis of the Low bandgap Polymers



Optical properties of various polymers

Polymer	Abs. $\lambda_{\text{max.}}$	Opt. band gap
PFTTB BT	530 nm	1.98 eV
PCPDTBT	718 nm	1.52 eV
PFTTB BT	868 nm	1.19 eV
PCPDTBBT	950 nm	0.92 eV

# Comparison of UV-vis Absorption of Low Bandgap Polymers



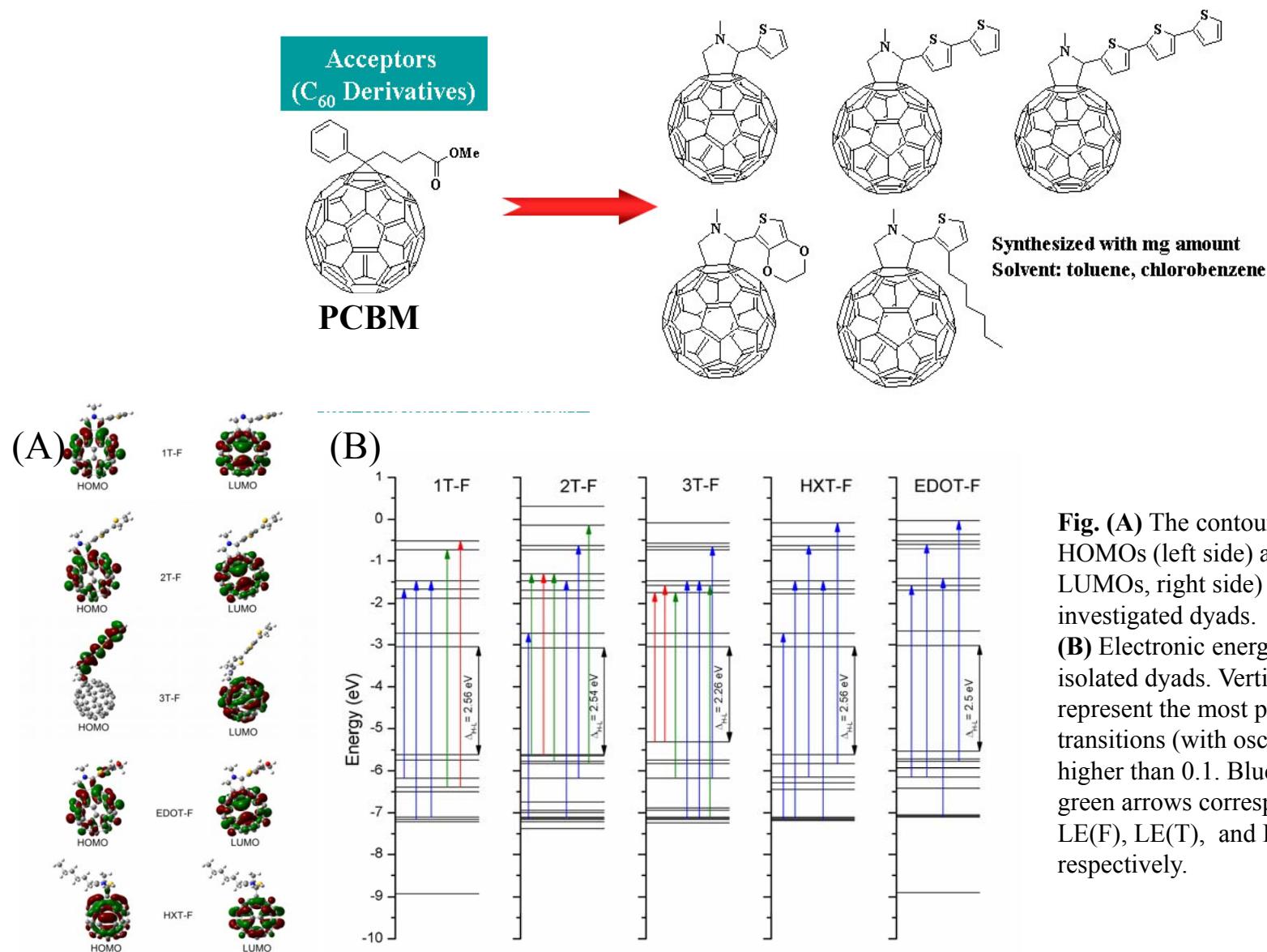
UV-vis absorption spectra of the PCPDTBT and PCPDTBTT Polymer films and the polymer-PCBM blend films

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- Introduction
- Quantum Dot-based Hybrids
  - \* Enhancing the Photocurrent Density
  - \* Photopatternable Quantum Dots
- Low Bandgap Polymers
- C<sub>60</sub> Derivatives
- Summary

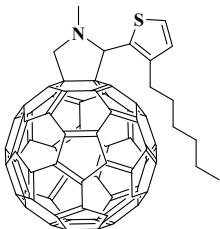
# C<sub>60</sub> Derivatives as Acceptors in Hybrid Bulk Heterojunction Solar Cells



**Fig. (A)** The contour plots of the HOMOs (left side) and the LUMOs, right side) of all investigated dyads.

**(B)** Electronic energy levels of an isolated dyads. Vertical arrows represent the most probable transitions (with oscillator strength higher than 0.1. Blue, red, and green arrows corresponds to the LE(F), LE(T), and ICT transitions, respectively.

# Electron Mobility of New n-Type C60 Derivative (Fu-Hexyl)

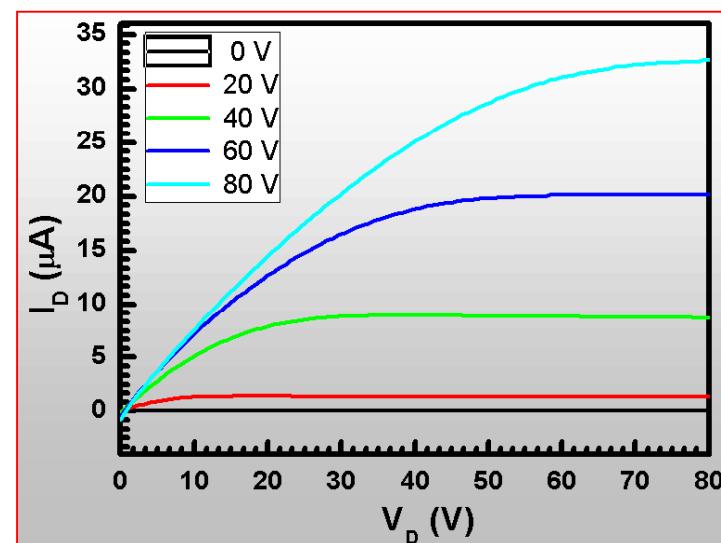
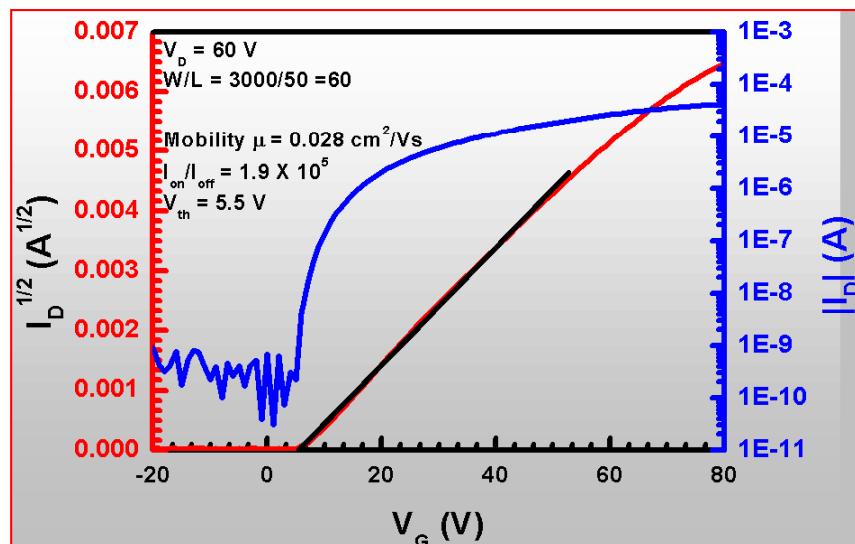
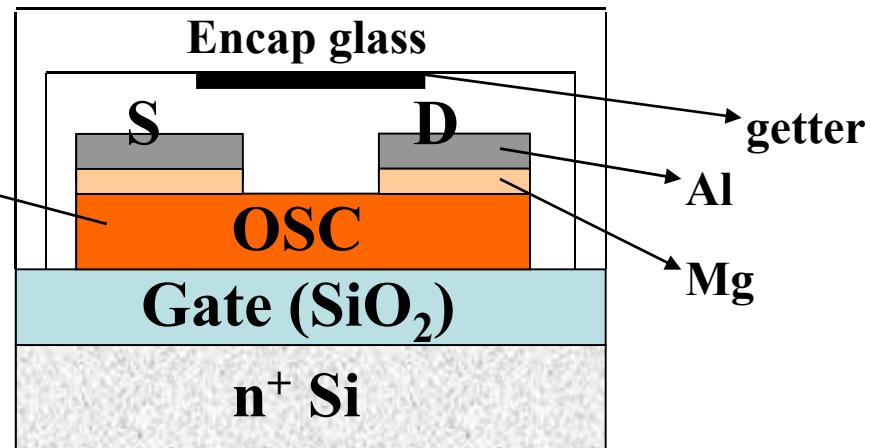


Fu-Hexyl

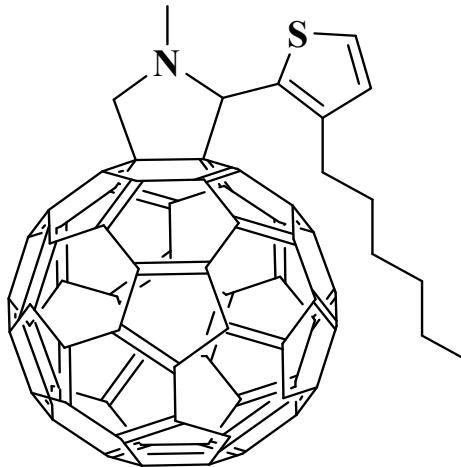
Inkjet printing  
(0.5wt% **Fu-Hexyl** in  
chlorobenzene)

$$\mu = 0.028 \text{ cm}^2/\text{V}\cdot\text{s} \quad (0.0058 \text{ cm}^2/\text{V}\cdot\text{s} \text{ for PCBM})$$

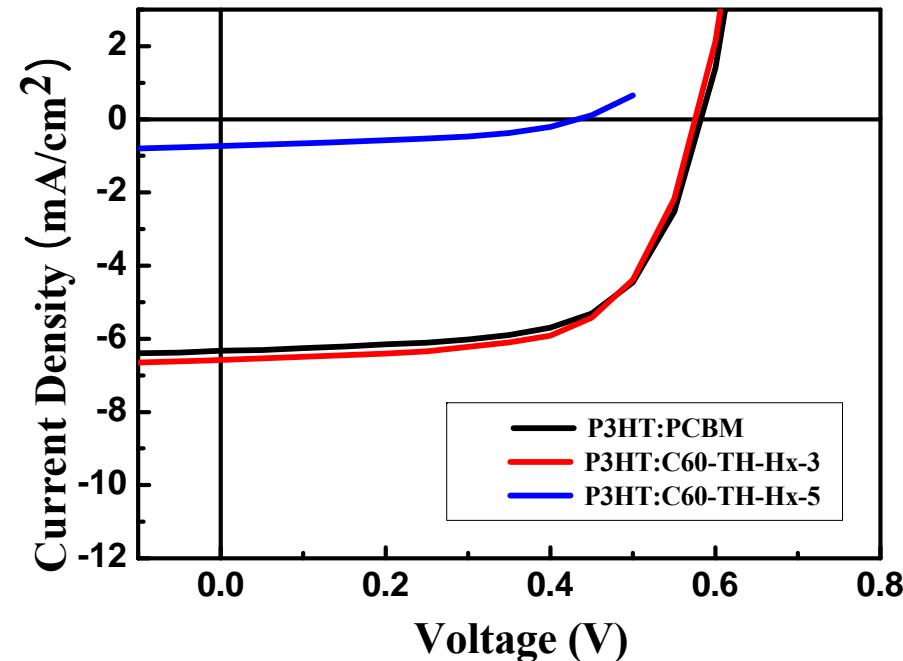
On-off ratio =  $1.9 \times 10^5$



# Efficiency of ITO/PEDOT:PSS/P3HT:C60 Derivative/TiO<sub>x</sub>/Al Solar Cells

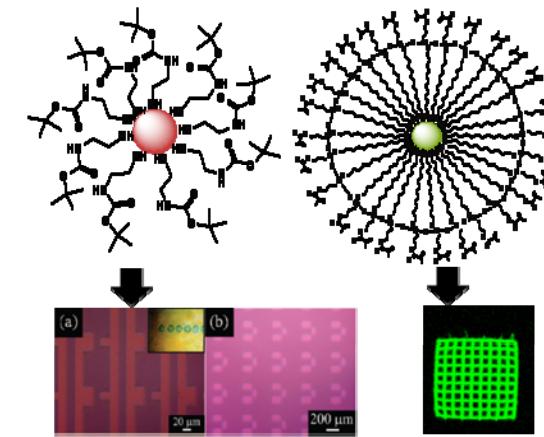
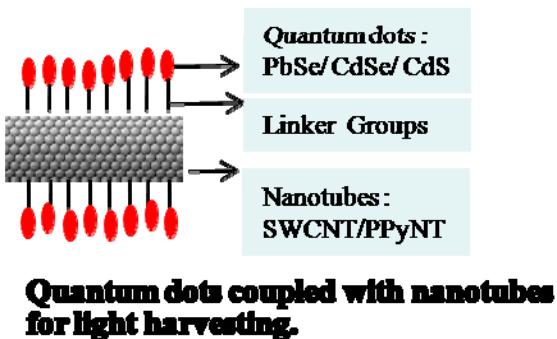
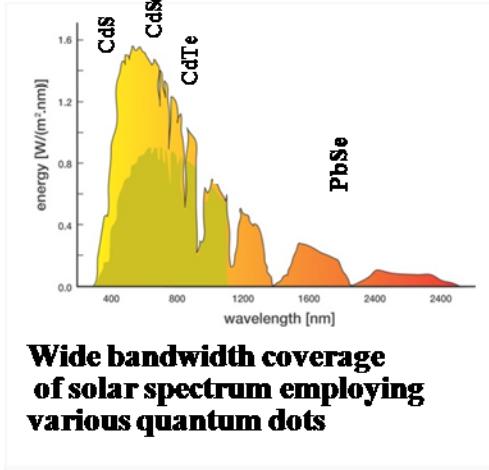


**Fu-Hexyl**

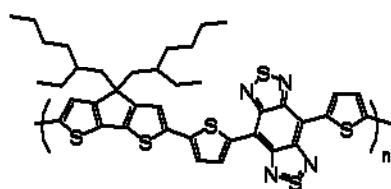


Compounds	$J_{sc}$ (mA/cm <sup>2</sup> )	$V_{oc}$ (V)	FF	E <sub>ff</sub> (%)
P3HT:PCBM	6.32	0.582	0.65	2.39
P3HT:C60-TH-Hx-3	6.58	0.575	0.64	2.44

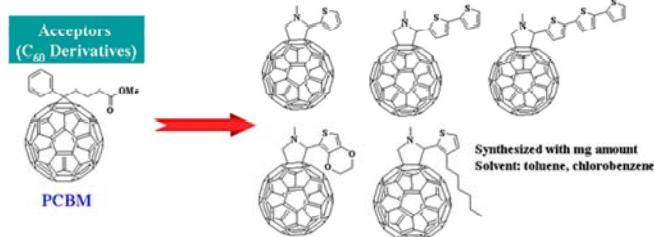
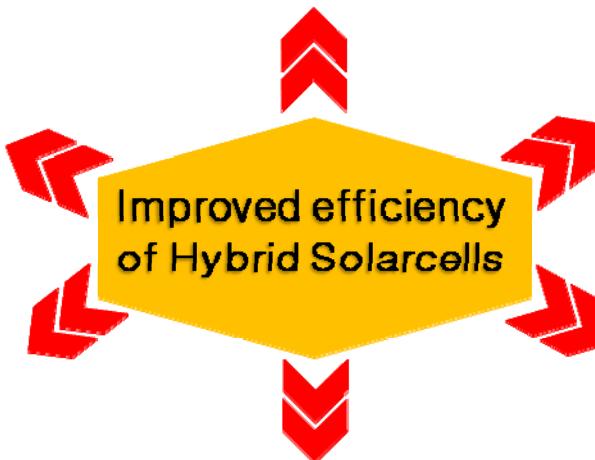
# Graphic Summary



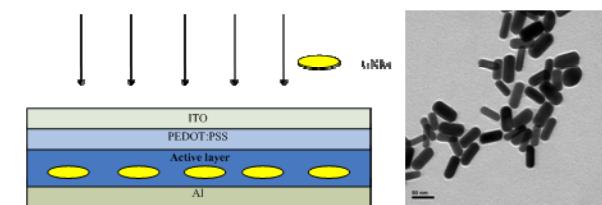
Surface modification of QDs for devices with photopatterned active layers.



Low bandgap polymers for light harvesting in NIR region of solar spectrum



New fullerene derivatives as electron acceptors in bulk heterojunction solar cells



Gold nanorod generated surface plasmon enhanced light harvesting in hybrid photovoltaics

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