



Nanostructures for Solar Cells

Donghwan Kim, Hyunju Lee, and *Jeunghee Park

Department of Materials Science and Engineering

***Department of Chemistry**

Korea University

donghwan@korea.ac.kr



Korea University



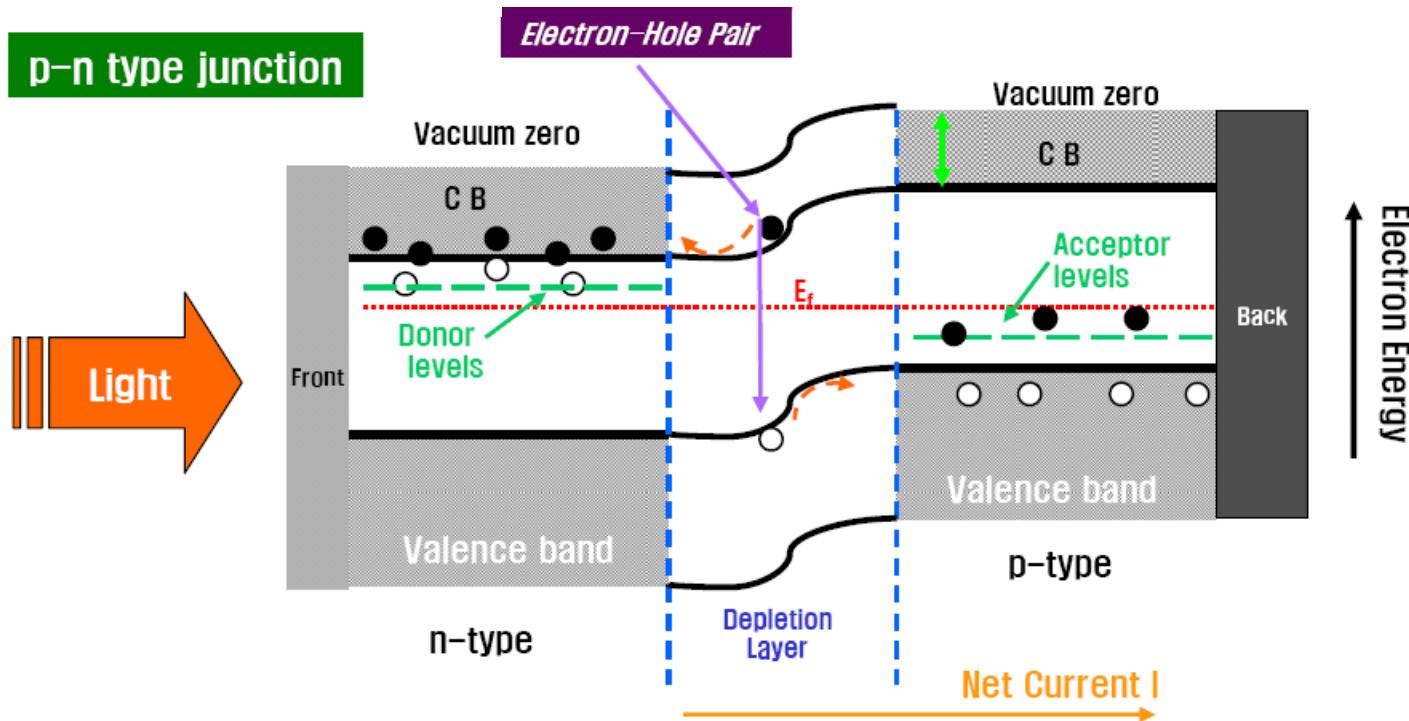
Solar Cells Lab.

- Principle of Operation
- Motivation
- Nanomaterials
- Previous research
- Hybrid nanostructures
- Material Synthesis
- Solar Cell fabrication
- Conclusions
- References



Principle of Operation

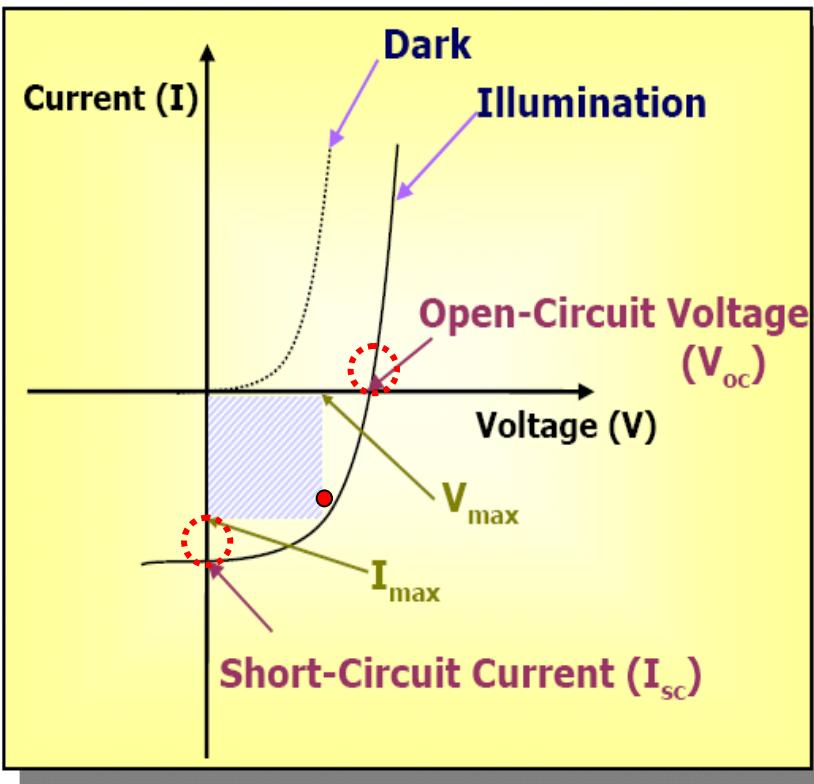
Semiconductor p-n junction



- Band gap enables photogeneration of carriers (ideal: 1.34 eV)
- n and p type doping \Rightarrow asymmetric electrodes (exciton separation)



Solar cell characteristics



- Short-Circuit Current (I_{sc}) For $V = 0$, $I = I_{sc}$
- Open-Circuit Voltage (V_{oc}) For $I = 0$, $V = V_{oc}$
- Fill Factor (FF) $= (I_{max} \times V_{max}) / (I_{sc} \times V_{oc})$
 $= P_{max} / (I_{sc} \times V_{oc})$
- Power Conversion Efficiency (η)
 $= P_{max} / P_{in}$
 $= (I_{max} \times V_{max}) / P_{in}$
 $= FF \times \{ (I_{sc} \times V_{oc}) / P_{in} \}$

$$* P_{in} = 100 [\text{mW/cm}^2]$$

$$P_{sun} = \int b(E)E dE$$



Motivation

Silicon Solar Cell

- Power conversion efficiency up to 24% in lab.¹
- High energy consumption at fabrication
- Cost-intensive fabrication

Organic Solar Cell

- Low energy consumption at fabrication
- Inexpensive fabrication
- Mechanically flexible on appropriate substrates
- Only 5% achieved in lab.¹

Issues

To harvest more photons from a wide spectral range:

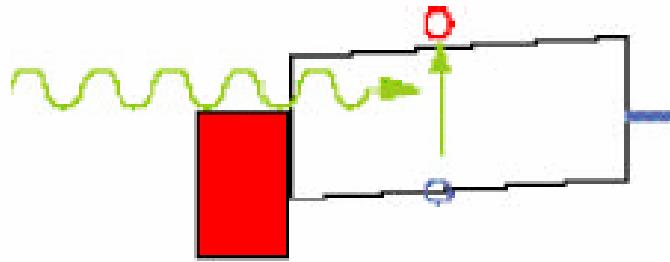
- Develop new absorber system that responds to IR

To improve the efficiency beyond the limit of 5%:

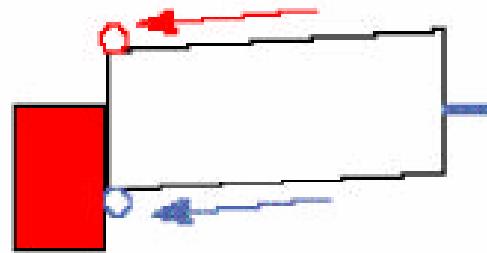
- Fabricate ordered nanostructures to improve photo-induced charge separation and transport



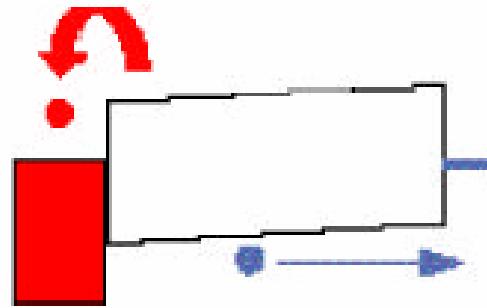
Issues in Organic Photovoltaics



*Photoabsorption
& exciton generation*



*Exciton transport
& separation*



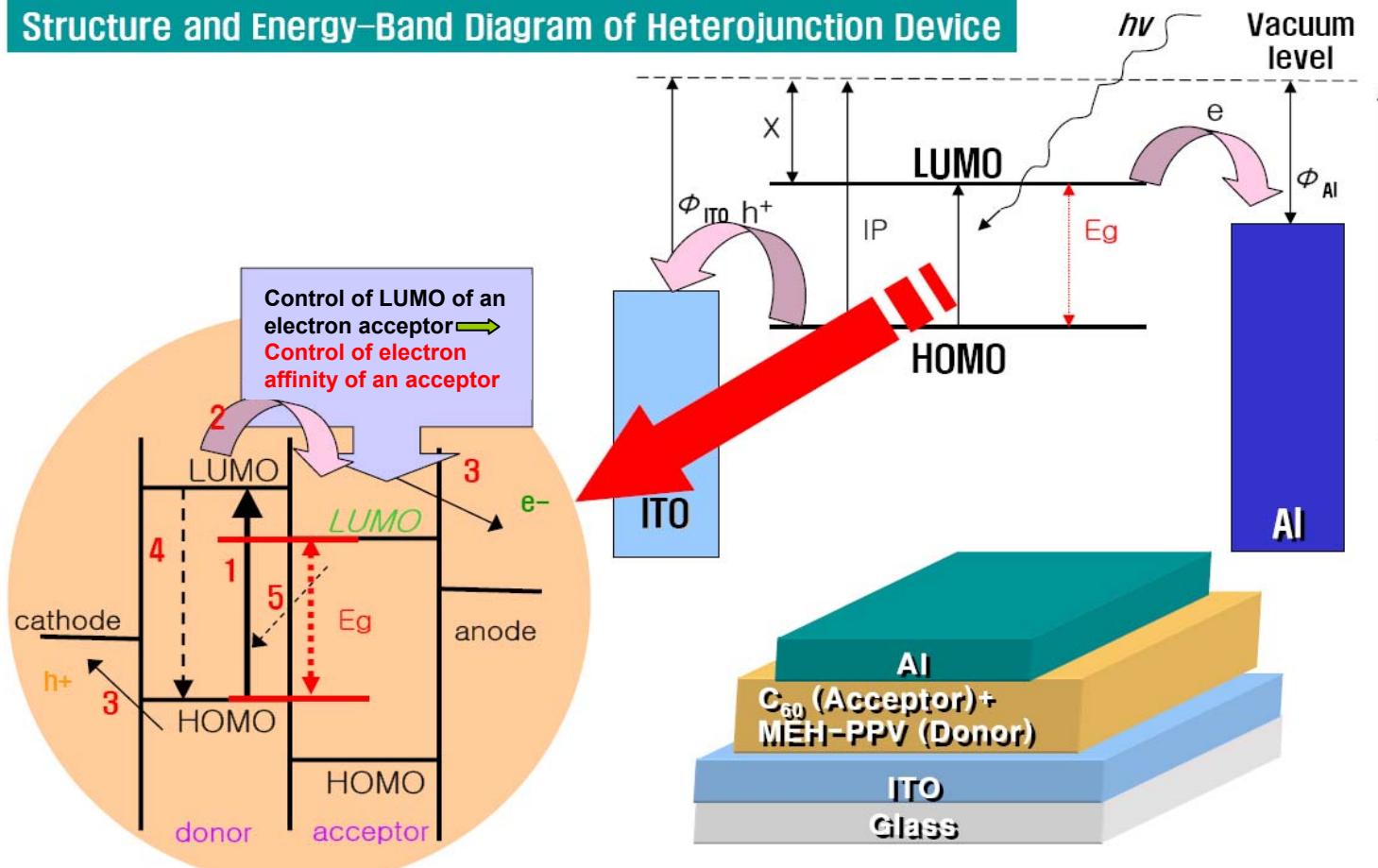
Hole transport

J. Nelson, Materialstoday, May 2002, p20



Organic photovoltaics

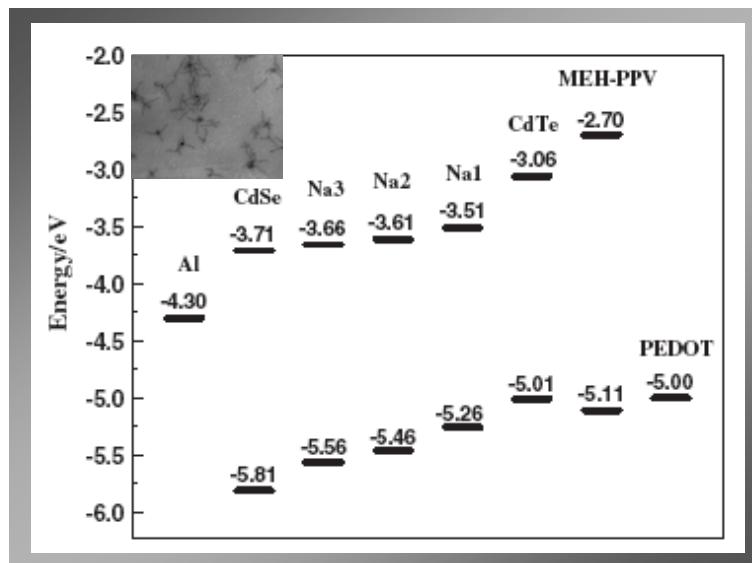
Structure and Energy-Band Diagram of Heterojunction Device



Nanomaterials

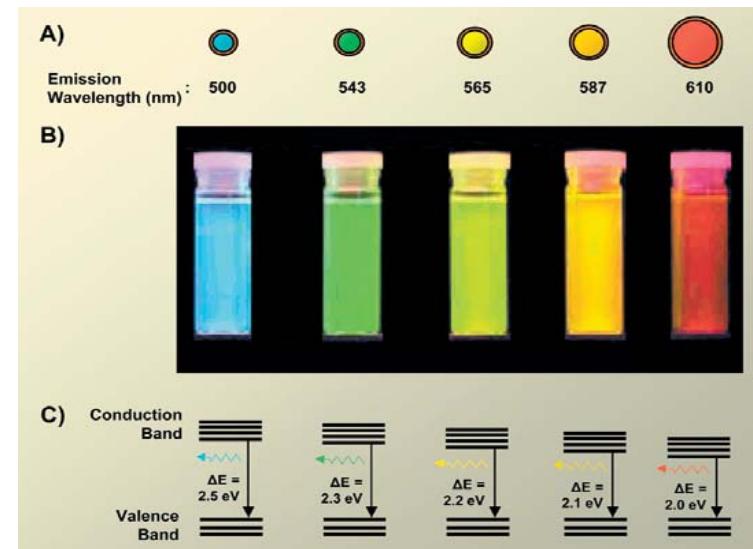
Unique properties of nanomaterials

- High electrical conductivity (10^4 S/cm for metallic (10,10) SWNTs²) and electron affinity (CdS, CdSe, CdTe, InP, GaAs etc.)
- 1-, 2- and 3-D assemblies
- Controlled electronic and surface properties by various components, morphologies, and chemical modifications^{3,4}



Samples	Na1	Na2	Na3
Molar ratio of Se/Te in the mixed precursors (Se + Te)	1/4 (0.25)	1/1 (1)	4/1 (4)
Molar ratio of Se/Te in the alloyed nanocrystals, determined by ICP	0.30	1.13	3.55
Se content in $\text{CdSe}_x\text{Te}_{1-x}$, $x =$	0.23	0.53	0.78

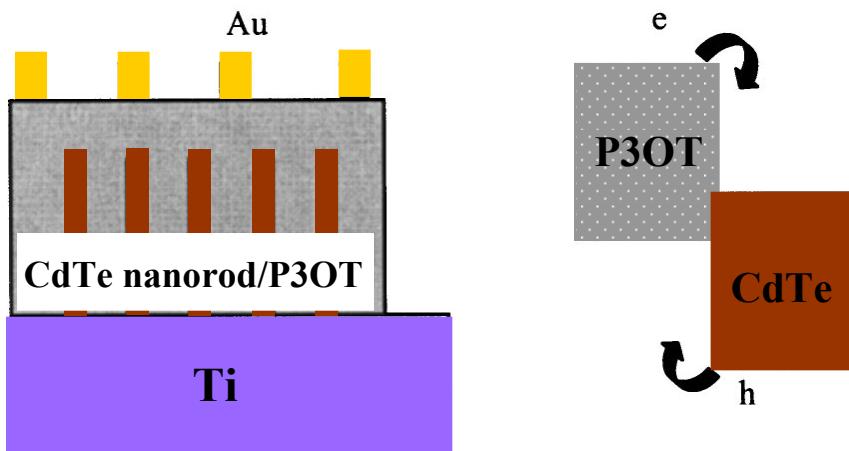
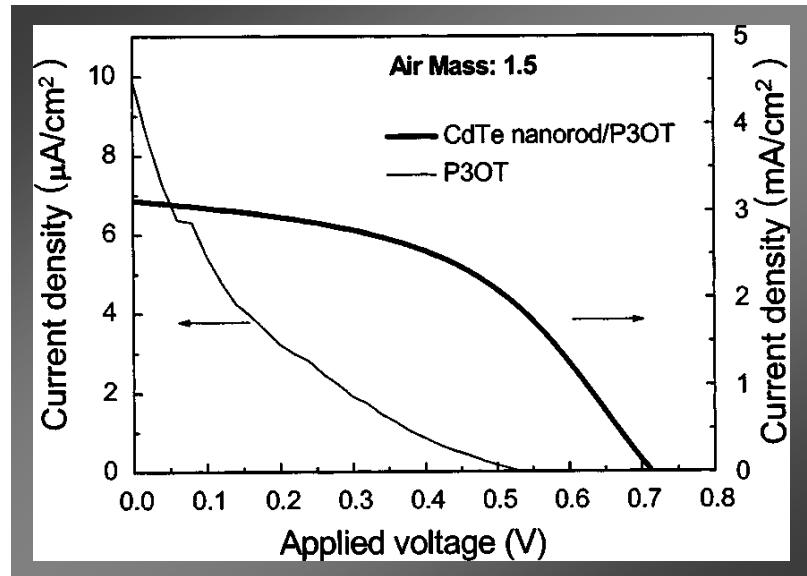
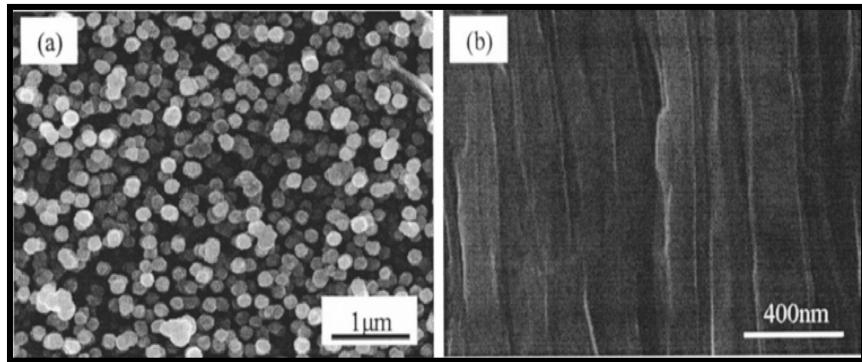
Electrical band gap energy of $\text{CdSe}_x\text{Te}_{1-x}$ tetrapodal nanocrystals²



Optical property of CdSe quantum dots with different size³

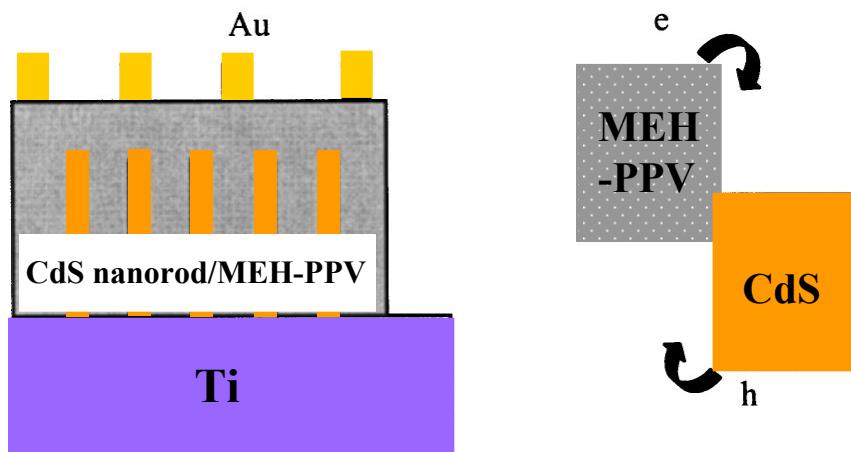
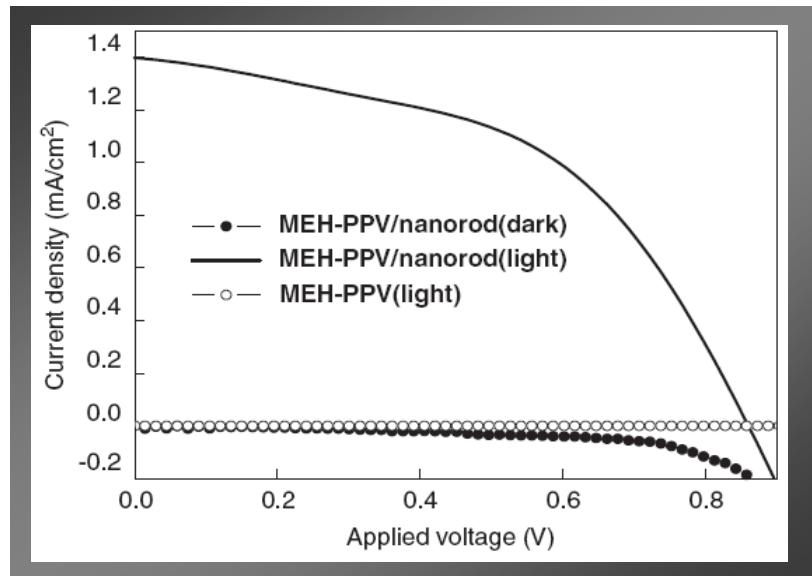
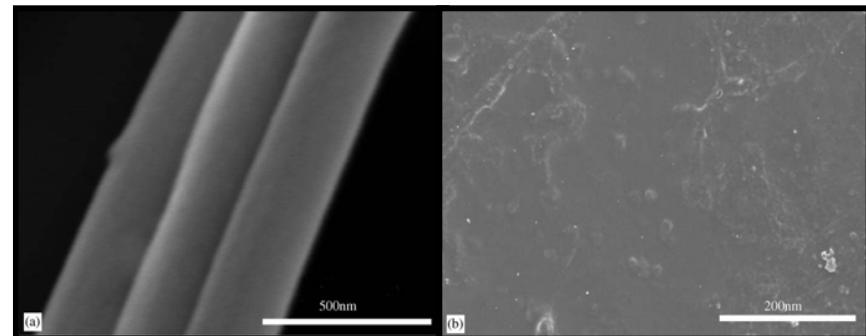
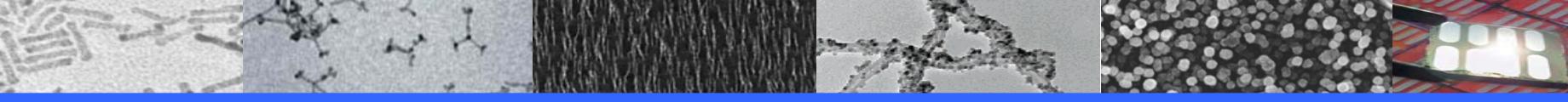


Previous research



Short-circuit current (J_{sc}): 3.12 mA/cm²
Open-circuit voltage (V_{oc}): 0.714 V
Fill factor (FF): 47.7 %
Power conversion efficiency (η): 1.06 %
under AM 1.5 conditions.⁵

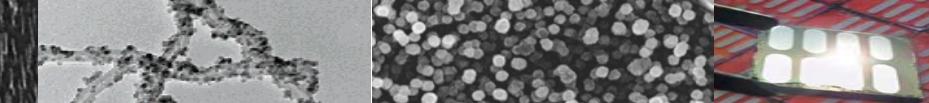




Short-circuit current (J_{sc}): 1.40 mA/cm²
Open-circuit voltage (V_{oc}): 0.858 V
Fill factor (FF): 49.6 %
Power conversion efficiency (η): 0.6 %
under AM 1.5 conditions.⁶



Hybrid nanostructures

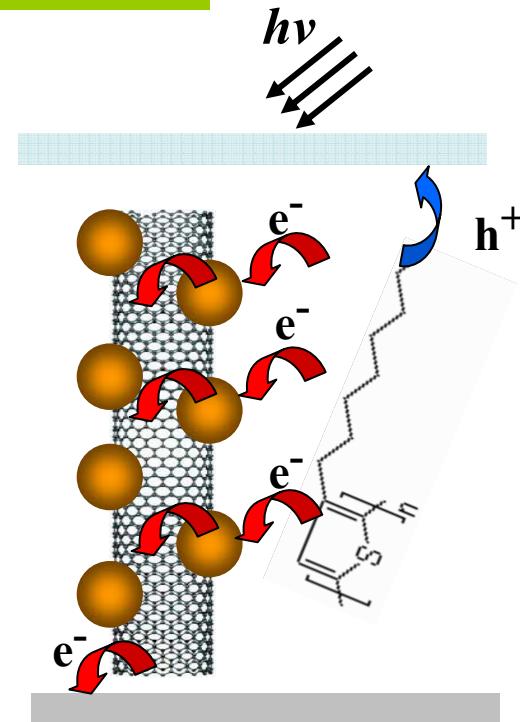
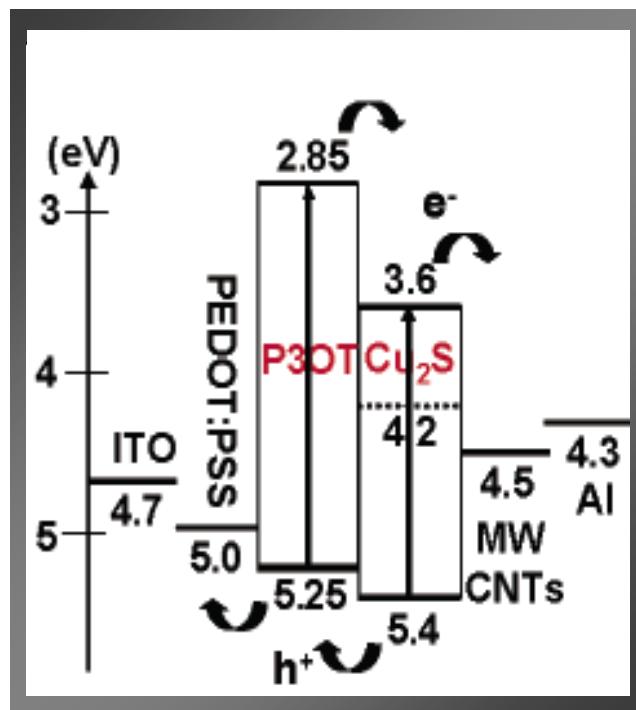


Quantum dots as charge separator and generator

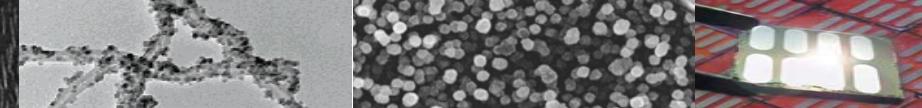
Carbon nanotubes as the supporter to improve charge separation and transport

Direct assembly

Hybrid nanostructures⁷



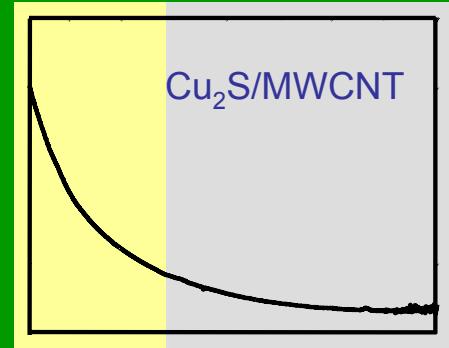
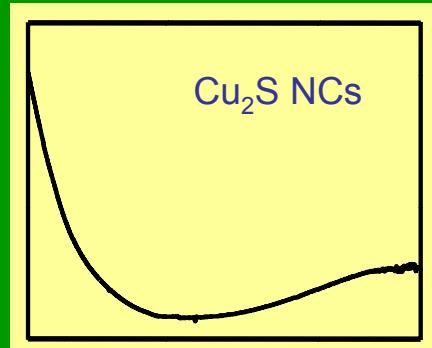
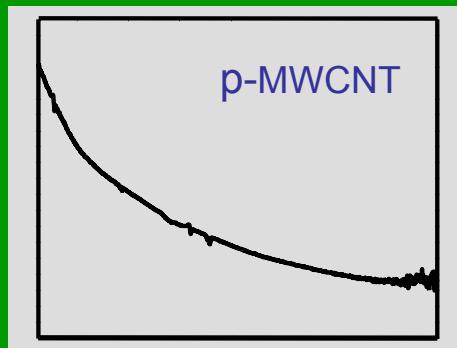
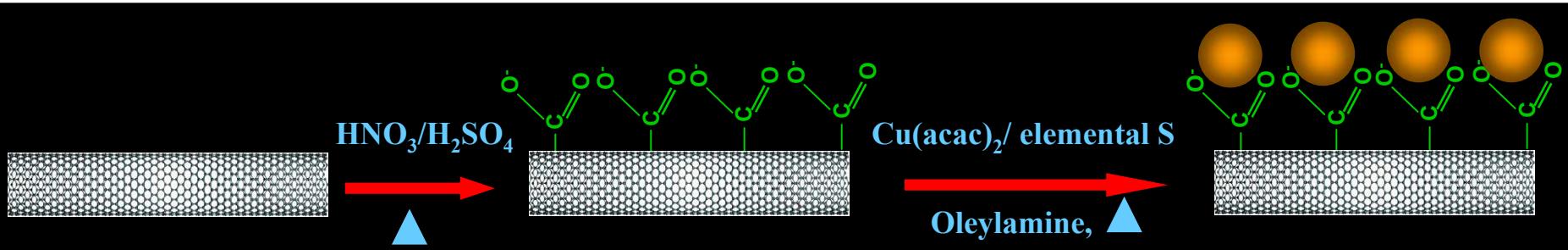
Material synthesis

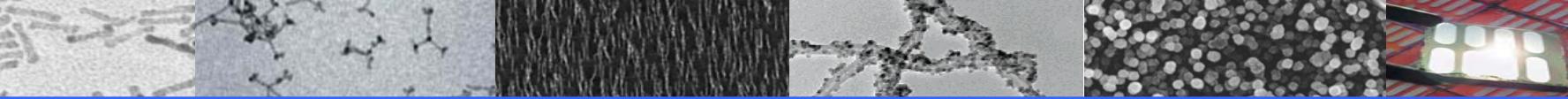


p-MWCNT in EtOH



Cu₂S/MWCNT in CHCl₃





0.05M Cu(acac)₂

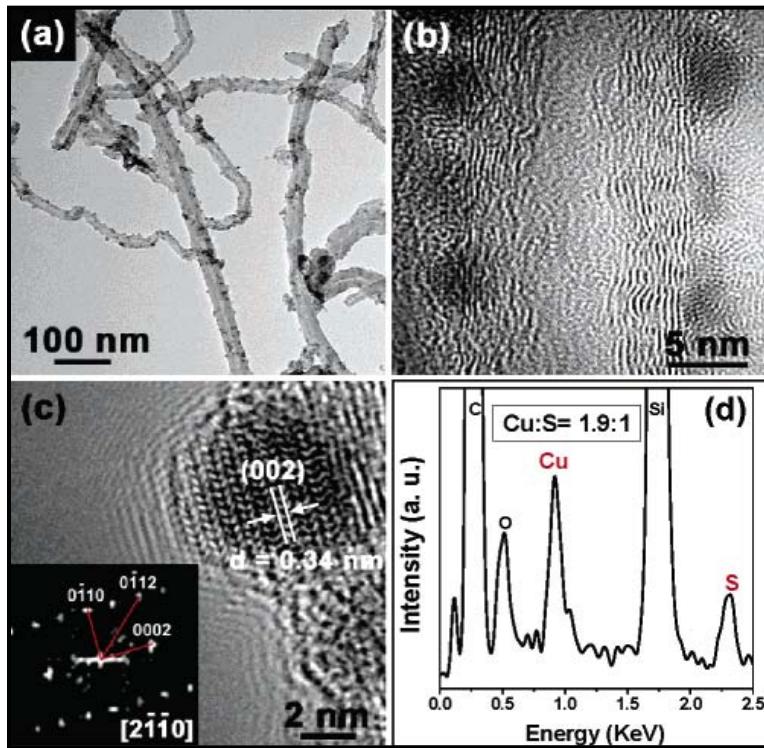


Fig.1

Hexagonal chalcocite β -Cu₂S (JCPDS no. 26-1116; $a = 3.961 \text{ \AA}$, $c = 6.722 \text{ \AA}$).

0.1M Cu(acac)₂

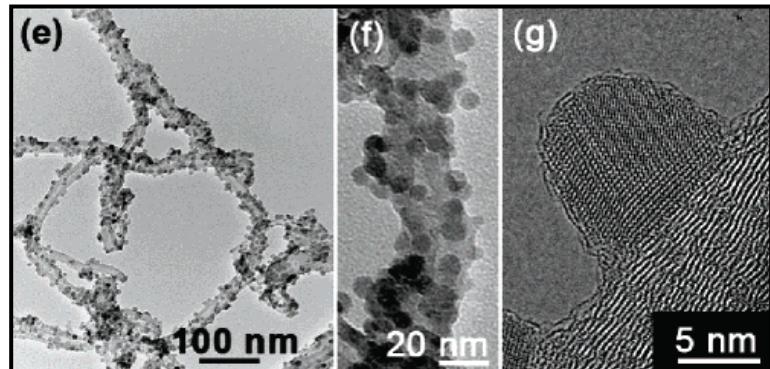


Fig.1

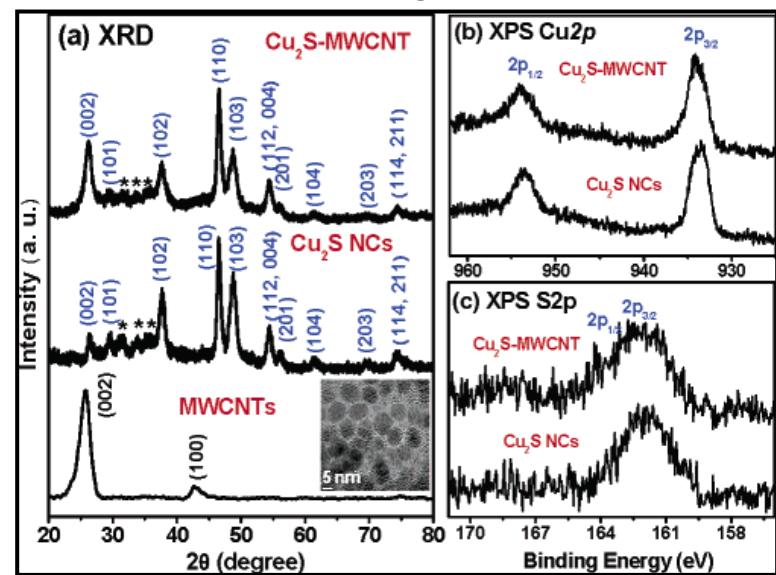


Fig.2



1.5M Cu(acac)₂

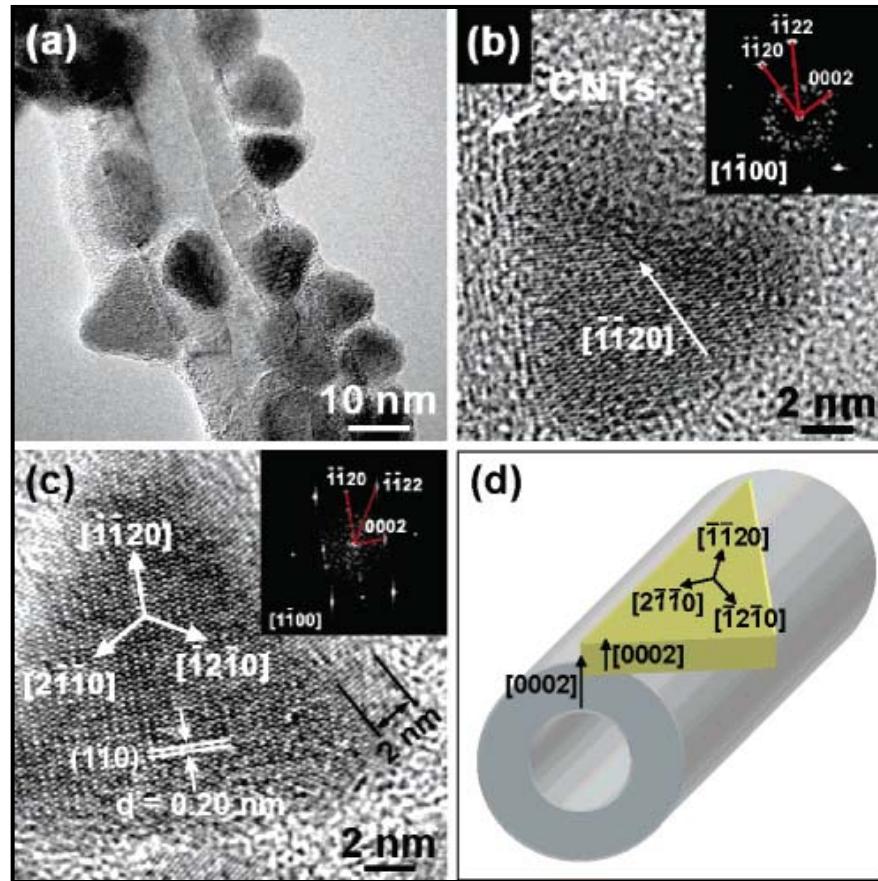


Fig.3



Solar cell fabrication

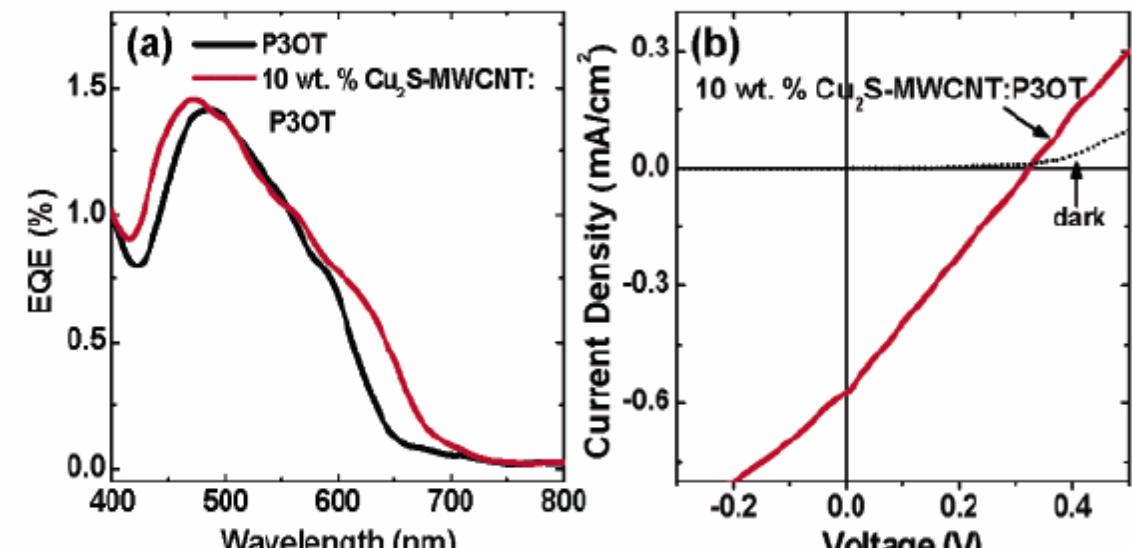
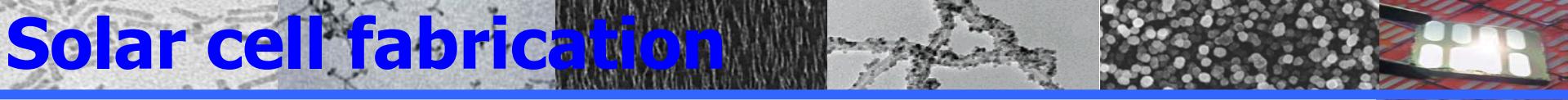
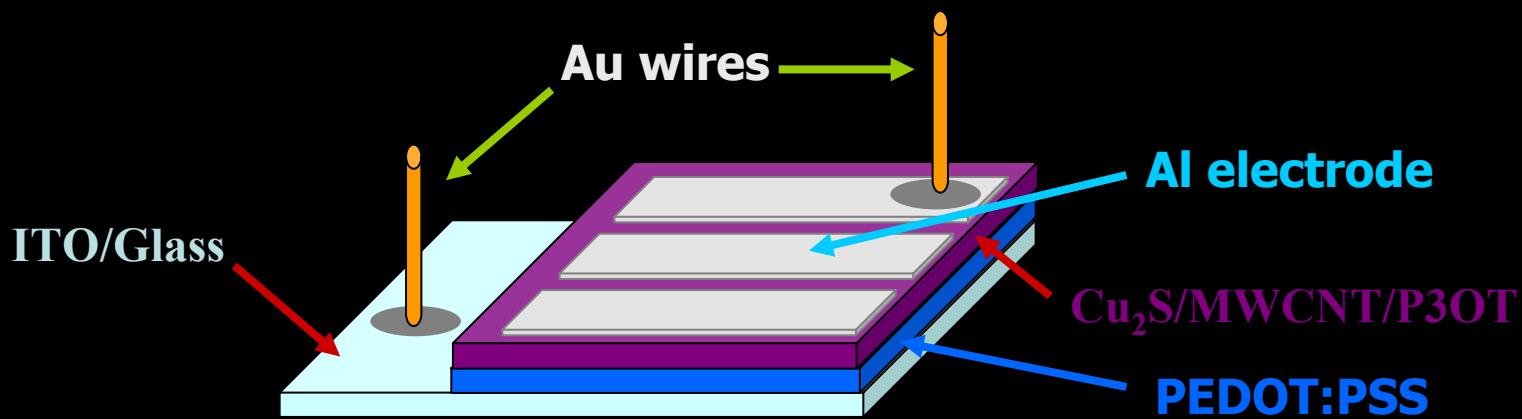


Fig.4

Short-circuit current (J_{SC}): 0.57 mA/cm²
Open-circuit voltage (V_{OC}): 0.32 V,
Fill factor (FF): 44 %
Power conversion efficiency (η): 0.08 %
under AM 1.5 conditions.

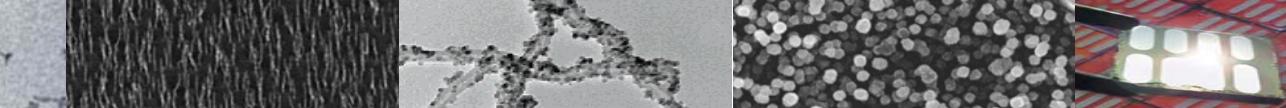


Conclusion

- 
1. Single-crystalline hexagonal-phase $\beta\text{-Cu}_2\text{S}$ NCs were grown *in situ* on acid-functionalized MWCNTs by the solvothermal method.
 2. The morphology of the Cu_2S NCs was varied from spherical particles (avg. size 4 nm) to triangular plates (avg. size 12 nm) by increasing the concentration of the precursors.
 3. The lattice matching between the (002) planes of Cu_2S and the (002) planes of the MWCNTs is thought to be an important factor in the growth of the Cu_2S NCs on the surface of the MWCNTs.
 4. The photovoltaic device fabricated from the blends of Cu_2S -MWCNT and P3OT responds more sensitively than those fabricated using Cu_2S NCs or MWCNTs alone.
 5. The maximum power conversion efficiency was found to be 0.08%. Direct, efficient electron transport from the photoexcited Cu_2S NCs to the MWCNTs would enhance their photocurrents.



References

- 
- [1] M. A. Green, K. Emery, D. L. King, Y. Hishikawa, W. Warta *Prog. Photovolt: Res. Appl.* **2007**, 15, 35.
 - [2] A. Thess, R. Lee, P. Nikolaev, H. Dai, P. Petit, J. Robert, C. Xu, Y.H. Lee, S.G. Kim, A. Rinzler, D.T. Colbert, G. Scuseria, D. Tomanek, J.E. Fischer, R. Smalley *Science* **1996**, 273, 483.
 - [3] (a) Y. Li, H. Zhong, R. Li, Y. Zhou, C. Yang, and Y. Li *Adv. Func. Mater.* **2006**, 16, 1705. (b) Y. Zhou, Y. Li, H. Zhong, J. Hou, Y. Ding, C. Yang and Y. Li *Nanotechnology* **2006**, 17, 4041.
 - [4] S. Kim, Y. T. Lim, E. G. Soltesz, A. M. De Grand, J. Lee, A. Nakayama, J. A. Parker, T. Mihalijevic, R. G. Laurence, D. M. Dor, L. H. Cohn, M. G. Bawendi, J. V. Frangioni, *Nat. Biotechnol.* **2004**, 22, 93.
 - [5] Y. Kang, N. -G. Park, D. Kim, *Appl. Phys. Lett.* **2005**, 86, 113101.
 - [6] Y. Kang, D. Kim, *Sol. Energy Mater. Sol. Cells* **2006**, 90, 166.
 - [7] H. Lee, S. W. Yoon, E. J. Kim, J. park, *Nano Lett.* **2007**, 7, 778.

