

### **Managing Emissions from Fossil Resources**

A Challenge to Technology and Policy



#### *IPCC Model Simulations of CO<sub>2</sub> Emissions*





# Resource Will Not Run Out

|                             | Consumption |      |             |                        | Resource          | Additional  |
|-----------------------------|-------------|------|-------------|------------------------|-------------------|-------------|
|                             | 1860–1994   | 1994 | Reserves    | Resources <sup>b</sup> | base <sup>c</sup> | occurrences |
| Oil                         |             |      |             |                        |                   |             |
| Conventional                | 103         | 3.21 | 150         | 145                    | 295               |             |
| Unconventional              | 6           | 0.16 | 183         | 336                    | 519               | 1,824       |
| Natural gas                 |             |      |             |                        |                   |             |
| Conventional <sup>d</sup>   | 48          | 1.87 | 141         | 279                    | 420               |             |
| Unconventional              |             |      | 192         | 258                    | 450               | 387         |
| Clathrates                  |             |      | 10 <u> </u> |                        |                   | 18,759      |
| Coal                        | 134         | 2.16 | 1,003       | 2,397                  | 3,400             | 2,846       |
| Total fossil<br>occurrences | 291         | 7.40 | 1,669       | 3,415                  | 5,084             | 23,815      |

 Table 9
 Aggregation of global fossil energy sources—all occurrences, in Gtoe<sup>a</sup>

<sup>a</sup>Sources: Historial consumption (46). Reserves, resources, and occurrences, see Tables 2–8. — = negligible volumes.

<sup>b</sup>Reserves to be discovered or resources developed to resources.

<sup>c</sup>Resource base is the sum of reserves and resources.

<sup>d</sup>Includes natural gas liquids.

H.H. Rogner, 1997

#### **Carbon as a Low-Cost Source of Energy**

![](_page_4_Figure_1.jpeg)

![](_page_4_Picture_2.jpeg)

H.H. Rogner, 1997

### Fossil fuels are fungible

![](_page_5_Figure_1.jpeg)

### The Challenge: Holding the Stock of CO<sub>2</sub> constant

![](_page_6_Figure_1.jpeg)

![](_page_6_Picture_2.jpeg)

![](_page_7_Figure_0.jpeg)

Pacala and Socolow, *Science* **305**, 968 – 972, (2004)

![](_page_7_Picture_2.jpeg)

# Orders of Magnitude

![](_page_8_Picture_1.jpeg)

![](_page_8_Picture_2.jpeg)

# A Triad of Large Scale Options

- Solar
  - Cost reduction and mass-manufacture
- Nuclear
  - Cost, waste, safety and security
- Fossil Energy
  - Zero emission, carbon storage and interconvertibility

Markets will drive efficiency, conservation and alternative energy

![](_page_9_Picture_8.jpeg)

# Small Energy Resources

- Hydro-electricity
  - Cheap but limited
- Biomass
  - Sun and land limited, severe competition with food
- Wind
  - Stopping the air over Colorado every day?
- Geothermal
  - Geographically limited
- Tides, Waves & Ocean Currents
  - Less than human energy generation

![](_page_10_Picture_11.jpeg)

#### **CCS** is technically feasible

It is affordable It can start today It is likely to be a major contributor to CO<sub>2</sub> reductions worldwide

# CARBON DIOXIDE CAPTURE AND STORAGE

Summary for Policymakers and Technical Summary

![](_page_11_Picture_4.jpeg)

![](_page_11_Picture_5.jpeg)

![](_page_11_Picture_6.jpeg)

Intergovernmental Panel on Climate Change

![](_page_11_Picture_8.jpeg)

# **Dividing The Fossil Carbon Pie**

![](_page_12_Figure_1.jpeg)

![](_page_12_Picture_2.jpeg)

![](_page_12_Picture_3.jpeg)

# Removing the Carbon Constraint

![](_page_13_Figure_1.jpeg)

![](_page_14_Picture_0.jpeg)

![](_page_15_Picture_0.jpeg)

Mineral carbonate disposal

![](_page_15_Picture_2.jpeg)

# Underground Injection

![](_page_16_Figure_1.jpeg)

![](_page_16_Picture_2.jpeg)

# Gravitational Trapping Subocean Floor Disposal

![](_page_17_Figure_1.jpeg)

![](_page_17_Picture_2.jpeg)

# **Energy States of Carbon**

![](_page_18_Figure_1.jpeg)

![](_page_18_Picture_2.jpeg)

![](_page_19_Figure_0.jpeg)

![](_page_19_Picture_1.jpeg)

### **Mineral Sequestration**

# $Mg_{3}Si_{2}O_{5}(OH)_{4} + 3CO_{2}(g) \rightarrow 3MgCO_{3} + 2SiO_{2} + 2H_{2}O(I) + 63kJ/mol CO_{2}$

### **Rockville Quarry**

### Belvidere Mountain, Vermont Serpentine Tailings

## Oman Peridotite

![](_page_22_Picture_1.jpeg)

Photo: Juerg Matter

![](_page_22_Picture_3.jpeg)

![](_page_23_Picture_0.jpeg)

# Many Different Options

- Flue gas scrubbing
  - MEA, chilled ammonia
- Oxyfuel Combustion
  - Naturally zero emission
- Integrated Gasification Combined Cycle
  - Difficult as zero emission
- AZEP Cycles
  - Mixed Oxide Membranes
- Fuel Cell Cycles
  - Solid Oxide Membranes

![](_page_24_Picture_11.jpeg)

![](_page_25_Picture_0.jpeg)

![](_page_26_Figure_0.jpeg)

![](_page_26_Picture_1.jpeg)

### Carbon makes a better fuel cell

 $C + O_2 \rightarrow CO_2$ no change in mole volume entropy stays constant  $\Delta G = \Delta H$ 

#### $2H_2 + O_2 \rightarrow 2H_2O$ large reduction in mole volume entropy decreases in reactants made up by heat transfer to surroundings $\Delta G < \Delta H$

![](_page_27_Picture_3.jpeg)

# **Proposed Membrane**

![](_page_28_Figure_1.jpeg)

#### **Multi-Phase Equilibrium**

 $CO_2 + O^{2-} = CO_3^{2-}$ 

![](_page_28_Picture_4.jpeg)

![](_page_29_Picture_0.jpeg)

![](_page_29_Picture_1.jpeg)

# Air Capture: A Different Paradigm

- Leave existing infrastructure intact
- Retain quality transportation fuels
- Eliminate shipping of CO<sub>2</sub>
- Open remote sites for CO<sub>2</sub> disposal
- Enable fuel recycling with low cost electricity

### Separate Sources from Sinks

![](_page_30_Picture_7.jpeg)

# Relative size of a tank

![](_page_31_Picture_1.jpeg)

LENFEST CENTER FOR SUSTAINABLE ENERGY

# Challenge: CO<sub>2</sub> in air is dilute

- Energetics limits options
  - Work done on air must be small
    - compared to heat content of carbon
    - 10,000 J/m<sup>3</sup> of air
- No heating, no compression, no cooling
- Low velocity 10m/s (60 J/m<sup>3</sup>)

Solution: Sorbents remove CO<sub>2</sub> from air flow

![](_page_32_Picture_8.jpeg)

# **CO<sub>2</sub> Capture from Air**

![](_page_33_Figure_1.jpeg)

### 1 m<sup>3</sup>of Air

40 moles of gas, 1.16 kg

wind speed 6 m/s

$$\frac{mv^2}{2} = 20 \,\mathrm{J}$$

0.015 moles of CO<sub>2</sub> produced by 10,000 J of gasoline

### Ca(OH)<sub>2</sub> as an absorbent

![](_page_34_Figure_1.jpeg)

Commess transfer is limited by diffusion in air boundary layer

![](_page_35_Picture_0.jpeg)

(6m/sec)

*Wind area that carries 10 kW* 

*O.2 m*<sup>2</sup> *for CO*<sub>2</sub> *Wind area that carries 22 tons of CO*<sub>2</sub> *per year* 

50 cents/ton of  $CO_2$  for contacting

![](_page_35_Picture_5.jpeg)

80 m<sup>2</sup>

for Wind Energy

![](_page_36_Figure_0.jpeg)

![](_page_37_Figure_0.jpeg)

(1) 
$$2NaOH + CO_2 \rightarrow Na_2CO_3 + H_2O$$
  
(2)  $Na_2CO_3 + Ca(OH)_2 \rightarrow 2NaOH + CaCO_3$   
(3)  $CaCO_3 \rightarrow CaO + CO_2$   
(4)  $CaO + H_2O \rightarrow Ca(OH)_2$   
(5)  $CH_4 + 2O_2 \rightarrow CO_2 + 2H_2O$   
(6)  $H_2O$  (1)  $\rightarrow H_2O$  (g)

 $\Delta H^{O} = -171.8 \text{ kJ/mol}$   $\Delta H^{O} = 57.1 \text{ kJ/mol}$   $\Delta H^{O} = 179.2 \text{ kJ/mol}$   $\Delta H^{O} = -64.5 \text{ kJ/mol}$   $\Delta H^{O} = -890.5 \text{ kJ/mol}$   $\Delta H^{O} = 41. \text{ kJ/mol}$ 

![](_page_37_Picture_3.jpeg)

![](_page_38_Figure_0.jpeg)

![](_page_38_Picture_1.jpeg)

![](_page_38_Picture_2.jpeg)

# Cost of CO<sub>2</sub> from Air

![](_page_39_Figure_1.jpeg)

# Sorbent Choices

![](_page_40_Figure_1.jpeg)

# Cost of CO<sub>2</sub> from Air (rescaled)

![](_page_41_Figure_1.jpeg)

![](_page_41_Picture_2.jpeg)

![](_page_42_Figure_0.jpeg)

60m by 50m 3kg of CO<sub>2</sub> per second 90,000 tons per year 4,000 people or 15,000 cars

*Would feed EOR for 800 barrels a day.* 

250,000 units for worldwide CO<sub>2</sub> emissions

![](_page_43_Picture_0.jpeg)

# The first of a kind

![](_page_44_Picture_1.jpeg)

![](_page_44_Picture_2.jpeg)

### Materially Closed Energy Cycles

![](_page_45_Figure_1.jpeg)

![](_page_46_Figure_0.jpeg)

![](_page_47_Figure_0.jpeg)

![](_page_48_Figure_0.jpeg)

![](_page_49_Figure_0.jpeg)

# Carbon Capture and Storage for Carbon Neutral World

- CCS simplifies Carbon Accounting
  - Ultimate Cap is Zero
  - Finite amount of carbon left

![](_page_50_Picture_4.jpeg)

![](_page_51_Figure_0.jpeg)

![](_page_52_Picture_0.jpeg)