# Advanced Nano-Composite Lithium-Metal-Oxide Electrodes for

## High Energy Lithium-Ion Batteries\*

### Sun-Ho Kang and Michael Thackeray

Argonne National Laboratory, 9700 S. Cass Ave., Argonne, IL 60439 sunho.kang@anl.gov

#### ABSTRACT

Layered lithium cobalt oxide (LiCoO<sub>2</sub>) and nickel-substituted derivatives are the most widely used positive electrode materials in lithium-ion batteries. However, their limited capacity (150-160 mAh/g), thermal instability in the charged state and the recent global interest in lithium-ion batteries for electrically-powered vehicles have emphasized the need to develop advanced cathode materials that will offer higher energy and improved safety. In efforts to develop advanced cathode materials for high energy cells, we have adopted a unique approach of integrating lithium metal oxides in nano-composite structures, examples of which are: (1) 'layered-layered' electrodes with layered  $Li_2MnO_3$ (Li<sub>1.33</sub>Mn<sub>0.67</sub>O<sub>2</sub>) and LiMO<sub>2</sub> (M=Ni,Co,Mn) components, represented generically as xLi<sub>2</sub>MnO<sub>3</sub>•(1-x)LiMO<sub>2</sub> and (2) 'layered-layered-spinel' electrodes comprised of layered Li<sub>2</sub>MnO<sub>3</sub>, layered LiMO<sub>2</sub> and spinel LiM'<sub>2</sub>O<sub>4</sub> (M'=Ni, Mn) components. These cathode materials deliver a capacity in excess of 200 mAh/g and they provide a high-voltage stability (>4.4 V vs.  $Li^+/Li$ ); they also possess superior thermal stability over LiCoO<sub>2</sub> and LiCo<sub>1-x</sub>Ni<sub>x</sub>O<sub>2</sub> electrodes due to their high Mn content. The structural compatibility of Li<sub>2</sub>MnO<sub>3</sub> with layered LiMO<sub>2</sub> and spinel LiM'<sub>2</sub>O<sub>4</sub> materials, which is a consequence of their common cubic-close-packed oxygen arrays, allows the integration of the Li<sub>2</sub>MnO<sub>3</sub>-LiMO<sub>2</sub>- and spinel LiM'<sub>2</sub>O<sub>4</sub> components at the atomic level. A detailed account of the electrochemical and structural properties of these highly complex nano-composite materials, as determined by various experimental techniques, such as NMR, TEM, X-ray diffraction and absorption, will be presented.

### Acknowledgments

This work was supported by the Office of Vehicle Technologies of the U.S. Department of Energy.

<sup>\*</sup> The submitted manuscript has been created by UChicago Argonne, LLC, Operator of Argonne National Laboratory ("Argonne"). Argonne, a U.S. Department of Energy Office of Science laboratory, is operated under Contract No. DE-AC02-06CH11357. The U.S. Government retains for itself, and others acting on its behalf, a paid-up, nonexclusive, irrevocable worldwide license in said article to reproduce, prepare derivative works, distribute copies to the public, and perform publicly and display publicly, by or on behalf of the Government.