

Artificial Photosynthesis – from Light Absorption to Solar Fuels



Devens Gust
*Department of Chemistry and
Biochemistry
Arizona State University
Tempe, AZ 85287-1604 USA*

**ASU Center for Bioenergy and
Photosynthesis**

**ASU Center for Bio-Inspired Solar
Fuel Production**

ASU® ARIZONA STATE
UNIVERSITY

Best current “technology” for solar energy conversion- photosynthesis



- The source of most of our current energy – fossil fuels
- 10^{14} kg of carbon removed from the atmosphere each year and converted to fuel
- 10^{21} Joules of energy each year

~125 TW

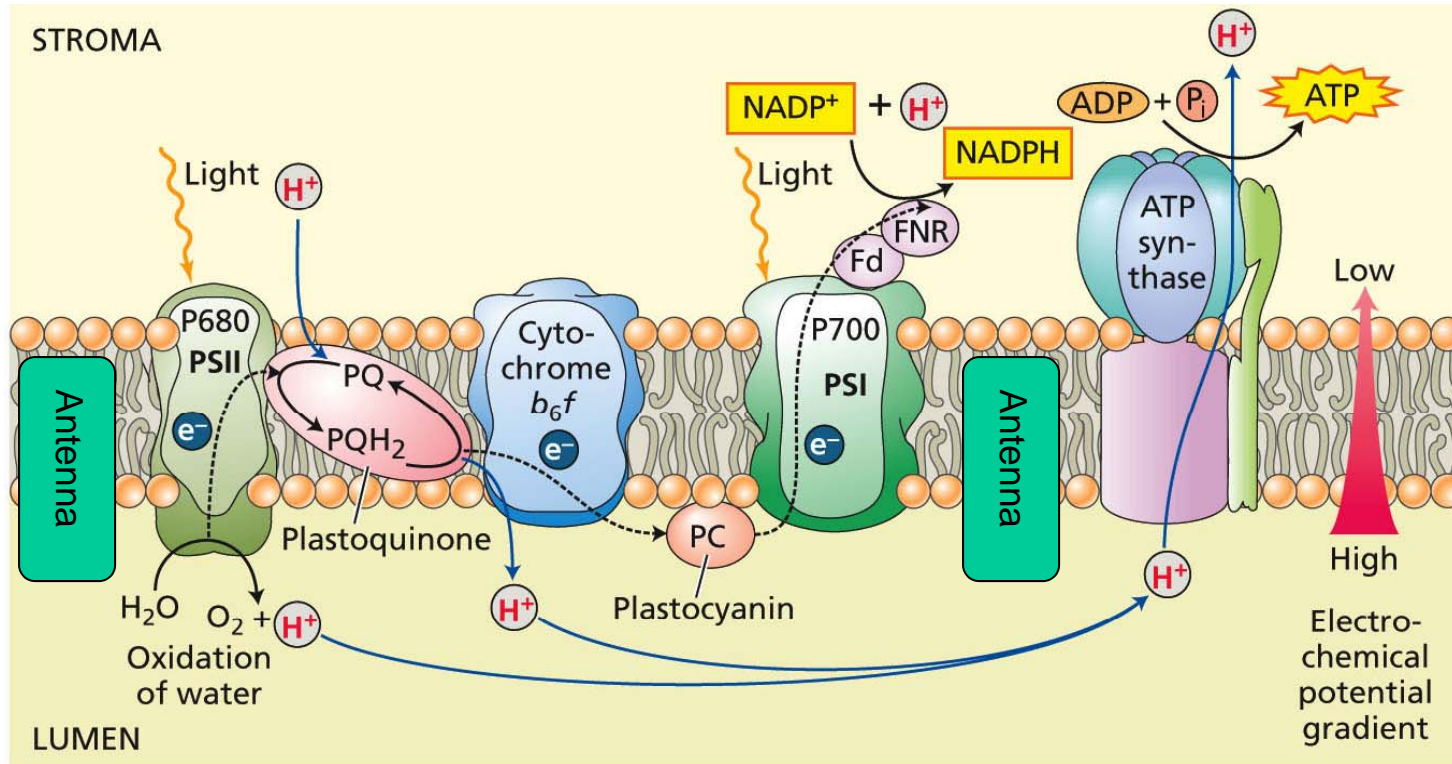
(10 times human needs)



Artificial photosynthesis

- Exploiting the physics and chemistry underlying photosynthesis for technological purposes

How does photosynthesis work?



© Sinauer Associates, Inc.

Antennas

Light capture, excitation energy migration, regulation and photoprotection

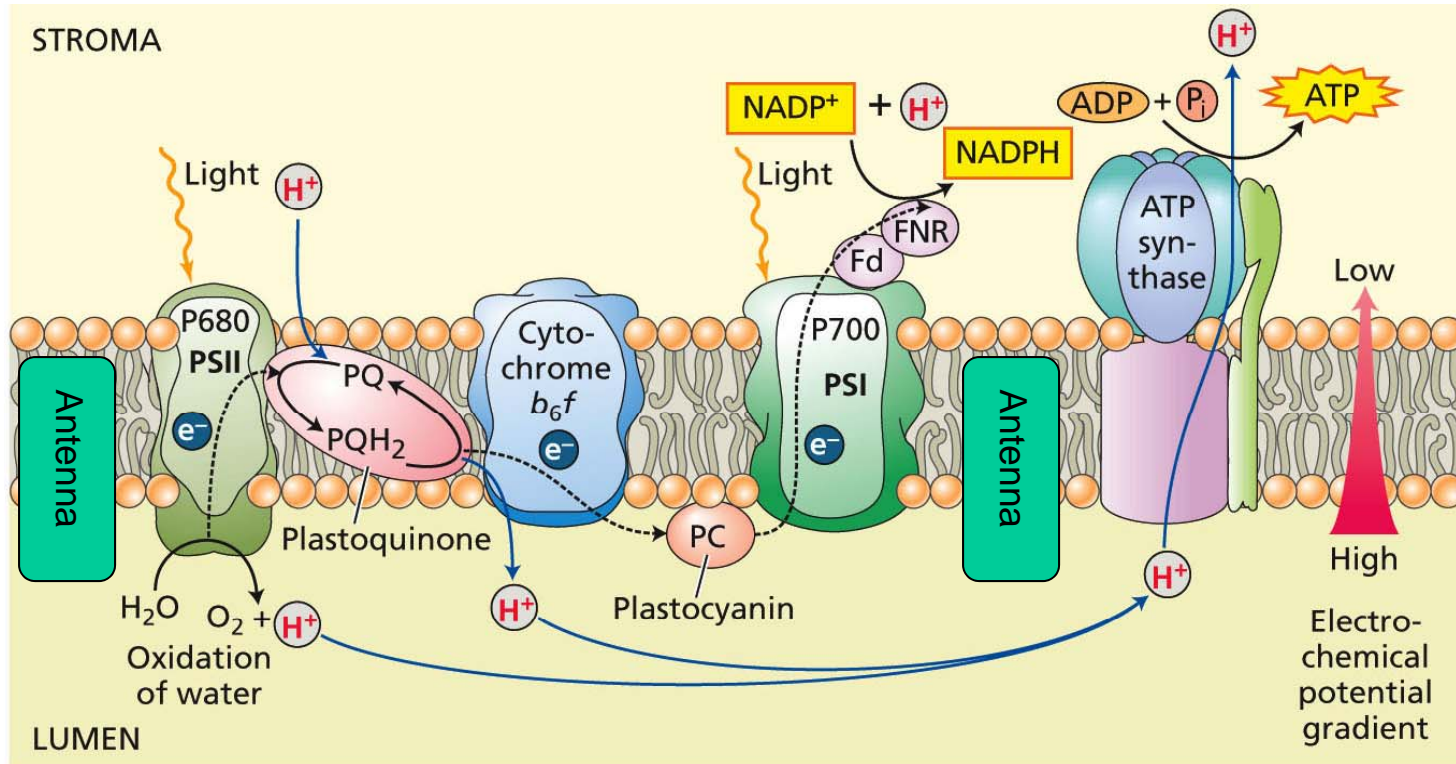
Reaction centers

Charge separation, electrochemical energy

Dark chemical reactions

Catalysts for H_2O oxidation, CO_2 reduction and biofuel production, (carbohydrate, biodiesel, hydrogen)

How does photosynthesis work?



© Sinauer Associates, Inc.

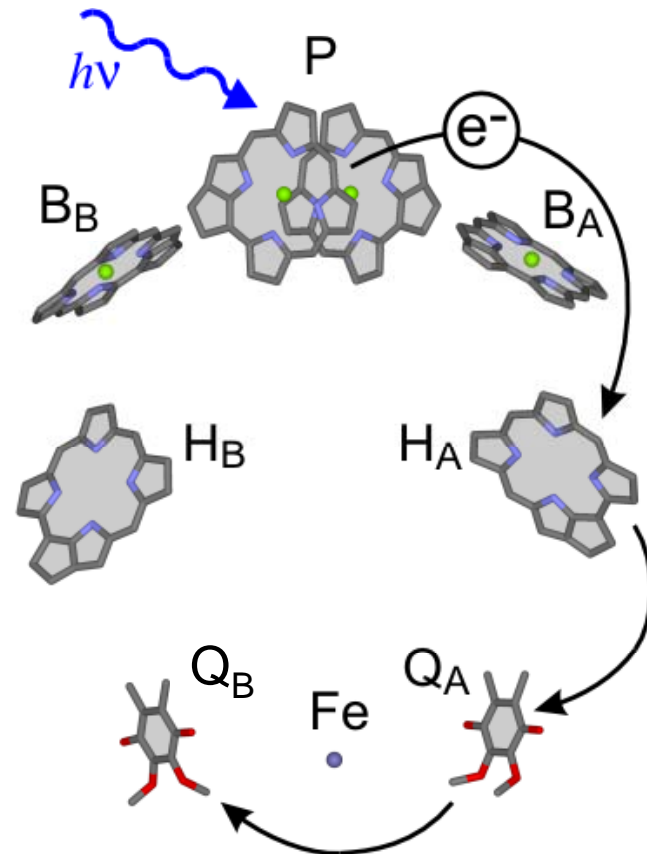
Natural Nanotechnology

Reaction centers



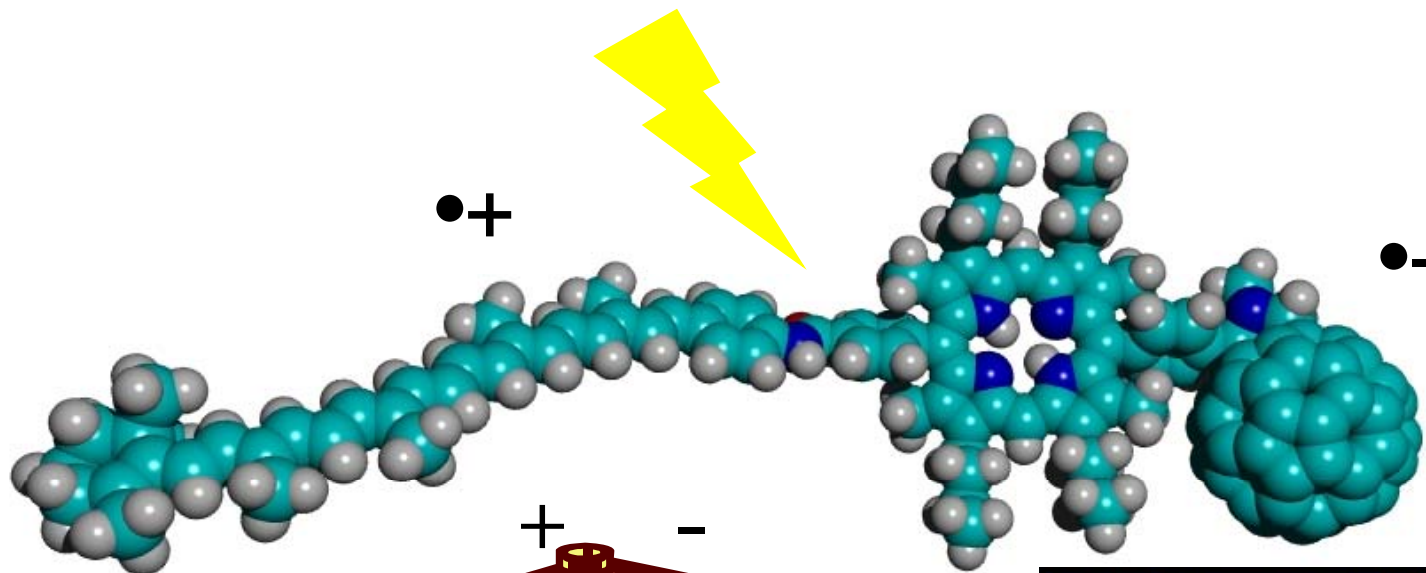
Bacterial reaction center

N. Woodbury, Department of Chemistry and Biochemistry,
Arizona State University



<http://www.fuchs-research.net>

Artificial reaction centers

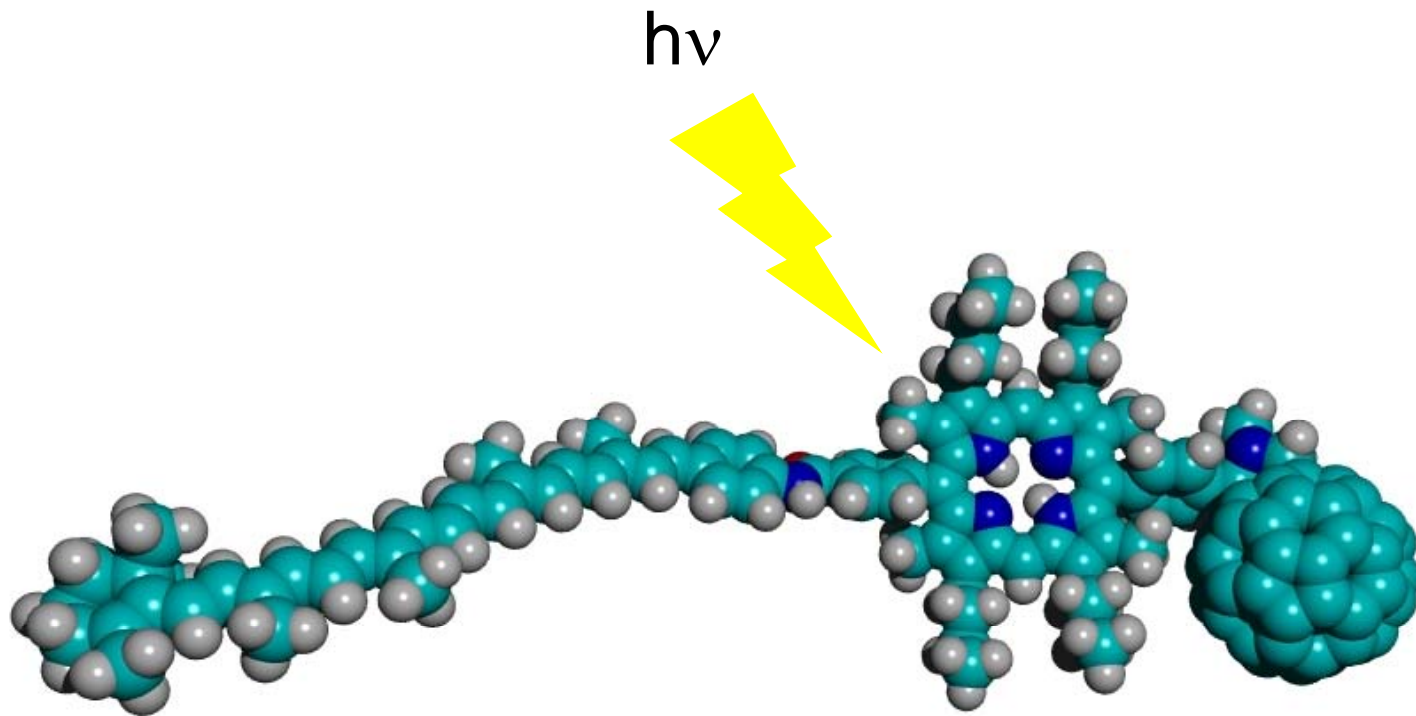


Molecular photovoltaics



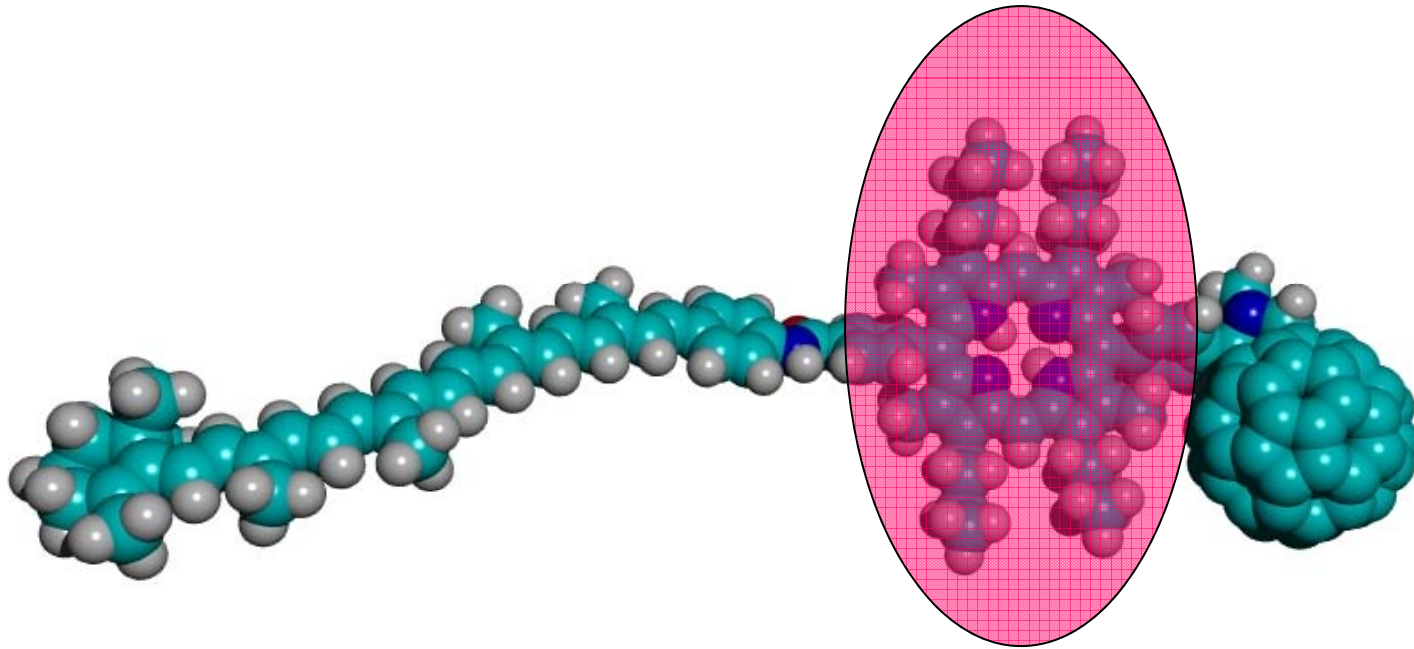
Kuciauskas, D.; Liddell, P. A.; Lin, S.;
Stone, S. G.; Moore, A. L.; Moore, T. A.;
Gust, D. *J. Phys. Chem. B* **2000**, *104*,
4307

Light absorption



C-P-C₆₀

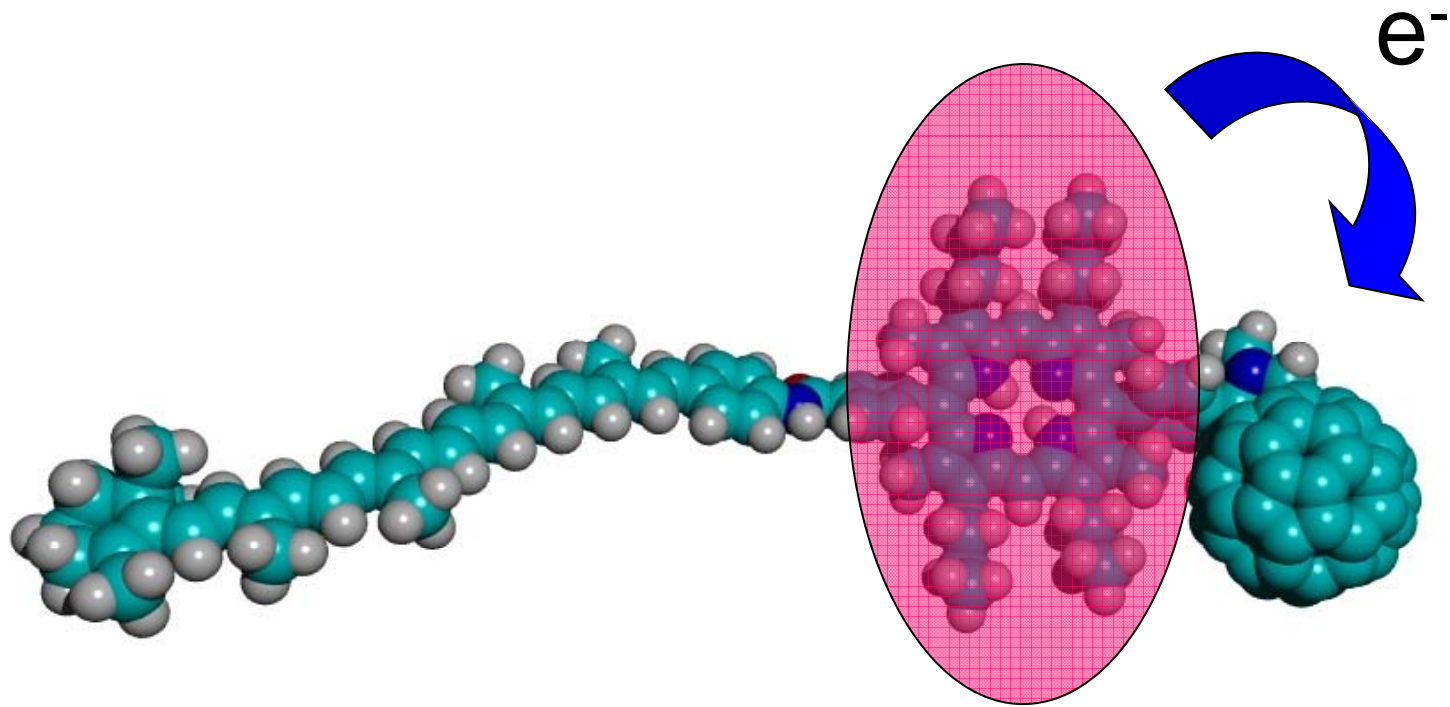
Light absorption



C-¹P-C₆₀

Porphyrin first excited singlet state

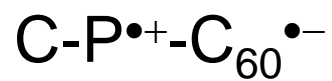
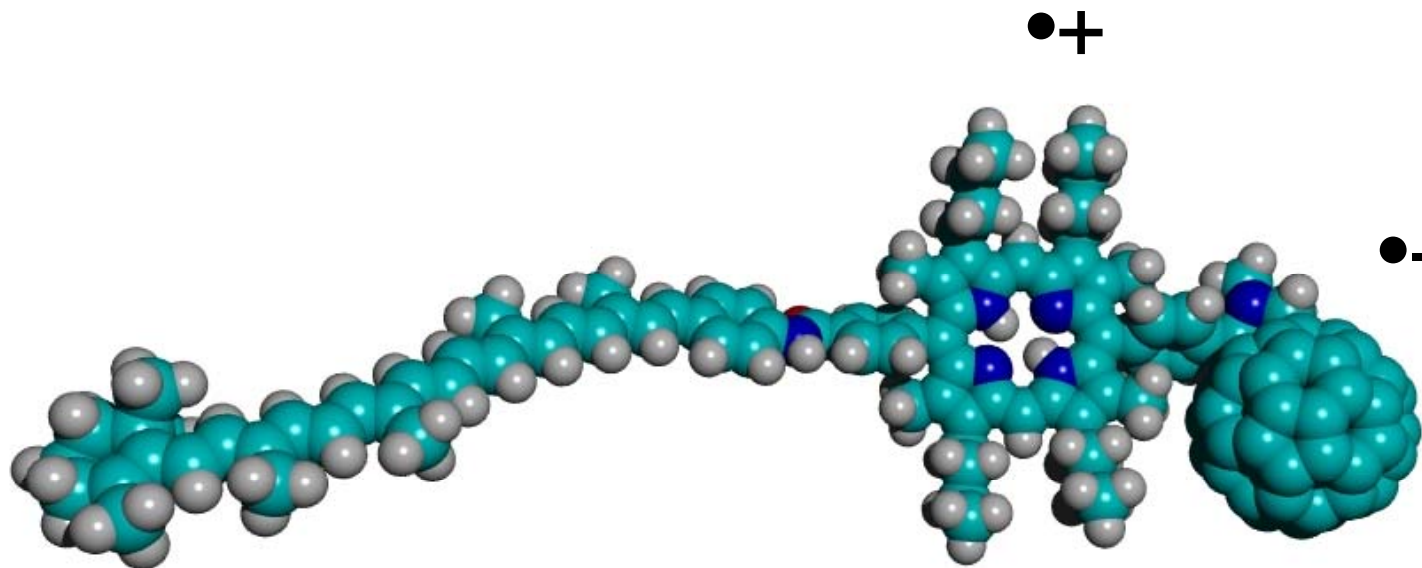
Photoinduced electron transfer



3 picoseconds

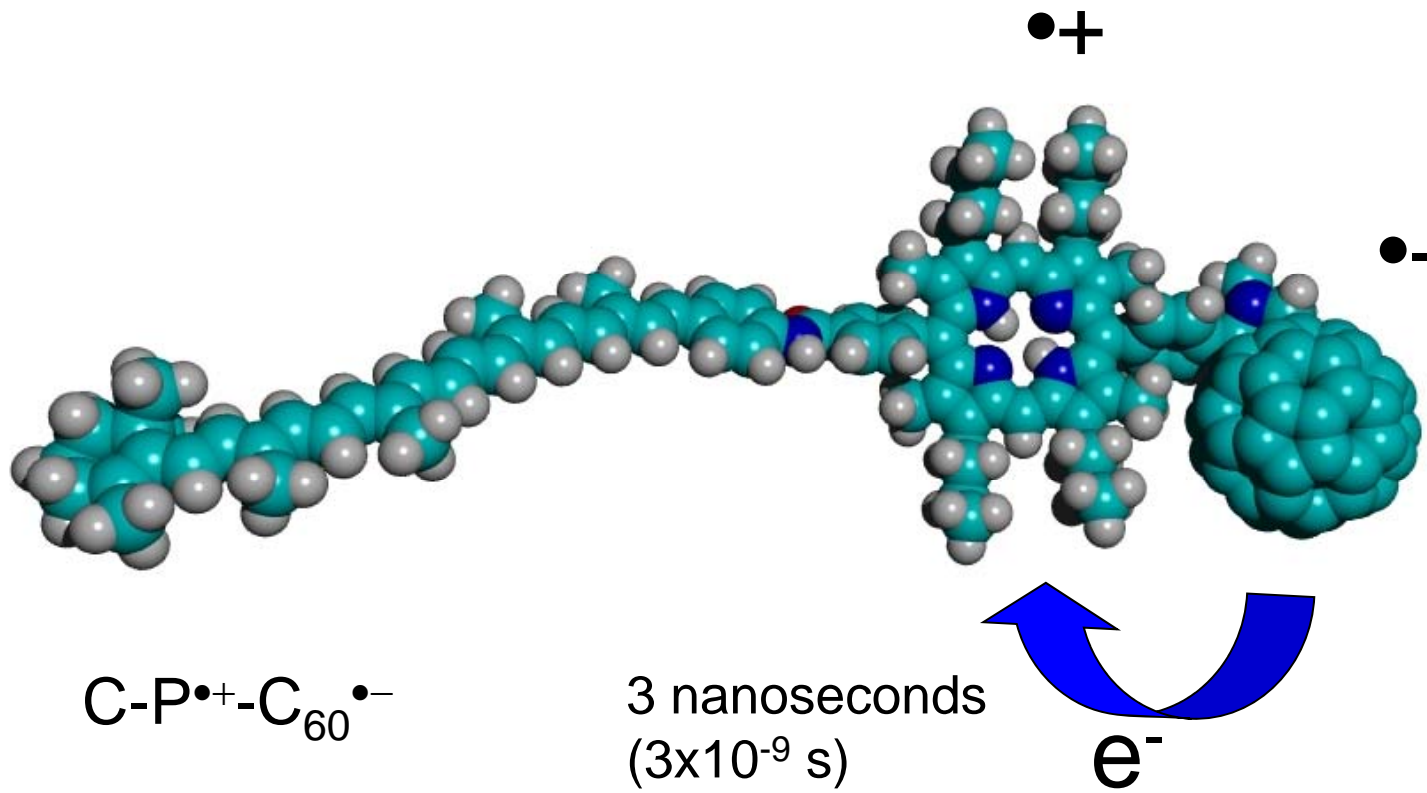
$(3 \times 10^{-12} \text{ s})$

Photoinduced electron transfer

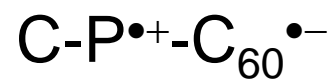
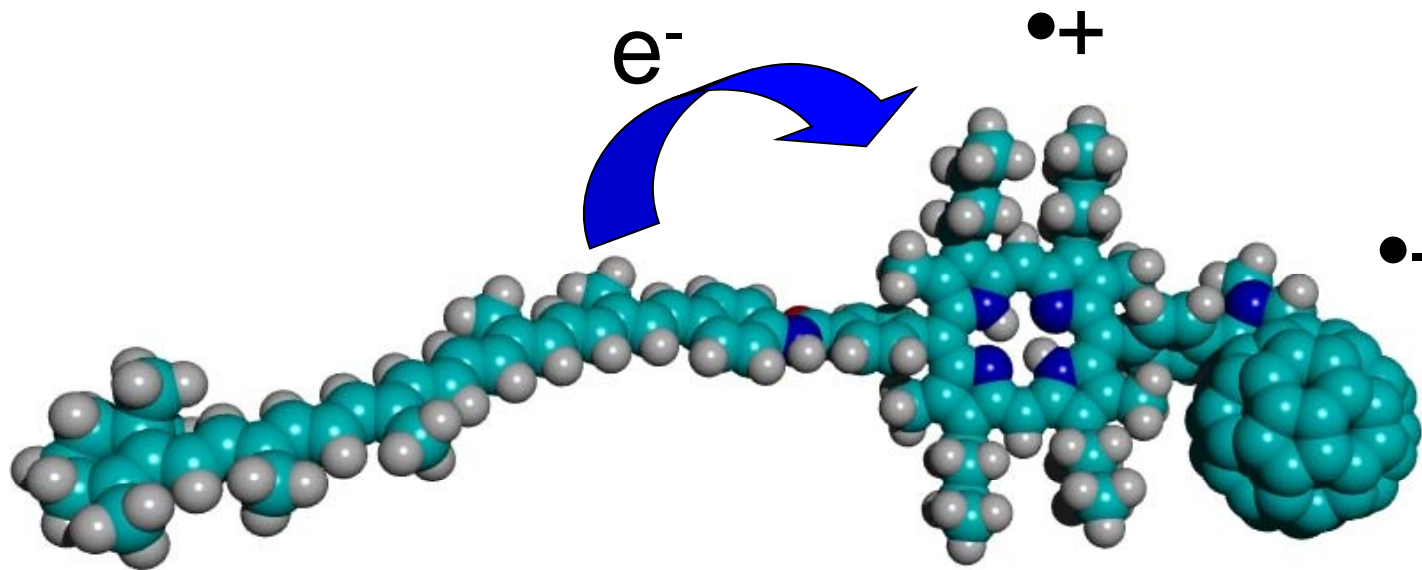


Charge-separated state

Charge recombination

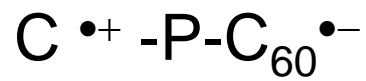
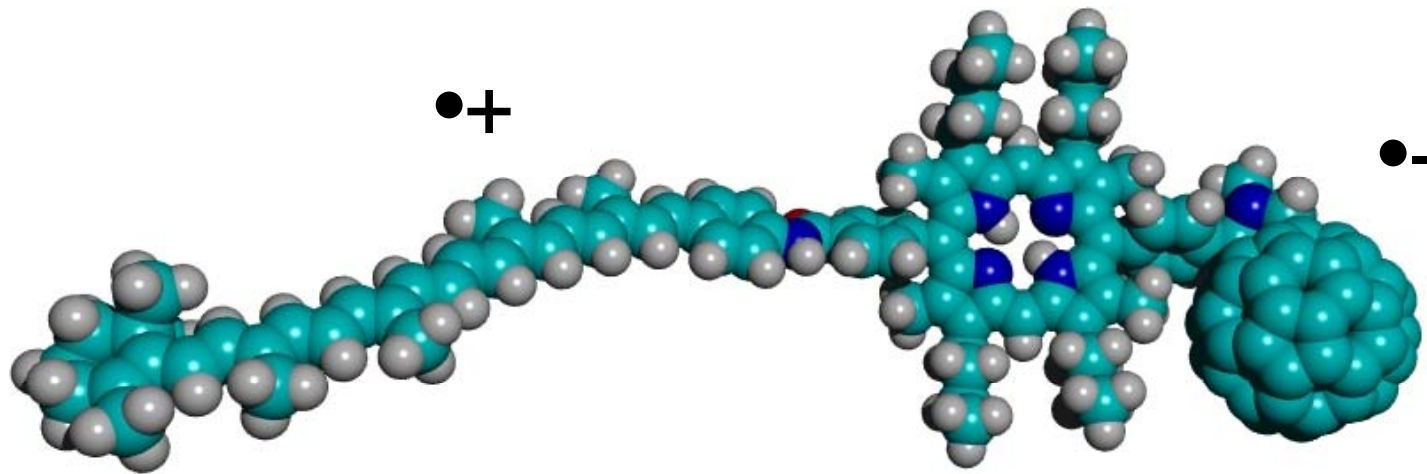


Charge shift reaction



67 picoseconds

Light energy stored as electrochemical energy



Final charge-separated state

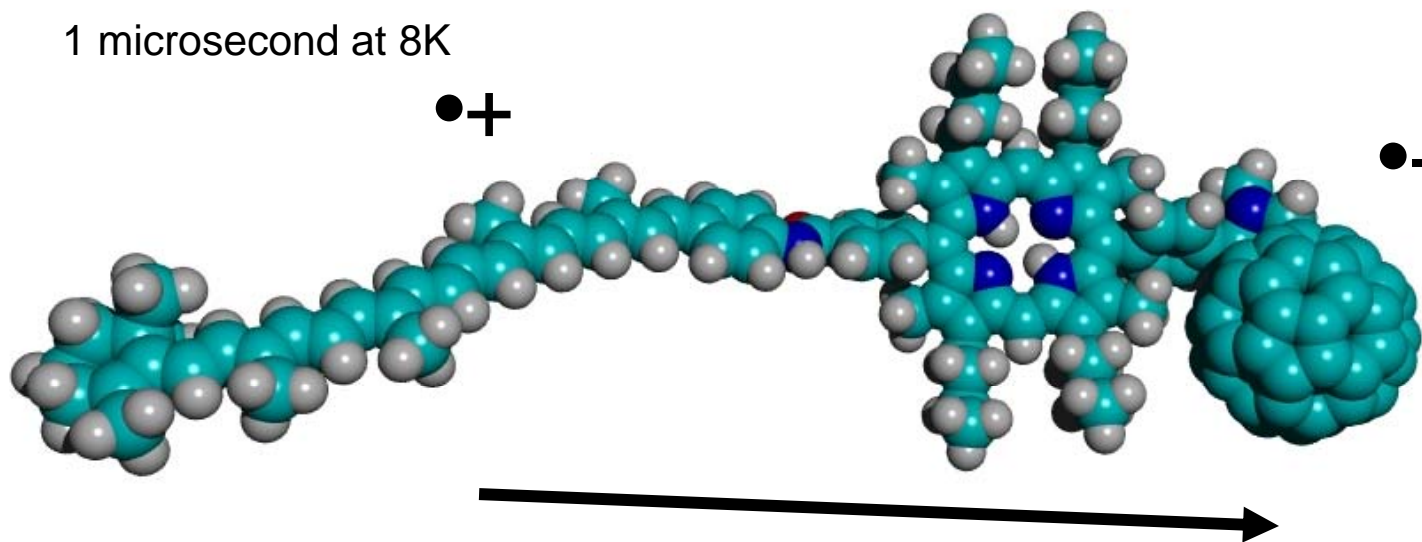
Light energy stored as electrochemical energy

Yield of charge separated state ~ 100%

Stored energy ~1.0 electron volt

Lifetime = hundreds of ns at room temp.

1 microsecond at 8K



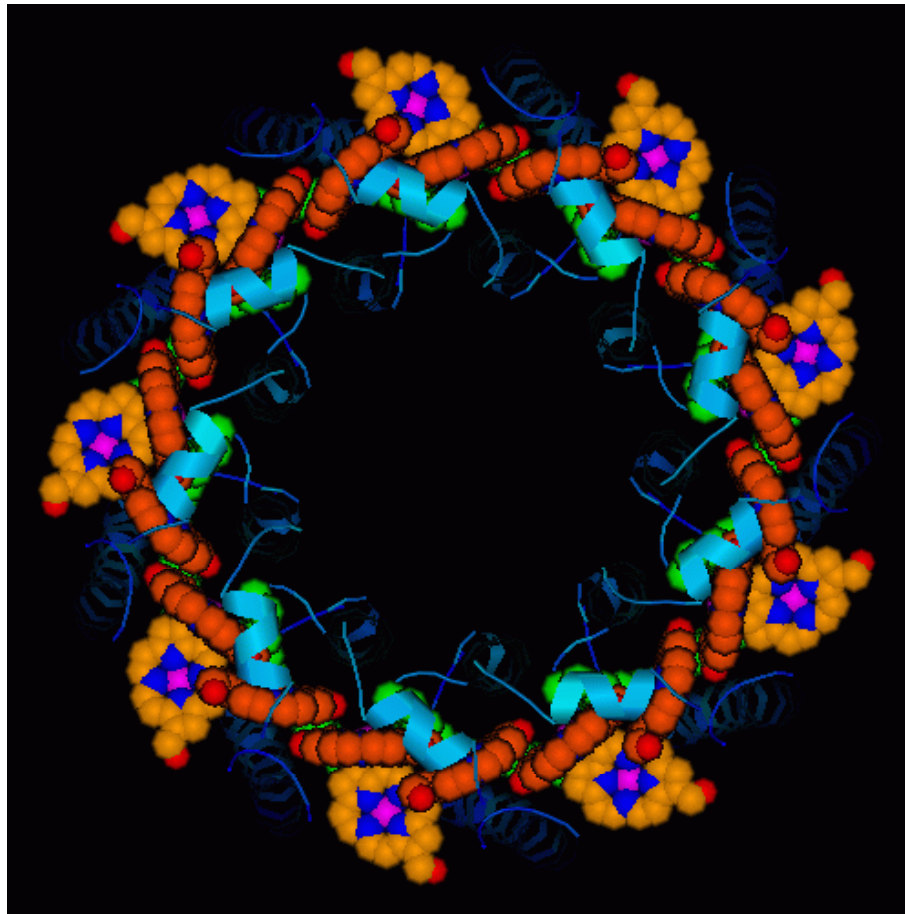
Dipole moment ~160 D

Smirnov, S. N.; Liddell, P. A.; Vlassiouk, I. V.; Teslja, A.; Kuciauskas, D.; Braun, C. L.; Moore, A. L.; Moore, T. A.; Gust, D. *J. Phys. Chem. A*, **2003**, *107*, 7567

Artificial antenna systems

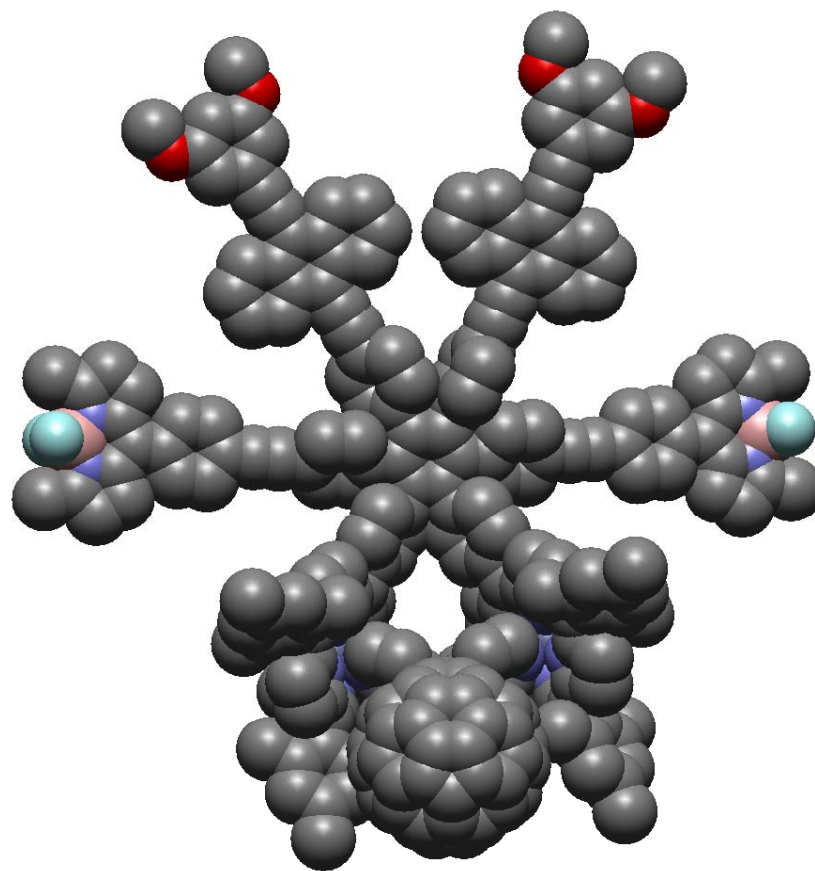
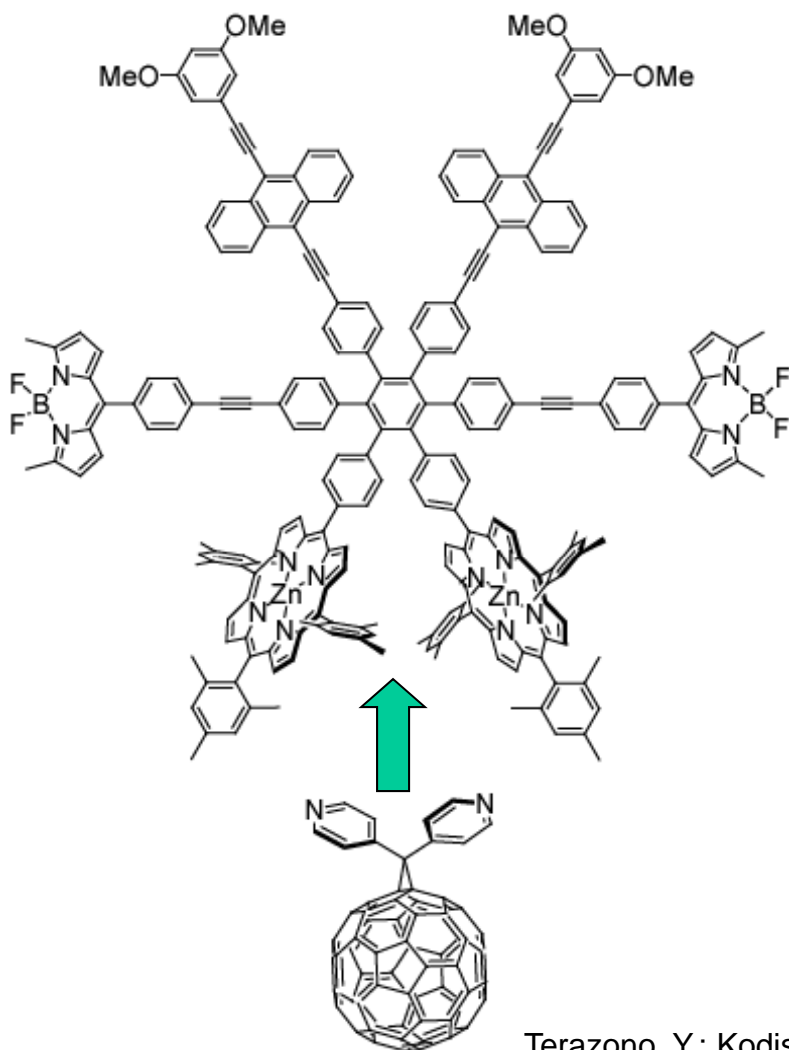
LH2 Photosynthetic antenna

Rhodospseudomonas acidophila



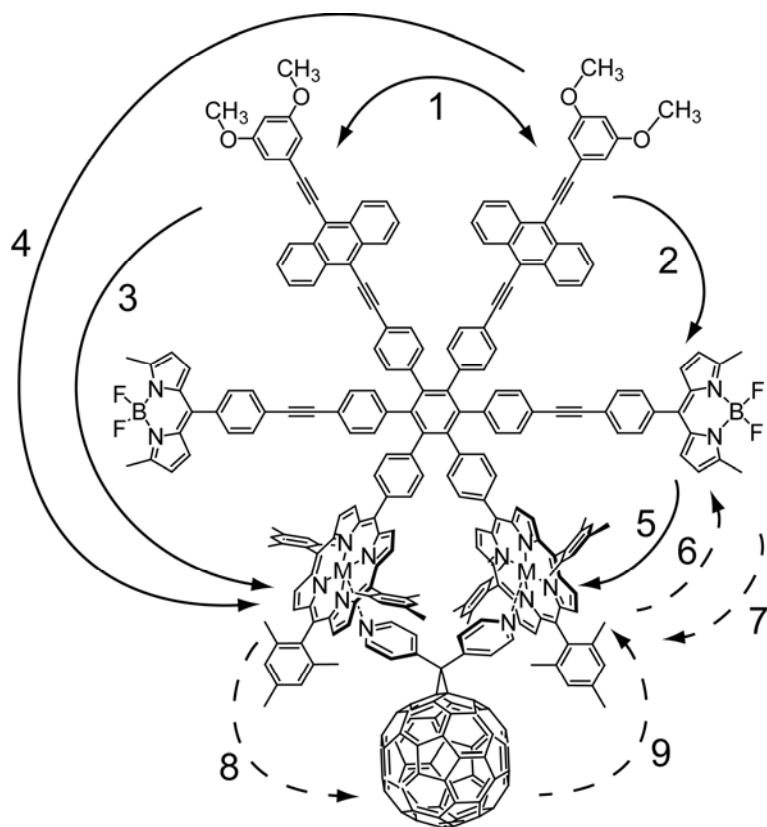
R. Cogdell,
N. Isaacs

A funnel-like antenna – RC complex



Terazono, Y.; Kodis, G.; Liddell, P. A.; Garg, V.; Moore, T. A.; Moore, A. L.; Gust, D. J. *Phys. Chem. B* **2009**, *113*, 7147-7155

Energy and electron transfer



Time constants for photochemical processes (ps)

process	porphyrin hexad		zinc	porphyrin fullerene	
	free base - 2-meTHF	2-meTHF	hexad 1,2-DFB	heptad - 1,2-DFB	
step 1	0.4	0.4	0.4 ^d	0.4	
step 2	5 - 14	5 - 14	5 - 13	5 - 13	
step 3	8	7	7 ^d	7	
step 4	4	6	6	6	
step 5	5 - 26	4 - 21	2 - 15	2 - 15	
step 6		570	230	230	
step 7		4800	1500	1500	
step 8				3	
step 9				230	

Both energy transfer to the porphyrins and photoinduced electron transfer are highly efficient.

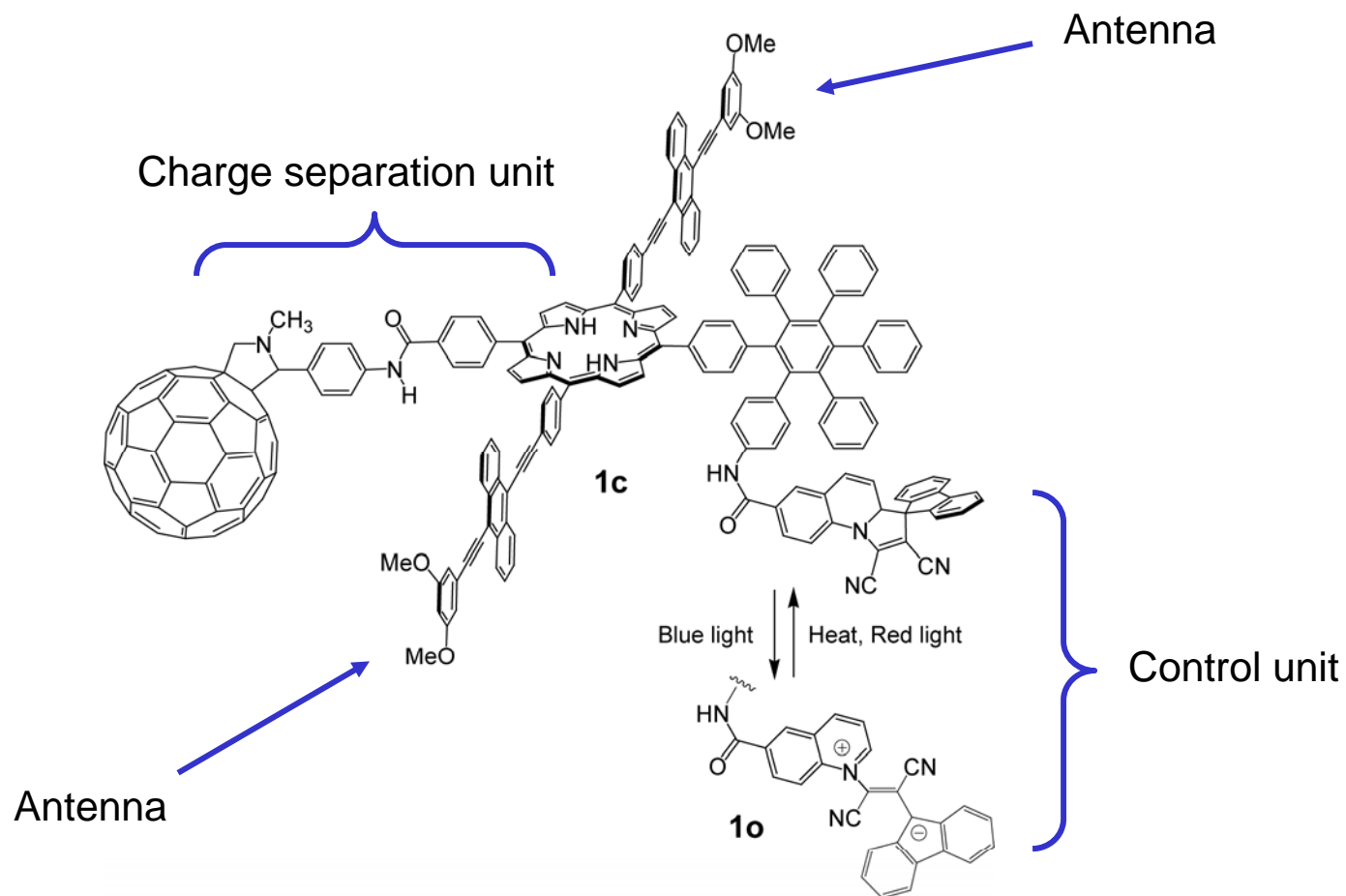
Regulation and photoprotection

- Sunlight is diffuse, so photosynthesis uses antennas to maximize delivery of excitation energy to reaction centers.
- If the sunlight intensity suddenly increases, photosynthesis is overdriven. Singlet oxygen, reactive radicals, redox intermediates, and other destructive species are produced.
- Photosynthetic organisms have evolved regulatory mechanisms to down-regulate delivery of excitation energy to reaction centers under such conditions. (e.g., “non-photochemical quenching, NPQ”)

Regulation in nanotechnology

- Functional nanoscale systems, including artificial photosynthesis, will need self-regulation and control circuits.
- How can we design photocatalysts or other molecular components that are not only functional, but also adaptive, responding to changes in their environment?

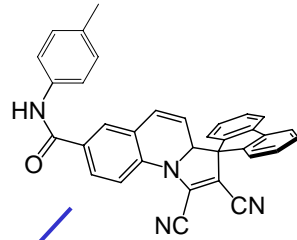
Self-regulation of photoinduced electron transfer by a molecular nonlinear transducer



Straight, S. D.; Kodis, G.; Terazono, Y.;
Hambourger, M.; Moore, T. A.; Moore, A. L.;
Gust, D. *Nature Nanotechnology* **2008**, 3, 280

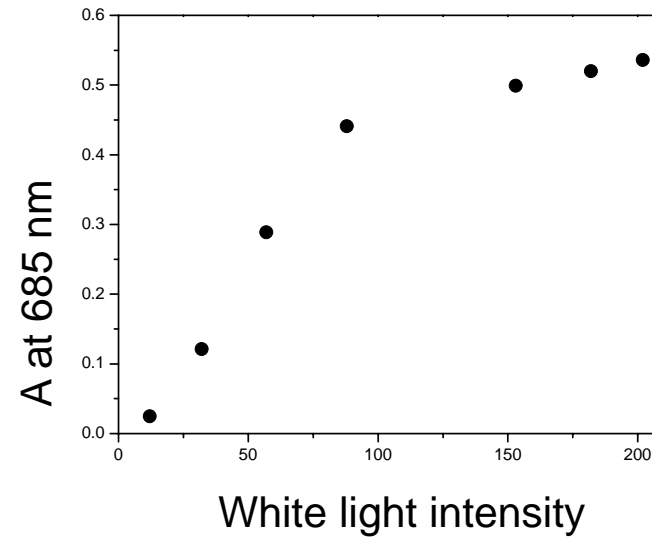
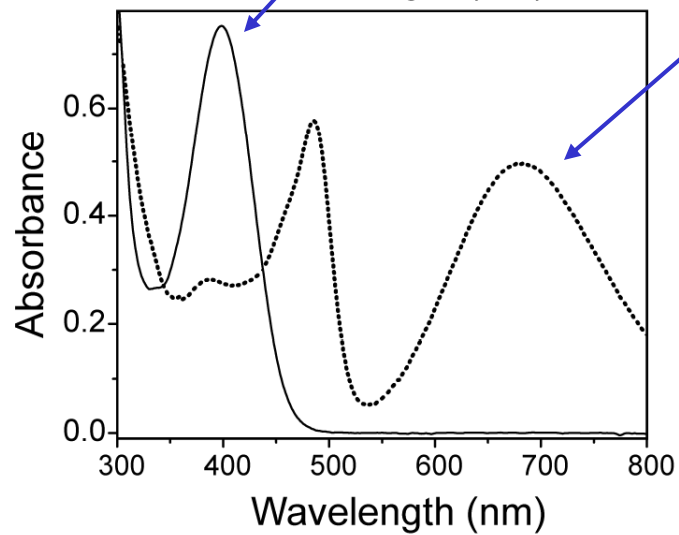
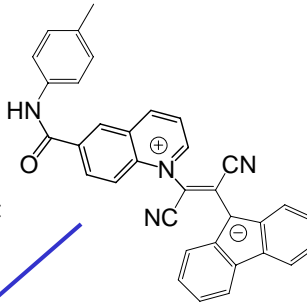
Photochromic control unit

DHI
dihydroindolizine



Blue light
Heat, Red light

BT
betaine



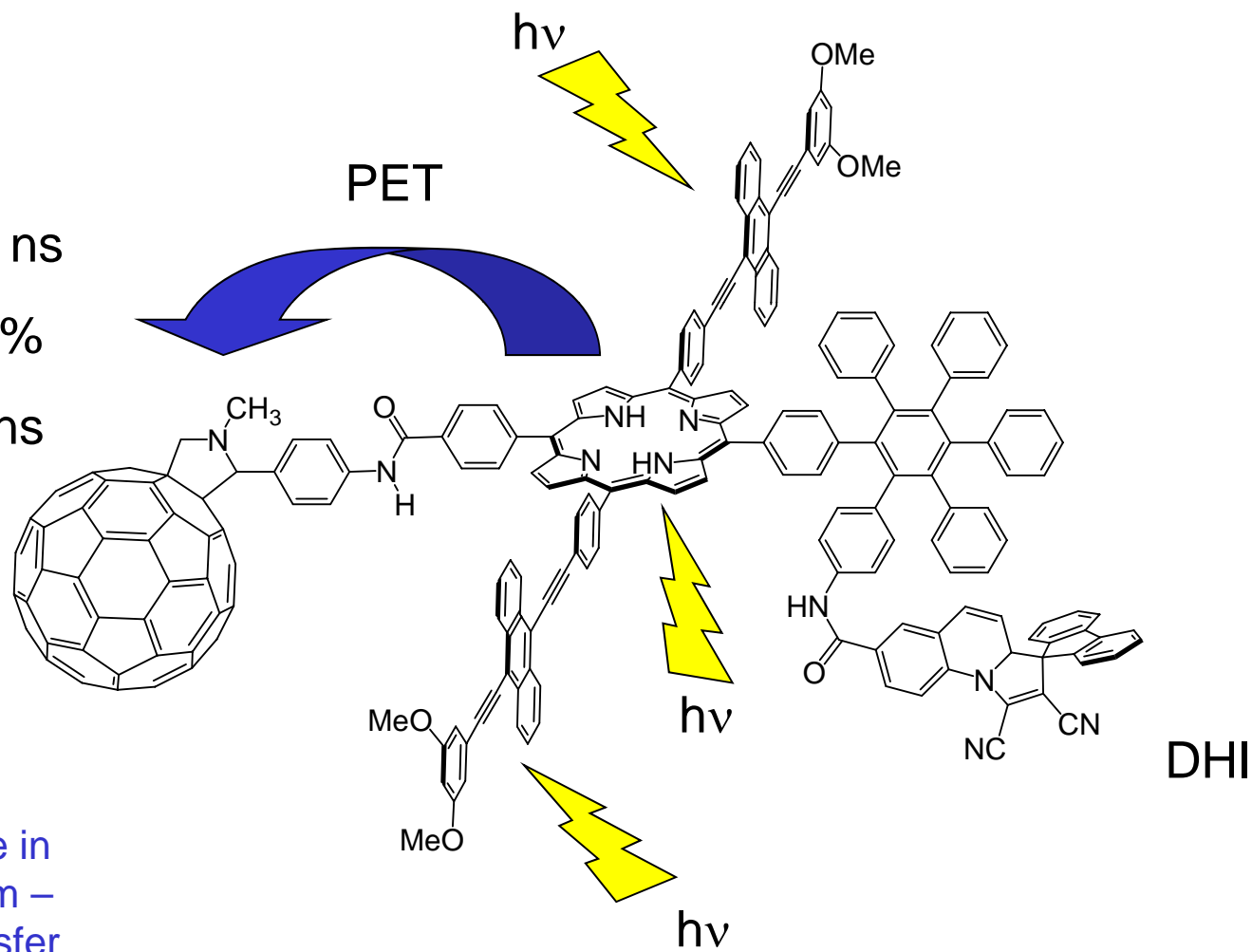
Photostationary
distribution

Principle of operation

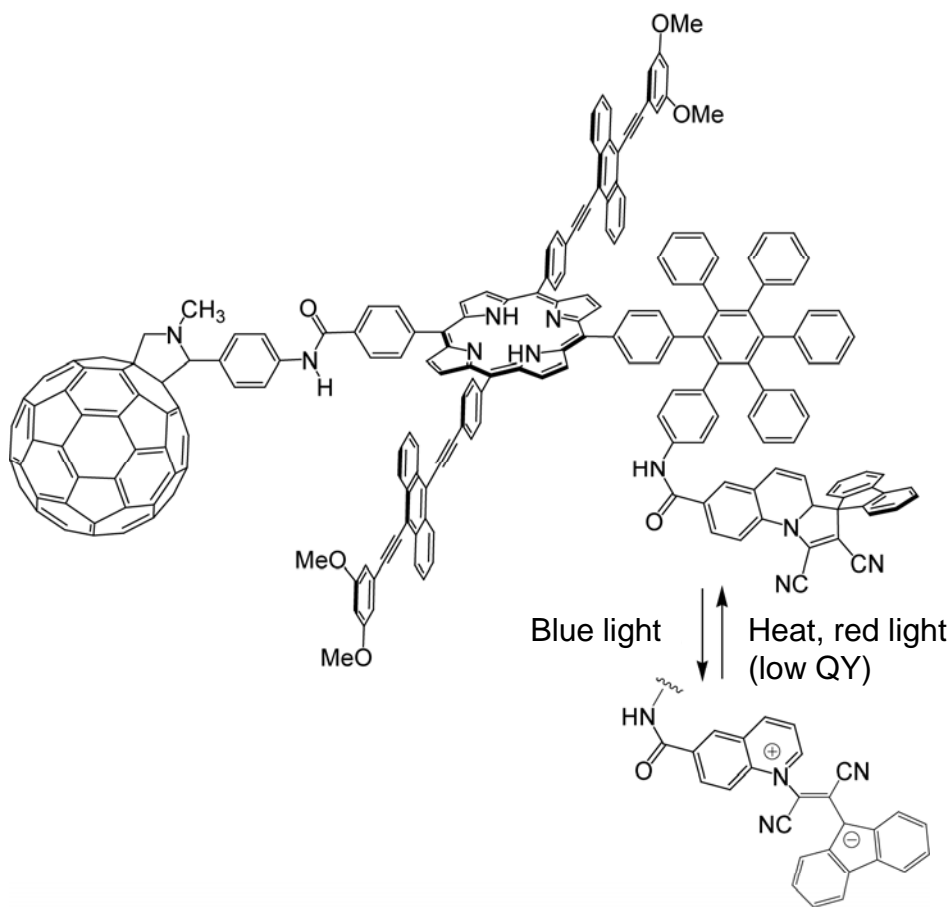
$$\tau_{cs} = 2.2 \text{ ns}$$

$$\Phi_{cs} = 82\%$$

$$\tau_{cr} = 14 \text{ ns}$$

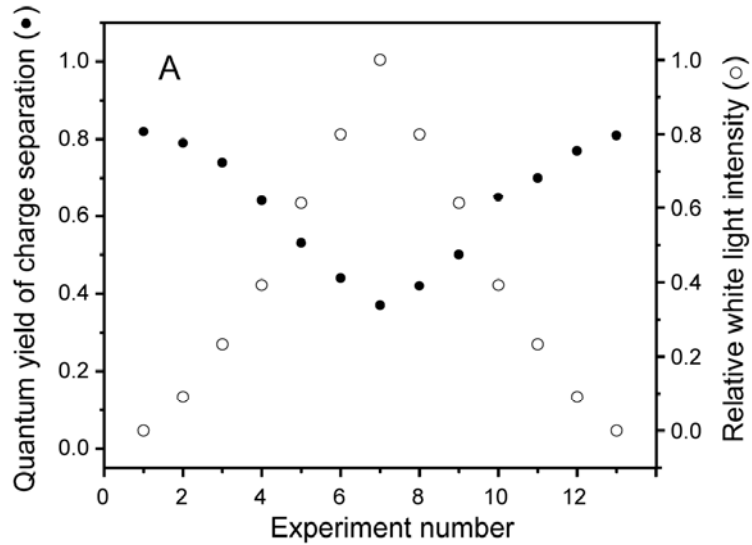


Principle of operation

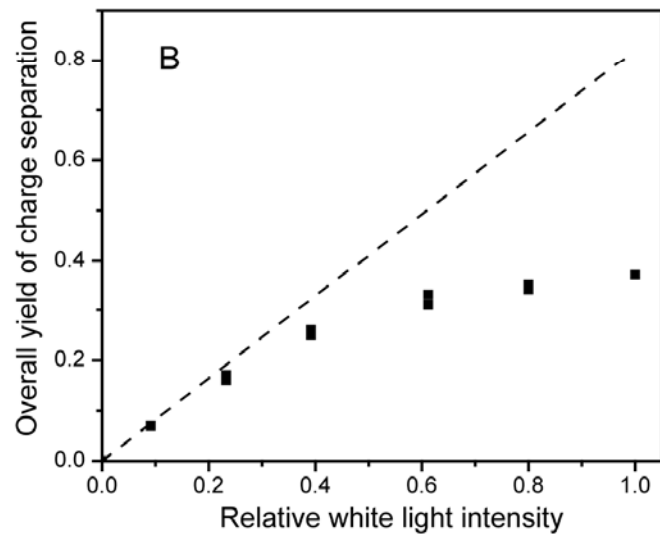


Isomer ratio, and therefore the amount of quenching, is determined by white light intensity and temperature

Operation

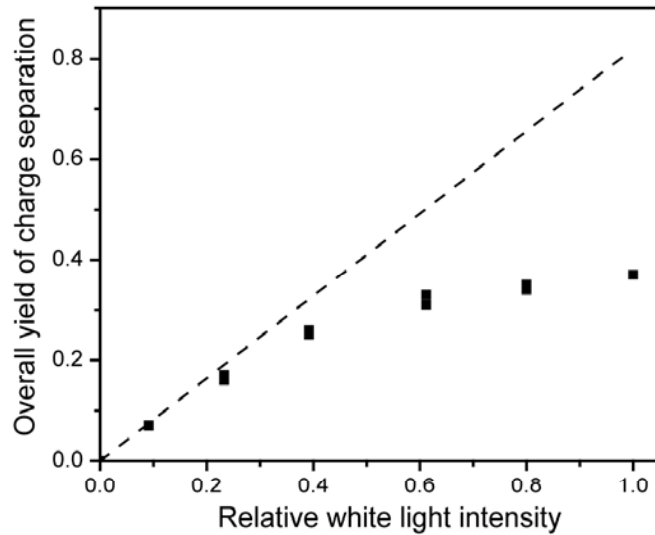


Quantum yield of charge separation is reduced from 82% at low light to as low as 27% under bright white light.

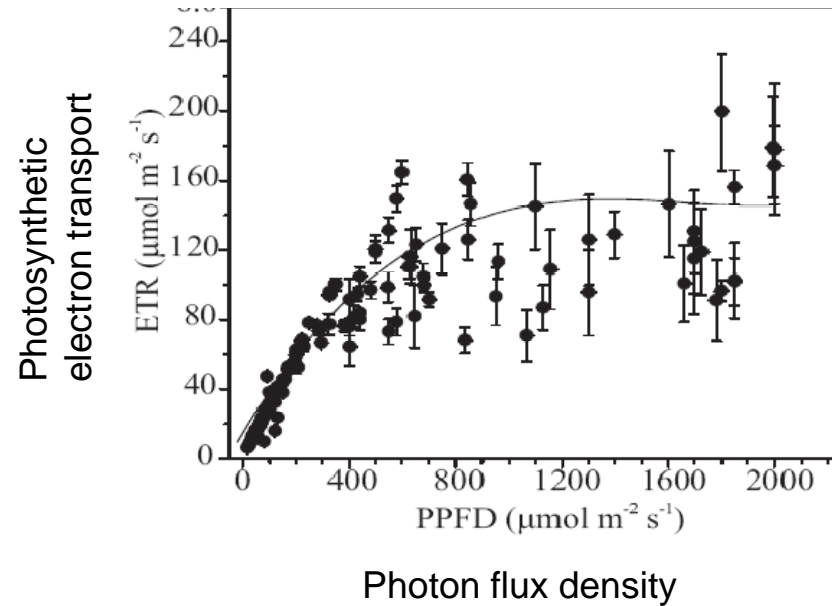


The overall yield of charge separation is a non-linear function of white light intensity

Comparison with natural NPQ

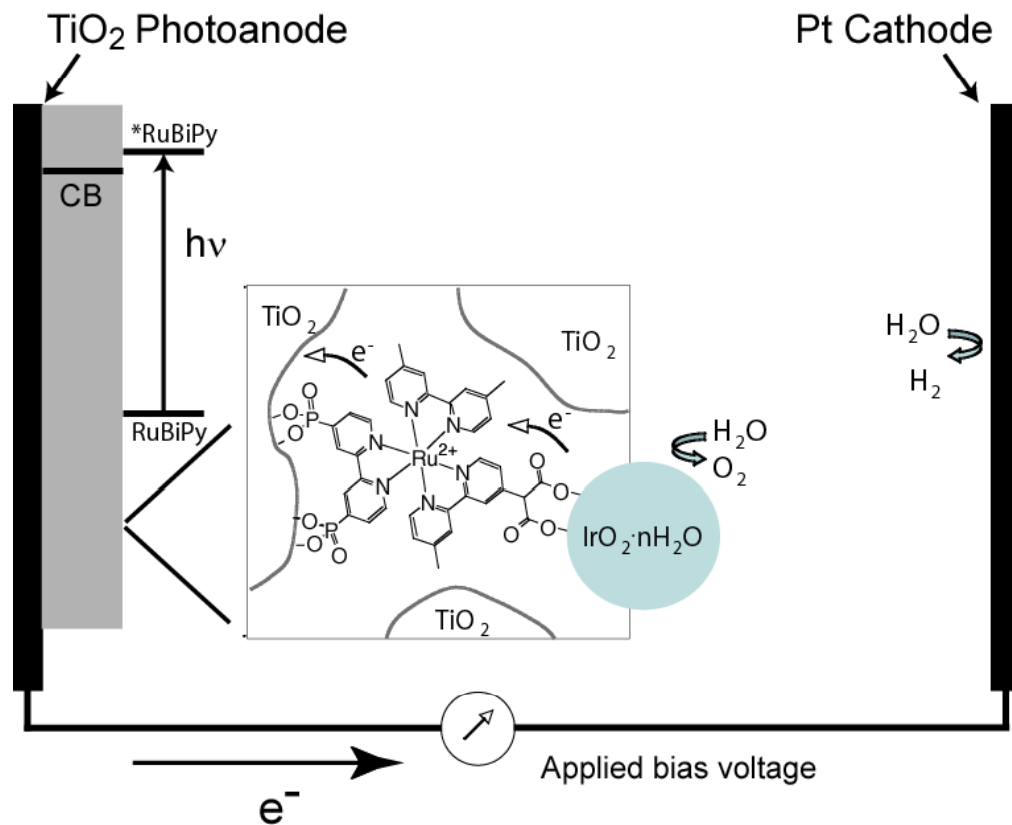
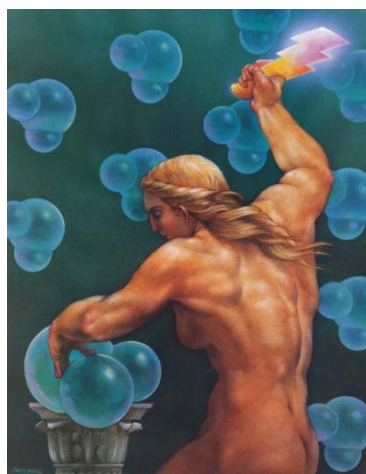


Self-regulation in
molecular system



Natural regulation in
orange trees

Photocatalytic water splitting



W. J. Youngblood, S.-H. A. Lee, Y. Kobayashi, E. A. Hernandez-Pagan, P. G. Hoertz, T. A. Moore, A. L. Moore, D. Gust, T. E. Mallouk, *J. Am. Chem. Soc.*, **2009**, 131, 926-927

Conclusions

- Photosynthetic principles can guide the design of artificial systems that convert sunlight into useful forms of energy
 - Antenna function
 - Photoinduced charge separation
 - Regulation and photoprotection
 - Production of useful fuels or electricity
- Such systems are in general not yet practical for technological applications – more research is needed

Many thanks to

- Tom Moore, Ana Moore, Tom Mallouk, Sergei Smirnov, Chuck Braun
- Many students and postdocs – listed on the slides
- The Department of Energy and National Science Foundation for funding